

TECHNICAL AREA STATUS REPORT  
FOR  
LOW-LEVEL MIXED WASTE  
FINAL WASTE FORMS

Volume II  
APPENDICES

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**MASTER**

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## ACRONYMS

|      |   |
|------|---|
| °C   | degrees centigrade                        |
| °F   | degrees Fahrenheit                        |
| ANS  | American Nuclear Society                  |
| ASME | American Society of Mechanical Engineers  |
| ASTM | American Society of Testing and Materials |
| AVM  | Ateliers de Vitrification de Marcoule     |
| BNL  | Brookhaven National Laboratory            |
| BNWL | Battelle Northwest Laboratory             |
| BWR  | Boiling Water Reactor                     |
| CFR  | Code of Federal Regulations               |
| CIF  | Consolidated Incineration Facility        |
| CRT  | cathode ray tube                          |
| DOE  | Department of Energy                      |
| DOT  | Department of Transportation              |
| DWPF | Defense Waste Processing Facility         |
| EDTA | Ethylene Diamine Triacetic Acid           |
| EP   | Extraction Procedure                      |
| EPA  | Environmental Protection Agency           |
| FWF  | Final Waste Form                          |
| HEME | High Efficiency Mist Eliminator           |
| HEPA | High Efficiency Particulate Air           |
| HLRW | High Level Radioactive Waste              |
| HLW  | High Level Waste                          |
| HWTF | Hanford Waste Treatment Facility          |
| IAEA | International Atomic Energy Agency        |
| ICPP | Idaho Chemical Processing Plant           |
| ID   | inner diameter                            |
| IFR  | Integral Fast Reactor                     |
| INEL | Idaho National Engineering Laboratory     |
| ISV  | In Situ Vitrification                     |
| LANL | Los Alamos National Laboratory            |
| LDPE | Low Density Polyethylene                  |
| LLMW | Low Level Mixed Waste                     |
| LLW  | Low Level Waste                           |
| LLRW | Low Level Radioactive Waste               |
| LOMI | Low Oxidation State Metal Ion             |
| MCC  | Material Characterization Center          |
| MW   | Mixed Waste                               |
| MWIP | Mixed Waste Integrated Program            |
| MWTP | Mixed Waste Treatment Program             |
| N/A  | Non-applicable                            |
| NDT  | Non-destructive Testing                   |

|      |  |
|------|--|
| NRC  | Nuclear Regulatory Commission              |
| OTD  | Office of Technology Development           |
| PCB  | Polychlorinated Biphenyls                  |
| PCC  | Portland Cement Concrete                   |
| PCT  | Product Consistency Test                   |
| PNL  | Pacific Northwest Laboratory               |
| PWR  | Pressurized Water Reactor                  |
| QA   | Quality Assurance                          |
| RCRA | Resource Conservation and Recovery Act     |
| RFP  | Rocky Flats Plant                          |
| SA/V | surface area over volume                   |
| SLS  | Soda-Lime-Silicate                         |
| SPC  | Sulfur Polymer Cement                      |
| SRL  | Savannah River Laboratory                  |
| SRP  | Savannah River Plant                       |
| SRS  | Savannah River Site                        |
| TASR | Technical Area Status Report               |
| TBP  | Tributyl Phosphate                         |
| TCLP | Toxicity Characteristic Leaching Procedure |
| TRU  | transuranic                                |
| UF   | Urea Formaldehyde                          |
| VES  | Vinyl-Ester Styrene                        |
| VRS  | Volume Reduction and Solidification        |
| WERF | Waste Experimental Reduction Facility      |
| WG   | Working Group                              |
| WHC  | Westinghouse Hanford Company               |
| WIPP | Waste Isolation Pilot Plant                |
| WSRC | Westinghouse Savannah River Company        |
| WVDP | West Valley Demonstration Project          |



## NOMENCLATURE

|                                |                             |
|--------------------------------|-----------------------------|
| Ag                             | silver                      |
| Al                             | aluminum                    |
| Al <sub>2</sub> O <sub>3</sub> | aluminum trioxide           |
| AlPO <sub>4</sub>              | aluminum phosphate          |
| Am                             | americium                   |
| As                             | arsenic                     |
| B                              | boron                       |
| B <sub>2</sub> O <sub>3</sub>  | boron trioxide              |
| Ba                             | barium                      |
| Be                             | beryllium                   |
| Br                             | bromine                     |
| BTU                            | British Thermal Units       |
| Ca                             | calcium                     |
| CaO                            | calcium oxide               |
| Cd                             | cadmium                     |
| Cl                             | chlorine                    |
| Co                             | cobalt                      |
| CO                             | carbon monoxide             |
| CO <sub>2</sub>                | carbon dioxide              |
| CoS                            | cobalt sulfide              |
| CoSe                           | cobalt selenide             |
| Cr                             | chromium                    |
| CrO <sub>2</sub>               | chromium dioxide            |
| Cs                             | cesium                      |
| Cu                             | copper                      |
| Eh                             | electro-chemical potential  |
| F                              | fluorine                    |
| Fe                             | iron                        |
| Fe <sub>2</sub> O <sub>3</sub> | iron trioxide               |
| g/cm <sup>2</sup>              | grams per square centimeter |
| gpm                            | gallons per minute          |
| H <sub>2</sub>                 | hydrogen                    |
| H <sub>2</sub> SO <sub>4</sub> | sulfuric acid               |
| Hg                             | mercury                     |
| HNO <sub>3</sub>               | nitric acid                 |
| hp                             | horsepower                  |
| hr                             | hour                        |
| kg                             | kilogram                    |
| kW                             | kiloWatt                    |
| lb                             | pound                       |
| Li                             | lithium                     |
| MGy                            | millions of Gray (100 rems) |

|   |                                |
|---|--------------------------------|
| Mn  | manganese                      |
| MnO   | manganese xide                 |
| Mo  | molybdenum                     |
| MPa   | megaPascal                     |
| mph   | miles per hour                 |
| Na  | sodium                         |
| NaCl  | sodium chloride                |
| NaK   | sodium potassium               |
| NaNO <sub>3</sub>                               | sodium nitrate                 |
| Na <sub>2</sub> O                               | sodium oxide                   |
| Ni  | nickel                         |
| NiS   | nickel sulfide                 |
| NiSe  | nickel selenide                |
| NiTe  | nickel teluride                |
| NL  | normalized log                 |
| NO  | nitrogen oxide                 |
| Np  | neptunium                      |
| Pb  | lead                           |
| PbI <sub>2</sub>                                | lead iodide                    |
| Pb <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> | lead phosphate                 |
| PO <sub>4</sub>                                 | phosphate                      |
| psi   | pounds per square inch         |
| Pu  | plutonium                      |
| Ru  | ruthenium                      |
| scfm  | standard cubic feet per minute |
| Sb  | antimony                       |
| Se  | selenium                       |
| Si  | silicon                        |
| SiO <sub>2</sub>                                | silica                         |
| SO <sub>2</sub>                                 | sulfur dioxide                 |
| SO <sub>4</sub>                                 | sulfate                        |
| Sr  | strontium                      |
| SrO <sub>3</sub>                                | strontium trioxide             |
| Tc  | technetium                     |
| TiO <sub>2</sub>                                | titanium dioxide               |
| U   | uranium                        |
| V   | vanadium                       |

## INTRODUCTION

The descriptions of the low-level mixed waste (LLMW) streams that are considered by the Mixed Waste Integrated Program (MWIP) are given in Appendix A. This information was taken from descriptions generated by the Mixed Waste Treatment Program (MWTP). Appendix B provides a list of characteristic properties initially considered by the Final Waste Form (FWF) Working Group (WG). A description of facilities available to test the various FWFs discussed in Volume I of DOE/MWIP-3 are given in Appendix C. Appendix D provides a summary of numerous articles that were reviewed on testing of FWFs. Information that was collected by the tests on the characteristic properties considered in this report are documented in Appendix D. The articles reviewed are not a comprehensive list, but are provided to give an indication of the data that are available.

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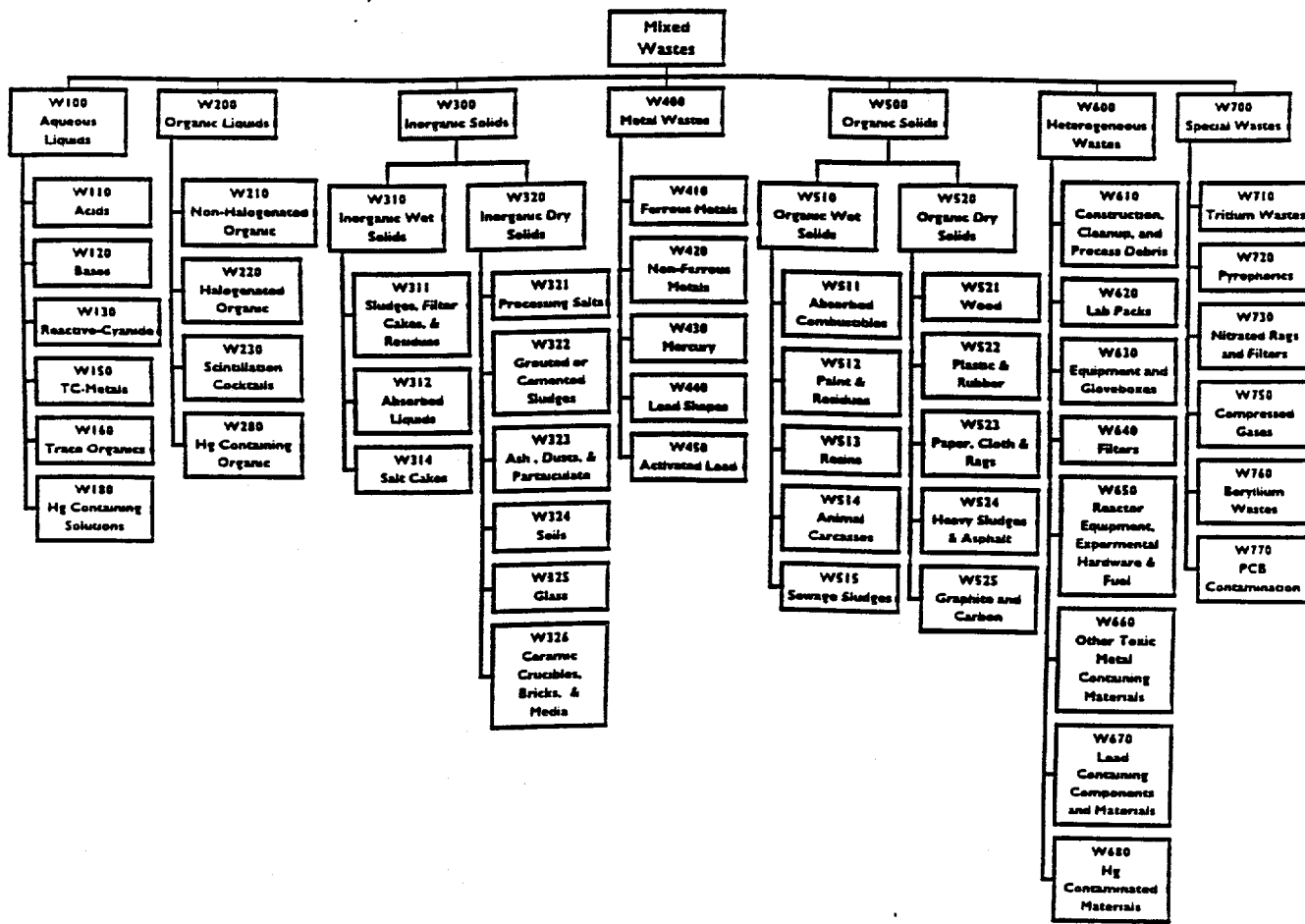
**APPENDIX A**  
**LLMW CHARACTERIZATION CODE DESCRIPTIONS**

A-2

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Wayne Ross at PNL has developed the following waste code descriptions for the Mixed Waste Treatment Project sponsored by DOE EM-30. The descriptions include LLMW streams from some of the DOE sites that store or generate LLMW. The waste has been divided into seven general waste categories and 52 subcategories. See figure below for a matrix of the categories described herein.

Fig. A-1. Mixed waste matrix characterization categories



| CODE # | TITLE                    | DESCRIPTION  |
|--------|--------------------------|--|
| W100   | Aqueous Liquids          | Aqueous solutions with less than 1% organic content. Solids must be pumpable but can be up to about 35-40% of the mass. Wastes in this specific category could not be assigned to a subcategory because of lack of information or because of multiple characteristics of the waste stream. Streams contain some of the following: As, Cd, Cr, Pb, Hg, and Ag. Contains trace hazardous organics. |
| W110   | Corrosive acids          | Principle sources of these wastes are general waste water cleanup, plating line solutions, and electropolishing activities. Streams contain some of the following: Cr, Pb, Hg, and Ag. Contains trace hazardous organics.  |
| W120   | Corrosives Bases         | These are solutions with a high pH generated from a variety of activities. Streams contain some of the following: As, Cd, Cr, and Pb.  |
| W130   | Reactive-Cyanide         | Any stream that contains cyanide as a significant component. Solutions will generally be basic. Streams contain some of the following: As, Cd, and Cr.   |
| W150   | TC-Metals                | Streams containing toxic metals. Streams contain some of the following: Cd, Cr, Pb, Hg, and Se. Contain trace hazardous organics.  |
| W160   | Trace Organics           | Streams with organic levels of less than 1%. Some streams also contain some lead.  |
| W180   | Hg Containing Solutions  | Any of the 100 series solutions that may contain Hg. Streams may also contain some lead.   |
| W200   | Organic Liquids          | Liquid streams containing more than 1% organic. Solids must be pumpable (except for code 230) but can be comprise of up to about 35-40% of the mass. Wastes in this specific category could not be assigned to any subcategory because of lack of information.   |
| W210   | Non-Halogenated Organics | Solvents free of F, Cl, Br, (e.g. oils, hexone, methanol). Wastes may contain Cd, Cr, Pb, Hg, and Se as trace contamination.   |



| CODE # | TITLE                              | DESCRIPTION   |
|--------|------------------------------------|---|
| W220   | Halogenated Organics               | Solvents containing F, Cl, Br, etc. Contaminated freon is one specific stream. Streams may contain Cd, Cr, Pb, Hg, and Ag as trace contaminants.  |
| W230   | Scintillation Cocktails            | Solutions used for scintillation counting. Solutions may be contained in the original glass or plastic analysis bottles. Streams may contain Cd, Cr, and Pb as trace contamination.   |
| W280   | Hg Containing Organics             | Any 200 series waste that may contain Hg. Streams may contain Se as a trace contaminant.  |
| W300   | Inorganic Solids                   | This category includes all inorganic solids except metal wastes. The inorganic materials are generally oxides, but also include salts. The major category is divided into wet solids, dry solids, and soils.  |
| W310   | Wet Inorganic Solids               | A solid mass, non pumpable, that contains liquids within the pore structure of the solid. Liquid fractions should be more than 5% of the mass of the material. Wastes in this specific category could not be assigned to any subcategory because of lack of information.  |
| W311   | Sludges, Filter Cakes and Residues | These materials are generally from wastewater cleanup or from settling ponds. They may contain organic materials in limited quantities from laundry or other sources. Heavy metals are present in some sludges. Cemented sludges are W322 wastes. Streams may contain Ba, Cd, Cr, Pb, Hg, Ag, V, and cyanide as trace contamination. Contains trace hazardous organics. |
| W312   | Absorbed Liquids                   | Aqueous or organic liquids absorbed onto a solid such as vermiculite or clay.   |
| W314   | Salt Cakes                         | Evaporated salt solutions, either nitrate and chloride, that may contain a high residual water content. Streams may contain Ba as trace contamination.  |

| CODE # | TITLE                                | DESCRIPTION  |
|--------|--------------------------------------|--|
| W320   | Inorganic Dry Solids                 | Inorganic solids without notable free liquid. Most wastes will contain some combined water or absorbed water. Liquid fractions will generally be less than 5% of the mass of the material. Wastes in this specific category could not be assigned to any subcategory because of lack of information. Streams may contain As, Ba, Pb, and Se as trace contamination. Contains trace hazardous organics.   |
| W321   | Processing Salts                     | Salts that have been used in processing; mostly F, Cl, NO <sub>3</sub> based salts. Streams may contain As, Ba, Cd, Cr, Pb, and Se as trace contamination. Contains trace hazardous organics.  |
| W322   | Grouted or Cemented Sludges          | Sludges that contain cement either as a water absorber or that are mixed with cement to produce a homogenous solid waste. Streams may contain Cd, Cr, and Pb as trace contamination. Contains trace hazardous organics.  |
| W323   | Ash, Dusts, and Particulates         | Fine particulate wastes. Typical wastes are ash from incinerators, dusts, and paint chips. Cemented particulates are included in code W322. Streams may contain As, Cd, Cr, Pb, Ag, and Se as trace contamination. Contains trace hazardous organics.  |
| W324   | Soils                                | Contaminated soils from spills, leaks, cleanups, and waste burial. Streams may contain As, Ba, Cd, Cr, Pb, Hg, Ag, and Se as trace contamination. Contains trace hazardous organics.   |
| W325   | Glass                                | Items composed primarily of glass, including process equipment, laboratory equipment, window materials, vessels, bottles, light bulbs, or glass beads or forms used within process equipment or for abrasion of surfaces. The glass may contain small amounts of organic or other inorganic materials. Streams may contain Ba, Cd, Pb, and Hg as trace contamination. Contains trace hazardous organics. |
| W326   | Ceramic, Crucibles, Bricks and Media | Oxide materials generally used as crucibles or refractories. They may also be beads or shapes used for catalysts, reactor beds, or for milling or grinding. Streams may contain As, Cr, Pb, Hg, and Ag as trace contamination. Contains trace hazardous organics.  |

| CODE # | TITLE                 | DESCRIPTION   |
|--------|-----------------------|---|
| W400   | Metal Wastes          | Inorganic solids classed as metals. This category includes wastes that are mixed ferrous and nonferrous metals or other defined metals. Streams may contain Cd, Cr, Pb, and Hg as trace contamination. Contains trace hazardous organics.                   |
| W420   | Nonferrous Metals     | Nonferrous metals are the principle component, such as aluminum, copper, and cadmium. May have trace Pb contamination.  |
| W430   | Mercury               | Liquid mercury pourable from containers.  |
| W440   | Lead Shapes           | Lead bricks, shipping casks, or shielding materials. The lead should only be surface contaminated. May have some Cr contamination. Contains trace hazardous organics.   |
| W450   | Activated Lead        | Lead activated from its use in radiations fields, such as from a reactor or accelerator.  |
| W500   | Organic Solids        | Waste streams that are predominately organic materials and candidates for direct treatment most likely by incineration. Wastes that are mixtures of the various subcategories or undefinable beyond this category are included.                             |
| W510   | Organic Wet Solids    | These are organic solids that have significant content of liquids generally combined with the waste matrix. Streams may contain As, Ba, Cd, Cr, Pb, Ag, and Se as trace contamination.  |
| W511   | Absorbed Combustibles | Rags or paper used to wipe up spills. Organic liquid spill cleanup materials. Streams may contain Ba, Cr, Pb, and Se as trace contamination. Contains trace hazardous organics.   |
| W512   | Paint and Residues    | New or removed paint. The paint may be liquid or have a liquid content either as original paint or a paint stripper. It may also only be paint chips. Streams may contain As, Ba, Cr, Pb, and Hg as trace contamination. Contains trace hazardous organics. |

| CODE # | TITLE                     | DESCRIPTION  |
|--------|---------------------------|--|
| W513   | Resins                    | Spent organic resins including spent carbon filters used in waste water cleanup. Streams may contain As, Ba, Hg, Ag, and Se as trace contamination. Contains trace hazardous organics. Some resins are not compatible with cement or grout systems.  |
| W514   | Animal Carcasses          | Dead animals or parts of animals. Most animals will have been used in testing and may contain agents to stabilize the remaining materials. Streams may contain As, Ba, Hg, Ag and Se as trace contamination.   |
| W520   | Organic Dry Solids        | These are organic solids that do not have significant content of liquids. Streams that may contain a variety of solid organic wastes such as mixtures of wood, plastic, and paper. Streams may contain Ba, Cd, Pb, and Ag as trace contamination. Contains trace hazardous organics.   |
| W521   | Wood                      | Wood timbers used for packaging or temporary structures similar to construction debris, however this would be wood as the single principle component. Some noncombustible tramp materials such as nails is to be expected in concentrations of less than 5%.   |
| W522   | Plastic and Rubber        | Plastic sheeting or components such as benelex or plexiglass. It also includes glovebox gloves. Leaded gloves would be included in W670, but this stream may include trace Pb. Contains trace hazardous organics.  |
| W523   | Paper, Cloth, and Rags    | Contaminated clothing and wipes. May include traces of metals. Streams may contain As, Ba, Cd, Cr, Pb, Hg, and Ag as trace contamination. Contains trace hazardous organics.   |
| W524   | Heavy Sludges and Asphalt | Heavy sludges are organic based materials that have such high viscosity that they can not be poured from a drum. Asphalt roadways or walkways that become contained with radioactivity. It would contain rock and organic binders. Streams may contain Cr, Pb, Hg, Ag, and Se as trace contamination. Contains trace hazardous organics. |
| W525   | Graphite and Carbon       | Crucibles or components of graphite or carbon.   |

| CODE # | TITLE                                     | DESCRIPTION   |
|--------|---|---|
| W600   | Heterogeneous Wastes                      | Solid materials that may contain a mixture (either as manufactured or from packaging) of organic, inorganic solid, and/or metallic materials. Wastes in this specific category could not be assigned to any subcategory because of lack of information or because they are a mixture of other waste types. Streams may contain Cd, Cr, Pb, Se and Ag as trace contamination. Contains trace hazardous organics.   |
| W610   | Construction, Cleanup, and Process Debris | Construction debris is generated from the remodeling of radioactive facilities and could include piping, wiring, wall materials, and flooring. Also includes wood and plastic sheeting used for temporary radioactive particulate containment and cleanup and process debris that are mixtures of various types of material that were commingled for disposal. A mixture of combustible organic, metal, and ceramic materials. Streams may contain As, Ba, Cd, Cr, Pb, Hg, Se and Ag as trace contamination. Contains trace hazardous organics. |
| W620   | Lab Packs                                 | Mixtures of chemical in drums. Chemicals are mostly solid, but can contain liquids. Packing materials such as vermiculite are commonly included. Streams may contain As, Ba, Cd, Cr, Pb, Hg, and Se as trace contamination. Contains trace hazardous organics.  |
| W630   | Equipment and Gloveboxes                  | Process equipment generally metallic, but that may contain oils, grease, or process materials. The equipment may include some associated electrical wiring and piping. Packaged gloveboxes that may contain equipment, windows, gloves, and other organic materials. They may also be contaminated with oils or sludges. Since many gloveboxes have been used with TRU materials some level of alpha contamination may be expected. Leaded gloves may be present. Streams may contain Cr and Pb as trace contamination.                         |

| CODE # | TITLE   | DESCRIPTION  |
|--------|---|--|
| W640   | Filters   | HEPA filters, and other process filters that may be metallic, organic or ceramic are included. Filters are contaminated with fine particulate. HEPA filters may be either wood or metal frame. Streams may contain As, Ba, Cd, Cr, Pb, Hg, Se and Ag as trace contamination. Contains trace hazardous organics.  |
| W650   | Reactor Equipment, Experimental Hardware and Fuel | Reactor equipment may contain listed chemicals and toxic metals. They can become highly radioactive during neutron irradiation in reactors. These materials may need special care as they contain activated metals and radiation that can not be removed by decontamination. Streams may contain As, Cd, Pb, and Hg as trace contamination. Contains trace hazardous organics. |
| W660   | Other Toxic Metal Containing Materials            | As, Ba, Cd, Cr, Se, and Ag contaminated materials. Lead may be present as a trace contamination. Contains trace hazardous organics.  |
| W670   | Lead Containing Components and Materials          | Glovebox gloves or shielding aprons containing lead or lead oxide in a rubber or plastic binder, which includes lead acid batteries. May also be lead wool or lead base solder materials.  |
| W680   | Hg-Contaminated Materials                         | Materials contaminated with Hg. Materials may be of any general type. Streams may contain Ba, Cd, Cr, Pb, Se and Ag as trace contamination.  |
| W700   | Special Wastes                                    | These are waste streams that will require extra care for treatment. They may require specific treatment equipment or processes not consistent with current radioactive materials handling guidance.  |
| W710   | Tritium Wastes                                    | Waste streams contaminated with tritium. They may be liquid or solids. Streams may contain Cr and Hg as trace contamination. Contains trace hazardous organics.  |
| W720   | Pyrophorics                                       | Reactive metals. They are typically sodium metal or sodium metal alloys, but can also be particulate fines of aluminum, uranium, beryllium, zirconium, or other pyrophoric materials and may be mixed with reaction preventing materials.  |

| CODE # | TITLE                     | DESCRIPTION  |
|--------|---------------------------|--|
| W730   | Nitrated Rags and Filters | Rags that have been used to absorb nitric acid and then been left in storage.  |
| W750   | Compressed Gases          | Aerosol cans and gas cylinders of any type of composition.   |
| W760   | Be Wastes                 | Be metal chips, dusts, or materials contaminated with Be. Contains trace hazardous organics.                                     |
| W770   | PCB Contamination         | Solids and liquids contaminated with PCBs. A wide range of material types. Streams may contain Cr and Pb as trace contamination. |

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**APPENDIX B**  
**FINAL LIST OF PERTINENT FINAL WASTE FORM CHARACTERISTICS**

B-2

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One of the activities of the Final Waste Form Working Group was to determine the characteristics that a final waste form should possess. A list of 21 pertinent waste form characteristics was initially generated, and is shown in Table B-1.

Table B-1. Initial List of Pertinent Final Waste Form Characteristics

|                      |                           |               |
|----------------------|---------------------------|---------------|
| Compressive Strength | Homogeneity               | Handling      |
| Thermal Stability    | Particle Size             | Melting Point |
| Radiation Stability  | Chemical Durability       | Solubility    |
| Biological Stability | Compositional Flexibility | Volatility    |
| Leach Resistance     | Gas Generation            | Criticality   |
| Immersion Stability  | Aging/Time Dependence     | Void Fraction |
| Free Liquid          | RCRA Compliance           | Waste Loading |

The above list of 21 characteristics was later revised to include only the eleven characteristics which are discussed in the TASR. Ten of the characteristics shown in Table B-1 were deleted due to redundancy, lack of methods for measurement, and inappropriateness considering the charter of the technical committee. A short description of these discarded characteristics and the rationale for deleting them follows:

Homogeneity - This characteristic refers to the degree of homogeneity of the waste and the binder. Although this characteristic may be important when considering representative sampling and the characterization of the final waste form, it does not address performance.

Particle Size - This characteristic was originally included to prevent the possibility of respirable particles being present and thus creating safety issues during storage and handling. A baseline was established that would only include "monolithic" waste forms, thus restricting the physical form and removing the need for this characteristic. This requirement is indirectly included in the compressive strength characteristic.

Aging/Time Dependence - This characteristic refers to the long-term performance of the waste form. Although it is an important characteristic, it was felt that it was too broad and could be more effectively quantified by more specific characteristics such as chemical stability, and leach resistance.

Handling - This characteristic was included to address the safety issues and the physical considerations associated with handling the waste form. This characteristic is not considered a performance criteria.

Melting Point - This characteristic addresses the need for the waste form to maintain its physical dimensions and to remain in a solid form at high

temperature. This characteristic is important under accident conditions and is addressed by thermal stability.

Volatility - This characteristic refers to the generation of volatile species under normal storage and disposal conditions. It was felt that this characteristic was covered under the gas generation and RCRA non-characteristic criteria.

Criticality - This characteristic addresses the potential for formation of a critical mass. Based on the concentration limits established for LLW the potential for criticality does not exist.

Void Fraction - This characteristic is referred to in the NRC guidance for LLW. It is, however, a consideration that addresses structural stability and economics. It was felt that the stability requirement is covered by the retained characteristics; the economic concern is beyond the scope of this report.

Waste Loading - This characteristic refers to the ability of the binder to effectively isolate large quantities of waste. Although this is an important characteristic of the final waste form, it primarily effects the economics of the process and therefore was outside the scope of this report.

Solubility - This characteristic refers to the "equilibrium" concentrations of species in solution. It is a major factor in leach resistance and, for the purposes of this review, is included in that characteristic.

**APPENDIX C**  
**SUMMARY OF CURRENT STATUS**

C-2

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GLASS

**TITLE:** Electrically Heated Glass Melter Connected to a Wet Scrubber Offgas System

**EQUIPMENT:** Joule-Heated Glass Melter

**LOCATION:** EG&G Mound Applied Technologies  
Miamisburg, Ohio

**CONTACT:** Larry Klingler: (513) 865-3078  
EG&G Mound

**CAPABILITIES:** Thermal Capacity -- 400,000 Btu/hr  
Temperature -- 730°C to 1,350°C  
Maximum Throughput -- N/A  
Treatable Waste -- RCRA hazardous and low-level radioactive wastes

**VERSATILITY:** N/A

**DESCRIPTION:** The system under study at Mound consists of an electrically heated glass melter connected to a wet offgas scrub system. The glass furnace was purchased from Penberthy Electromelt, Inc. The unit is equipped with feed systems designed to provide the flexibility to introduce a variety of waste types in accurately metered quantities.

**OFFGAS SYSTEM:** The offgas system features a primary and secondary wet scrubbing by means of a spray tank and a high efficiency venturi scrubber. The scrubber system is followed by a cyclone demister and several stages of HEPA filtration. A scrub liquid recycling system provides caustic solution for the scrubbing operations.

**STATUS:** Has operated but shutdown pending environmental assessment review.

C-4

GLASS

**TITLE:** High Bay Ceramic Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Hanford Site  
Richland, WA

**CONTACT:** Chris Chapman: (509) 376-6576  
Pacific Northwest Laboratory

**CAPABILITIES:** Thermal Capacity -- ~1,500,000 BTU/hr<sup>1</sup>  
Temperature -- 1,200°C  
Throughput -- Inorganic Solids: 570-710 kg/d; Combustibles:  
850-  
1,714 kg/d<sup>1</sup>; Slurries: 630-1,050 kg/d  
Treatable Waste -- Non-radioactive RCRA hazardous waste

**VERSATILITY:** Waste feeding systems will handle slurries (acid, bases, neutral), sludges, solids (soils, combustible trash, tramp metals). Device is not suitable for wastes with large metal pieces.

**DESCRIPTION:** Melter has a cylindrical melt cavity 2 ft in diameter and is 2.83 ft deep with a flat floor. Reaction chamber above the melt can be maintained at temperatures up to 1,000°C with unit's plenum heaters. Unit has a batch bottom drain and a conventional bottom take off, overflow discharge. Melter is enclosed in a water cooled stainless steel jacket. Electrodes are side entering plates (2 sets) made of Inconel 690. The refractory lining is made of Monofrax K-3. Feed ports are 6-in. ID on the current lid. Larger sizes can be achieved with a new lid. Control and data logging equipment are associated with the system. The melter is shown in Figure C-1.

**OFFGAS SYSTEM:** 800 SCFM offgas treatment capacity  
Submerged bed scrubbers  
High efficiency mist eliminator (HEMEs)  
Packed column  
Hydrosonic scrubber

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<sup>1</sup>Existing blower limits this to about 1,200,000 BTU/hr and 1,280 kg/d of 22,000 BTU/kg feed.



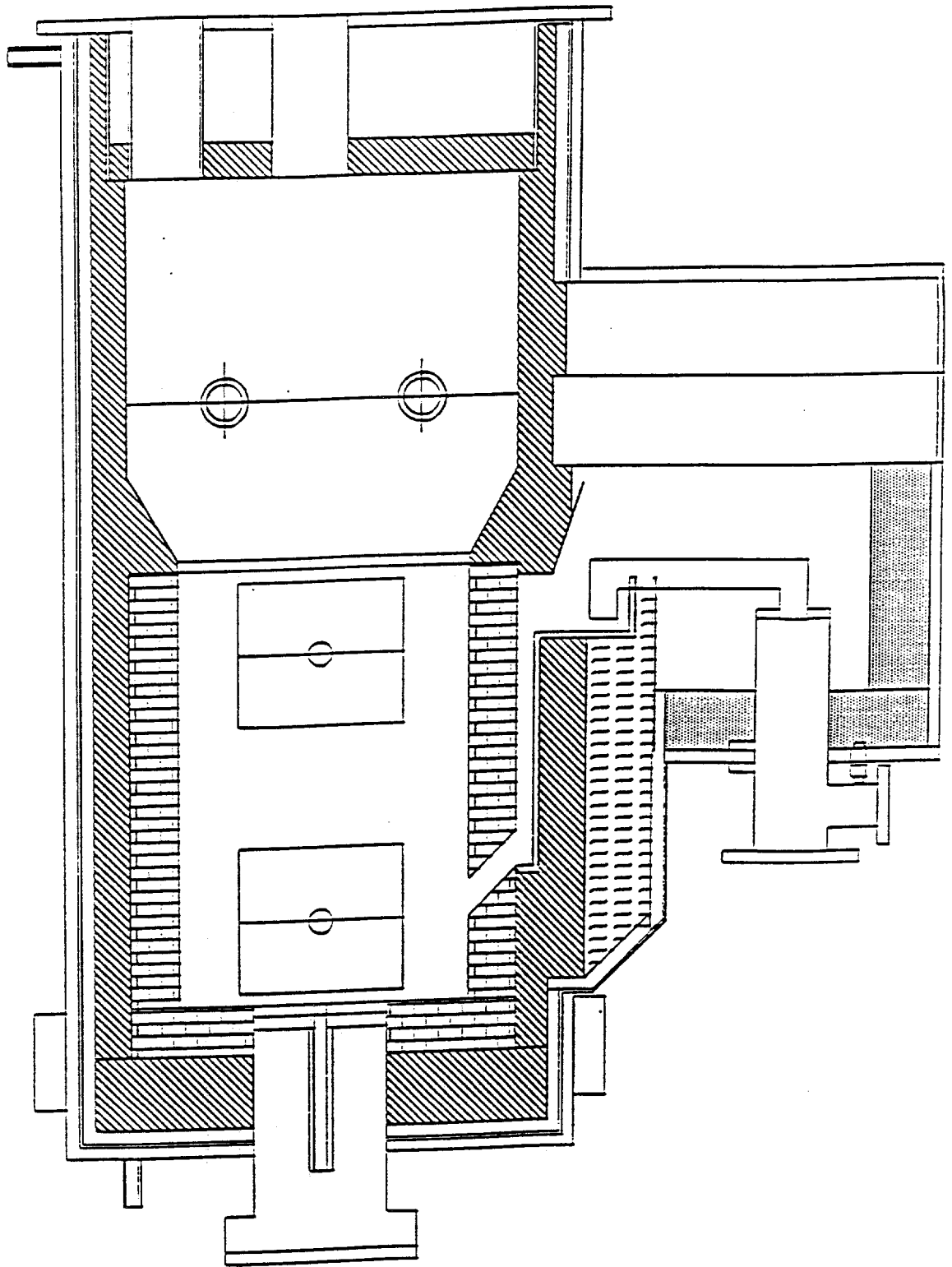
C-5

High pressure drop venturi scrubbers  
Activated carbon absorption  
High efficiency metal fiber filter  
HEPA  
Thermosyphon evaporator, concentrator

Offgas characterization equipment includes particulate size fraction characterization (0.3 to  $>10\mu\text{m}$ ) and compositions. Volatile organics and gases with mass spectrometry.

STATUS:

Current being upgraded for a new slurry feed system and an independent off gas system.



**Fig. C-1. High bay ceramic melter**

C-7

GLASS

**TITLE:** Advanced Test Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Hanford Site  
Richland, WA

**CONTACT:** Chris Chapman: (509) 376-6576  
Pacific Northwest Laboratory

**CAPABILITIES:** Thermal Capacity -- ~750,000 BTU/hr<sup>1</sup>  
Temperature -- 1,550°C  
Throughput -- Inorganic solids: 270-340 kg/d; combustibles:  
400- 800 kg/d<sup>1</sup>; slurries: 300-500 kg/d  
Treatable Waste -- Non-radioactive RCRA hazardous waste

**VERSATILITY:** Waste feeding systems will handle slurries (acid, bases, neutral), sludges, solids (soils, combustible trash, tramp metals). Device is suitable for wastes with large metal pieces.

**DESCRIPTION:** The Advanced Test Melter, shown in Figure C-2, has a rectangular cavity 1.5 ft long by 1 ft wide and is 1 ft deep with a flat floor. The unit has a controllable, freeze valve bottom drain and a conventional bottom take off, overflow discharge. There is a tilt pour mechanism for drain control. The melter is enclosed in a water cooled steel jacket. Electrodes are top entering molybdenum rods with oxidation resistant sheaths. Refractory lining is Monofrax K-3 and E. Feed ports are 6-in. ID on current lid. Larger sizes can be achieved with new lid. Control and data logging equipment are associated with the system.

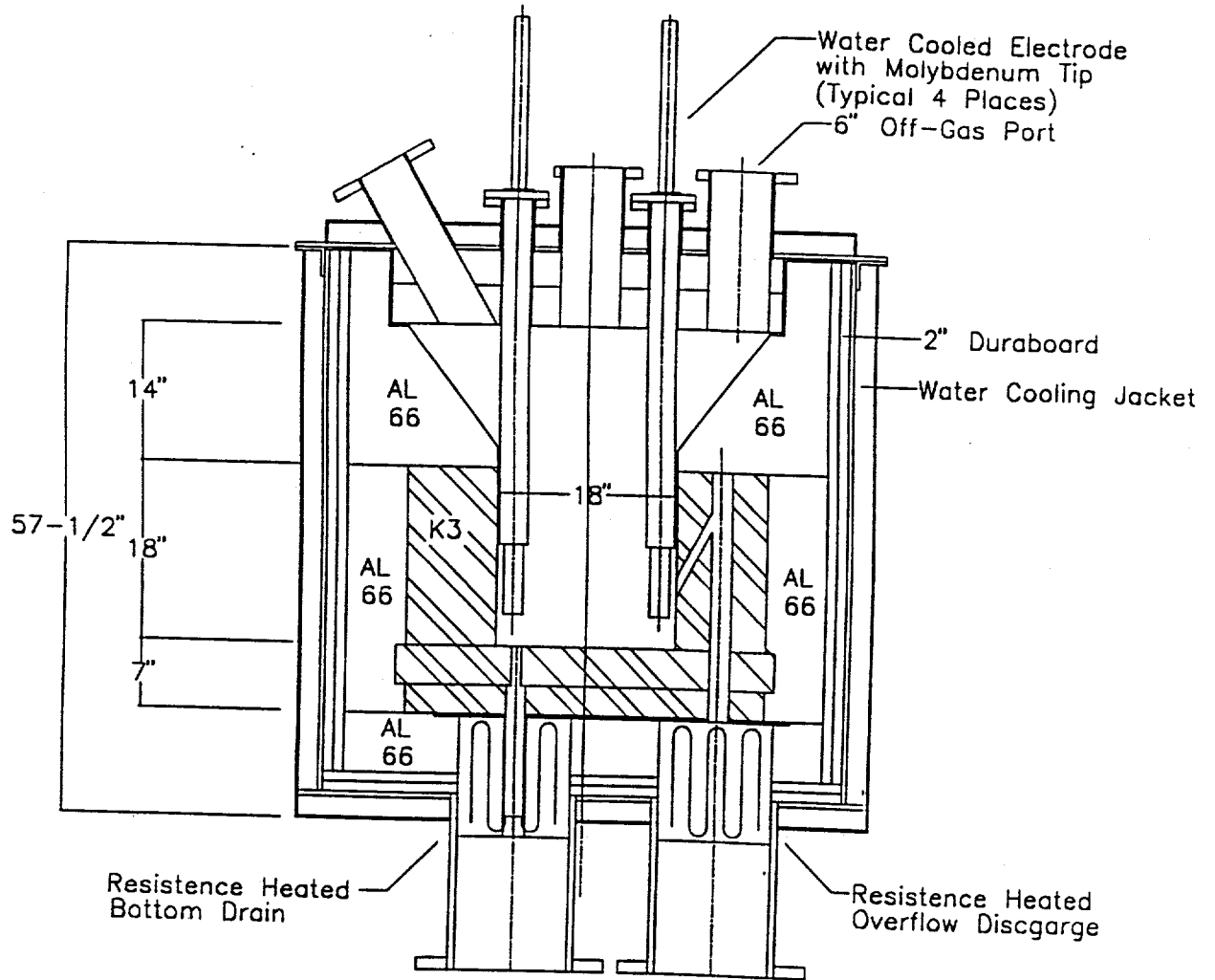
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<sup>1</sup>Existing blower limits this to about 1,200,000 BTU/hr and 1,280 kg/d of 22,000 BTU/kg feed.

OFFGAS SYSTEM: 800 SCFM offgas treatment capacity  
Submerged bed scrubbers  
HEMEs  
Packed column  
Hydrosonic scrubber  
High pressure drop venturi scrubbers  
Activated carbon absorption  
High efficiency metal fiber filter  
HEPA  
Thermosyphon evaporator, concentrator

Offgas characterization equipment includes particulate size fraction characterization (0.3 to  $>10\mu\text{m}$ ) and compositions. Volatile organics and gases with mass spectrometry.

STATUS: Constructed but not in operation.



**Fig. C-2. Advanced test melter (ATM)**

C-10

GLASS

**TITLE:** High Temperature Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Hanford Site  
Richland, WA

**CONTACT:** Chris Chapman: (509) 376-6576  
Pacific Northwest Laboratory

**CAPABILITIES:** Thermal Capacity -- ~6,000,000 BTU/hr<sup>1</sup>  
Temperature -- 1,550°C  
Throughput -- Inorganic Solids: 3,800-4,700 kg/d;  
Combustibles: 5,700-11,400 kg/d<sup>1</sup>; Slurries: 4,200-7,000  
kg/d  
Treatable Waste -- Non-Radioactive RCRA Hazardous Waste

**VERSATILITY:** Waste feeding systems will handle slurries (acid, bases, neutral), sludges, and solids (soils, combustible trash, tramp metals). Device is suitable for wastes with large metal pieces.

**DESCRIPTION:** Melter has a cylindrical cavity 5.2 ft in diameter and is 3.5 ft deep with a sloped floor. Unit has a controllable, freeze valve, bottom drain and a conventional bottom take off, overflow discharge. Reaction chamber above melt can be maintained at a temperature up to 1,000°C with plenum heaters. Melter is enclosed in a water cooled stainless steel jacket. Electrodes are top entering, molybdenum rods (2 sets) with oxidation resistant sheaths. The refractory lining is made of Monofrax K-3 and E. Monofrax electrodes are also tested. Feed ports is up to 24-in. ID. Provision for a central arc boosting or high capacity joule electrode is present. Control and data logging equipment is associated with the system. A diagram of this melter is provided in Figure C-3.

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<sup>1</sup>Existing blower limits this to about 1,200,000BTU/hr and 1,280 kg/d of 22,000 BTU/kg feed.

**OFFGAS SYSTEM:**

800 SCFM offgas treatment capacity  
Submerged bed scrubbers  
HEMEs  
Packed column  
Hydrosonic scrubber  
High pressure drop venturi scrubbers  
Activated carbon absorption  
High efficiency metal fiber filter  
HEPA  
Thermosyphon evaporator, concentrator

Offgas characterization equipment includes particulate size fraction characterization (0.3 to  $>10\mu\text{m}$ ) and compositions. Volatile organics and gases with mass spectrometry.

**STATUS:**

Under construction; due to be in operation 1st quarter of 1993.

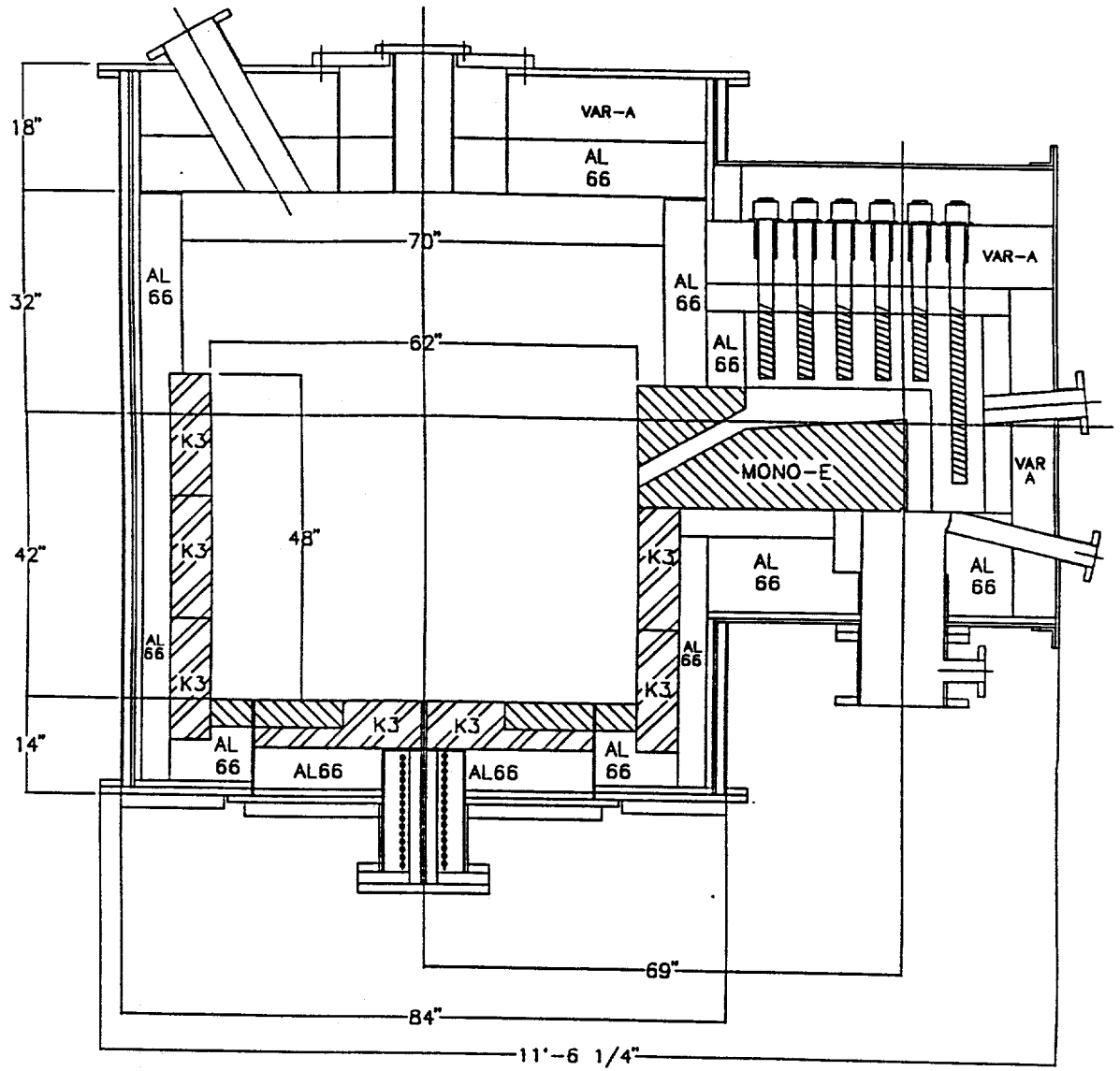


Fig. C-3. High temperature melter



C-13

GLASS

**TITLE:** Liquid Fed Ceramic Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Hanford Site  
Richland, WA

**CONTACT:** Chris Chapman: (509) 376-6576  
Pacific Northwest Laboratory

**CAPABILITIES:** Thermal Capacity -- ~5,600,000 BTU/hr<sup>1</sup>  
Temperature -- 1,200°C  
Throughput -- Inorganic solids: 2,000-2,500 kg/d;  
combustibles: 3,000-6,100 kg/d<sup>1</sup>; slurries: 2,250-3,750 kg/d  
Treatable Waste -- Non-radioactive RCRA hazardous waste

**VERSATILITY:** Waste feeding systems will handle slurries (acid, bases, neutral), sludges, solids (soils, combustible trash, tramp metals). Device is not suitable for wastes with large metal pieces that can not be oxidized in the feed pile.

**DESCRIPTION:** The Liquid Fed Ceramic Melter, shown in Figure C-4, has a rectangular melt cavity 4 ft long by 2.8 ft wide and 1.8 ft deep with a 46° sloped floor. Unit has a batch bottom drain and a conventional bottom take off, overflow discharge. Reaction chamber above melt can be maintained at a temperature up to 1,000°C with plenum heaters. The melter is enclosed in a water cooled stainless steel jacket. The electrodes are side and bottom entering plates (3 total) made of Inconel 690. The refractory lining is Monofrax K-3. Largest feed port is 10-in. diameter. Control and data logging equipment are associated with the system.

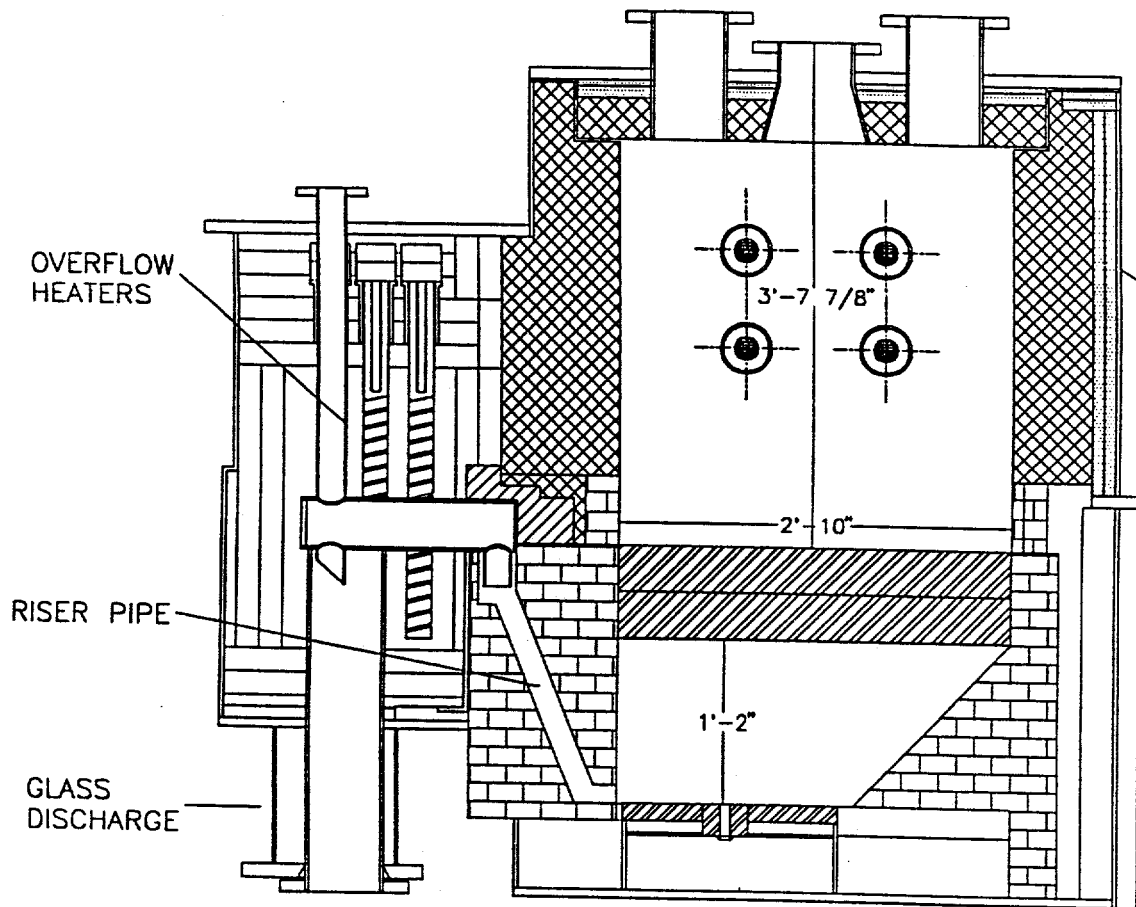
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<sup>1</sup>Existing blower limits this to about 1,200,000 BTU/hr and 1,280 kg/d of 22,000 BTU/kg feed.

**OFFGAS SYSTEM:** 800 SCFM offgas treatment capacity  
Submerged bed scrubbers  
HEMEs  
Packed column  
Hydrosonic scrubber  
High pressure drop venturi scrubbers  
Activated carbon absorption  
High efficiency metal fiber filter  
HEPA  
Thermosyphon evaporator, concentrator

Offgas characterization equipment includes particulate size fraction characterization (0.3 to  $>10\mu\text{m}$ ) and compositions. Volatile organics and gases with mass spectrometry.

**STATUS:** Operational



**Fig. C-4. Liquid fed ceramic melter**

C-16

GLASS

**TITLE:** Pilot Scale Ceramic Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Hanford Site  
Richland, WA

**CONTACT:** Chris Chapman: (509) 376-6576  
Pacific Northwest Laboratory

**CAPABILITIES:** Thermal Capacity -- ~1,800,000 BTU/hr<sup>1</sup>  
Temperature -- 1,200°C  
Throughput -- Inorganic solids: 1,450-1,850 kg/d;  
combustible: 2,200-4,500 kg/d<sup>1</sup>; slurries: 1,650-2,700 kg/d  
Treatable Waste -- Non-radioactive RCRA hazardous waste

**VERSATILITY:** Waste feeding systems will handle slurries (acid, bases, neutral), sludges, and solids (soils, combustible trash, tramp metals). Device is not suitable for wastes with large metal pieces that can not be oxidized in the feed pile.

**DESCRIPTION:** The Pilot Scale Ceramic Melter, shown in Figure C-5, has a rectangular melt cavity 3.4 ft long by 2.4 ft wide and 1.2 ft deep with a flat floor. It has a batch bottom drain and a conventional bottom take off, overflow discharge. The melter is enclosed in a water cooled stainless steel jacket. Electrodes are side entering plates (1 set) made of Inconel 690. The refractory lining is Monofrax K-3. Largest feed port is 12 in. wide by 18 in. long. Control and data logging equipment are associated with the system.

---

<sup>1</sup>Existing blower limits this to about 1,200,000 BTU/hr and 1,280 kg/d of 22,000 BTU/kg feed.

**OFFGAS SYSTEM:**

800 SCFM offgas treatment capacity  
Submerged bed scrubbers  
HEMEs  
Packed column  
Hydrosonic scrubber  
High pressure drop venturi scrubbers  
Activated carbon absorption  
High efficiency metal fiber filter  
HEPA  
Thermosyphon evaporator, concentrator

Offgas characterization equipment includes particulate size fraction characterization (0.3 to  $>10\mu\text{m}$ ) and compositions. Volatile organics and gases with mass spectrometry.

**STATUS:**

Operational

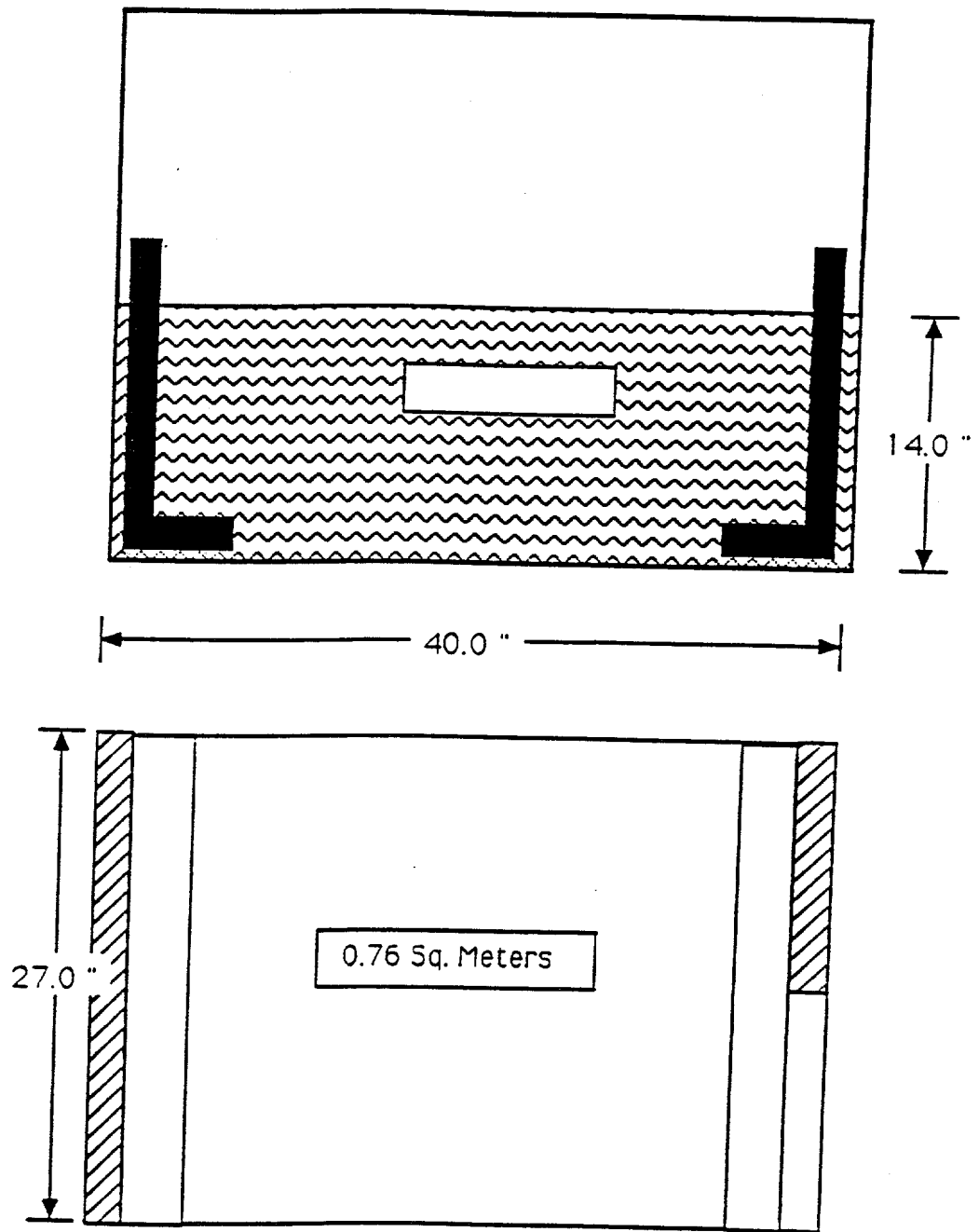


Fig. C-5. Pilot scale ceramic melter

GLASS

**TITLE:** HWTF Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Hanford Site  
Richland, WA

**CONTACT:** Chris Chapman: (509) 376-6576  
Pacific Northwest Laboratory

**CAPABILITIES:** Thermal Capacity -- ~3,400,000 Btu/hr<sup>1</sup>  
Temperature -- 1,550°C  
Throughput -- Inorganic solids: 1,450-1,850 kg/d;  
combustibles: 2,220-4,500 kg/d<sup>1</sup>; slurries: 1,650-2,700 kg/d  
Treatable Waste -- Non-radioactive RCRA hazardous wastes

**VERSATILITY:** Wastes can be solutions, slurries (acid, basic or neutral), or solids (soils, combustible trash or mixtures). Device is suitable for wastes with large metal pieces.

**DESCRIPTION:** The HWTF Melter, shown in Figure C-6, has a rectangular cavity 4 ft long and is 3.5 ft wide, with a glass depth of 2 ft and a sloped floor. The unit has a batch bottom drain and a conventional bottom take off, overflow discharge. Melter is enclosed in a water cooled steel jacket. Electrodes are side entering molybdenum rods (2 total). The refractory lining is Monofrax K-3 and E. Feed port is up to 24 in. wide and 24 in. long.

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<sup>1</sup>Existing blower limits this to about 1,200,000 BTU/hr and 1,280 kg/d of 22,000 BTU/kg feed.

**OFFGAS SYSTEM:** 800 SCFM offgas treatment capacity  
Submerged bed scrubbers  
HEMEs  
Packed column  
Hydrosonic scrubber  
High pressure drop venturi scrubbers  
Activated carbon absorption  
High efficiency metal fiber filter  
HEPA  
Thermosyphon evaporator, concentrator

**STATUS:** Design completed; shell fabricated.



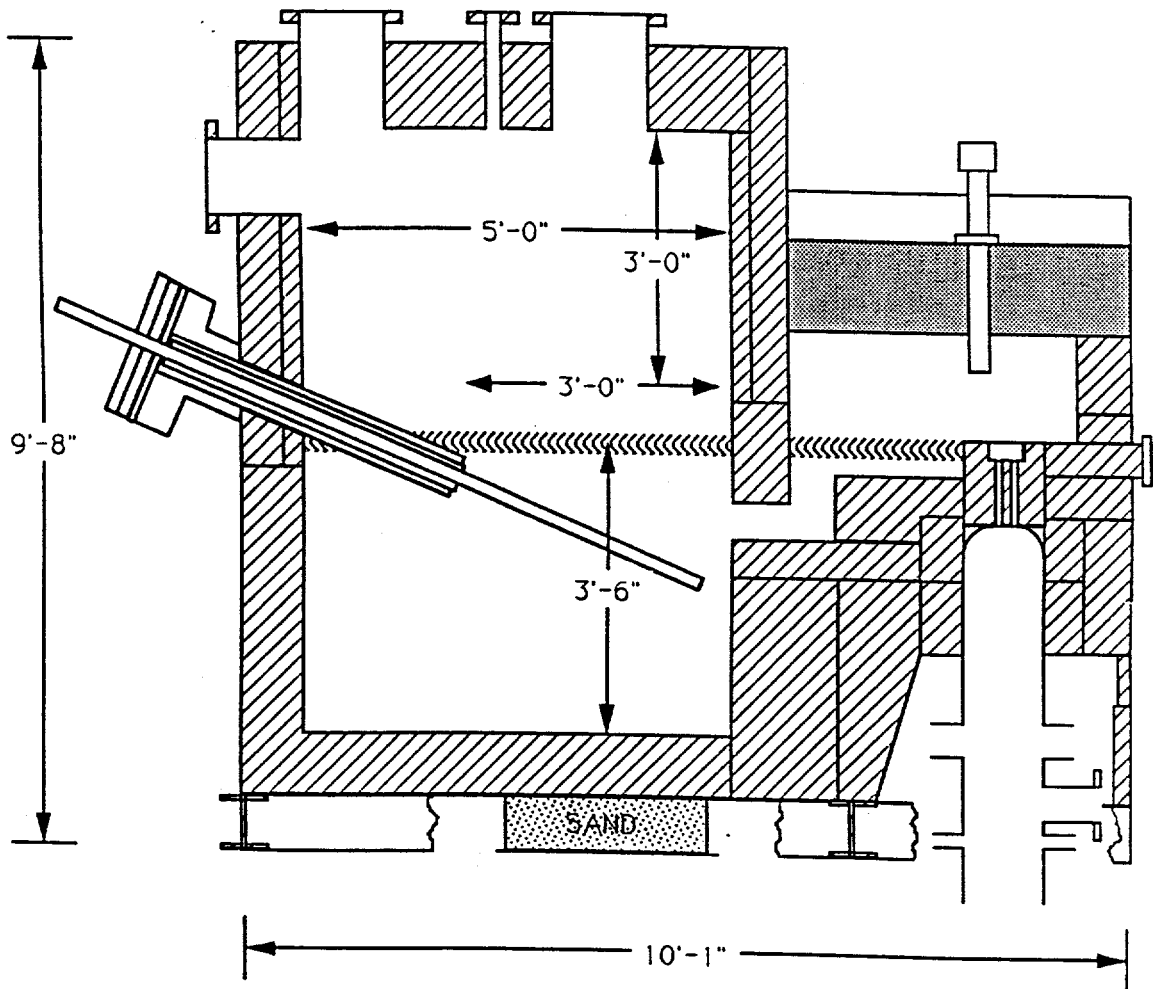


Fig. C-6. HWTF Melter

GLASS

**TITLE:** Fixed Hearth Plasma Process

**EQUIPMENT:** Plasma Arc Furnace

**LOCATION:** Retech, Inc.  
Ukiah, CA

**CONTACT:** Ray L. Geimer: (208) 528-2144  
Science Application International Corporation

**CAPABILITIES:** Thermal Capacity -- 3,000,000 BTU/hr  
Temperature -- 1,800°C  
Maximum Throughput -- 300 kg/hr  
Treatable Waste -- Bulk metals, soils, sludges, solid combustibles, ash, cemented sludges, solid graphite, bricks, concrete blocks. Can perform RCRA Treatability Tests.

**VERSATILITY:** Wastes can be non-homogeneous mixtures of metals, inert solids, and combustibles. Wastes can be feed in whole in 30-gal drums (55-gal drums can be used with some modification to the feed system). Waste with any heating value (0-20,000 BTU/lb) can be treated by this system. Waste can include solid such as bricks, concrete, or bulk metal objects. Sludge wastes and partially (or completely) solidified sludges can be processed.

**DESCRIPTION:** The primary furnace chamber has a 1.2 MW plasma torch that can be operated on He, Ar, N, O, air, or mixtures of these gases. The feed chamber is large enough to accommodate whole 30-gal drums and could be easily modified to accommodate 55-gal drums. The system operates in a batch mode and can operate for 2-4 hrs depending on feed composition. The primary furnace is a water cooled, refractory lined chamber capable of vacuum tight operation at temperatures up to 1,800°C. The secondary combustion chamber is a water cooled, refractory lined vessel capable of operating at temperatures up to 1,400°C. The secondary combustion chamber has an auxiliary natural gas burner that will deliver up to 1,000,000 BTU/hr. Combustion air is supplied to the primary and secondary chamber by a 7.5-hp blower rated at 1,000 scfm.

**OFFGAS SYSTEM:**

Gases exit the secondary combustion chamber and are immediately quenched by an air atomized water spray. Warm dry gases (200°C) exit the evaporative cooler and are filtered in a high temperature baghouse (51 m<sup>2</sup> of filter area). Flow of gas is induced by a 20-hp fan with a capacity at 5,000 scfm at 10-in. water column.

**STATUS:**

Operational

C-24

GLASS

**TITLE:** Joule-Heated Research Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Savannah River Site  
Aiken, SC

**CONTACT:** Dennis Bickford: (803) 725-3737  
Westinghouse Savannah River Company

**CAPABILITIES:** Thermal Capacity -- 8.4 kW  
Temperature -- 1,200°C  
Maximum Throughput -- 2 lb/hr (dry feed @ 50%)  
Treatable Waste -- simulated waste slurry and hazardous materials

**VERSATILITY:** Installed in chemical hood to permit handling of various toxic materials. Full lid heating capacity. Normally used with mercury, silver, lead, selenium, etc. Frequently used for waste solubility, waste form quality, glass redox control measurements. Major changes in batch chemistry are tried here before tests in pilot scale melters. Tilt to pour mechanism. Third generation, 15 yrs experience, repaired and replaced as necessary.

**DESCRIPTION:** The Research Melter is used for electric Joule heating of slurry which can contain hazardous materials, including mercury for melting temperatures up to 1,200°C. The enclosed melter system utilizes Inconel 690 electrodes and includes the complete offgas system. The melter produces 1-3 lb/hr output. No additional permits are required for experimental simulated waste.

**OFFGAS SYSTEM:** Fully scaled down DWPF system:  
Venturi scrubber  
Steam atomized scrubber and cyclone  
Condenser and demister  
HEPA filtration  
Filter disk and bubble scrubbers for special offgas analysis

**STATUS:** Operational

C-25

GLASS

**TITLE:** High Level Caves -- Joule-Heated Research Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Savannah River Site  
Aiken, SC

**CONTACT:** Dennis Bickford: (803) 725-3737  
Westinghouse Savannah River Company

**CAPABILITIES:** Thermal Capacity -- 8.4 kW  
Temperature -- 1150°C  
Maximum Throughput -- 2 lb/hr (dry feed @ 50%)  
Treatable Waste -- Radioactive waste slurry and hazardous materials

**VERSATILITY:** Full lid heating capacity. Installed in high-level cell to permit handling of high-level waste and various toxic materials. Frequently used for high-level waste verification tests. Third generation, 15 years operation, rebuilt and replaced as necessary. Tilt to pour mechanism.

Same facility has capability of performing leach testing, ICP-AES for elemental analysis, and radiological testing.

**DESCRIPTION:** Can vitrify radioactive waste slurry in a Research Melter; very similar to the Joule-Heated Research Melter described above.

**OFFGAS SYSTEM:** Venturi scrubber  
Condenser and demister  
HEPA filtration  
Filter disk and bubbler scrubbers for special offgas analyses

**STATUS:** Operational

GLASS

**TITLE:** High Temperature Melter

**EQUIPMENT:** Induction Melter

**LOCATION:** Savannah River Site  
Aiken, SC

**CONTACT:** Dennis Bickford: (803) 725-3737  
Savannah River

**CAPABILITIES:** Thermal Capacity -- N/A  
Temperature -- 1,900°C  
Throughput -- 20 lb/hr  
Treatable Waste -- Low-level mixed waste

**VERSATILITY:** Rapid heating to desired temperature, with maximum temperature about 2,000°C. Replaceable melt container. Has been used with depleted, natural, and enriched uranium. Tilt to pour. Especially applicable for tests where molten steel, stainless steel, iron, etc., are produced. Ideal for metal recycling and reclamation test. Temperatures are better controlled than with arc or plasma melting, minimizing volatilization of low boiling point metals such as cadmium and zinc. Can be fitted with air lance or oxygen lance for oxidation of organics. Has induction stirring for homogenization and improved metal recovery. A similar unit was used for production of plutonium alloys (design of containment facility available).

**DESCRIPTION:** Used for the induction melting of electrically insulating or conductive materials, and their mixtures. Equipment is located in a lab with a 2-in. metal shear and other furnaces operating from 1,000-1,600°C. Permitted for low-level radioactive testing and hazardous materials, with a limit of 500 grams enriched uranium. Similar unit could be made transportable. Off the shelf scale up to 50 tons/day.

**OFFGAS SYSTEM:** Exhauster, double HEPA filtration and stack. Licensed for use with LLMW.

**STATUS:** Installed and available. Offgas upgrade scheduled for January 1993.

C-27

GLASS

**TITLE:** Pilot Scale Joule-Heated Melter at the TNX Area

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Savannah River Site  
Aiken, SC

**CONTACT:** Dennis Bickford: (803) 725-3737  
Westinghouse Savannah River Company

**CAPABILITIES:** Thermal Capacity -- 56 kW  
Temperature -- 1,150°C  
Throughput -- 10 lb/hr  
Treatable Waste -- Simulated waste streams and hazardous materials

**VERSATILITY:** A \$16,000,000 fully integrated melter system duplicating DWPF process. All major feed preparation and offgas scrubbing equipment are run using instrumental controls similar to those of the DWPF. Most feed preparation and offgas system components are made of Hastelloy C-276, permitting full flowsheet amounts of corrosive acids, halides, sulfates, etc. Normally operates with Hg, Ag, Se, Cd, Pb, Cr, Cu, H<sub>2</sub>SO<sub>4</sub>, HNO<sub>3</sub>, etc.

**DESCRIPTION:** Complete pilot-scale facilities are available for pre-processing the waste stream, Joule-heat glass melting, process control, and waste glass evaluation. Expertise includes process design and modification, process evaluation, process evaluation, process control, process modeling and offgas design. Permits include mercury and hazardous materials.

**OFFGAS SYSTEM:** Extensive instrumentation of offgas system and feed preparation offgas allows monitoring for CO, CO<sub>2</sub>, H<sub>2</sub>, NO, N<sub>2</sub>O, SO<sub>2</sub>, and organics.

**STATUS:** Operating now.

GLASS

**TITLE:** High Temperature Glass Melter

**EQUIPMENT:** TECO Joule Heated Melter

**LOCATION:** DOE Industrial Center for Vitrification  
Clemson, SC

**CONTACT:** Dennis Bickford: (803) 648-6317  
Westinghouse Savannah River Company

**CAPABILITIES:** Thermal Capacity -- 100 kW  
Temperature --1,600°C  
Maximum Throughput -- 120 lb/hr  
Treatable Waste -- Simulated wastes and hazardous materials

**VERSATILITY:** Especially designed for rapid and inexpensive disassembly and reconfiguration (one week turn around time). Is being modified to permit recovery of molten copper alloys form circuit boards. Will be used to examine alternative electrode materials, and a range of oxidation reduction conditions.

**DESCRIPTION:** The melter was especially designed for hazardous waste vitrification by one of the world's largest suppliers of commercial electric melter. It is portable and has been used at 3 different sites for pilot scale tests. The manufacturer can supply similar unity with capacities of over 10 tons/day. A high intensity electric field provides locally high temperatures, while maintaining the low pollution characteristics of a cold top melter. Necessary permits are being obtained by Clemson University to operate for Sandia National Laboratories and Savannah River Site simulated waste.

**OFFGAS:** Portable and fully integrated DWPF system with  
Quencher/scrubber  
Offgas condensate tank  
Atomized supersonic scrubber and cyclone  
Condenser and mist Eliminator  
Reheater and HEPA Filtration  
Exhauster  
Stack (not portable)  
The offgas system is arranged with the necessary taps for particulate and gas emission measurements.

**STATUS:** Ready to ship to Clemson November 1992.



GLASS

**TITLE:** Lab Scale Stirred Glass Melter

**EQUIPMENT:** Glasstech 1/4 ft<sup>2</sup> Stirred Melter

**LOCATION:** DOE Industrial Center for Vitrification  
Clemson, SC

**CONTACT:** Dennis Bickford: (803)648-6317  
Westinghouse Savannah River Company

**CAPABILITIES:** Thermal Capacity -- 15 kW  
Temperature -- 1,200°C  
Maximum Throughput -- 8 lb/hr  
Treatable Waste -- Simulated wastes, wastes slurries, and hazardous materials

**VERSATILITY:** Highest throughput per unit size of any joule heated melter. Especially designed for low cost, rapid and inexpensive disassembly and reconfiguration (one week turn around time). Is being used to investigate direct vitrification of wastes under a range of oxidation reduction conditions, and for development of melter sensors.

**DESCRIPTION:** The melter was especially designed for hazardous waste vitrification in cooperation with the Savannah River Site. The manufacturer can supply similar units with capacities of over 6 tons/day. Stirring provides exceptionally high melting rates while minimizing the temperature and surface area available for volatilization of waste components. Can be operated as a batch process to partially maintain the low pollution characteristics of a cold top melter. Necessary permits are being obtained by Clemson University to operate for Sandia National Laboratory and Savannah River Site simulated waste.

**OFFGAS SYSTEM:** Normally vented to laboratory hood, can be fitted with necessary scrubbers and samplers for specific tests.

**STATUS:** Ready for shipment November 1992.

C-30

GLASS

**TITLE:** Production Scale Stirred Glass Melter

**EQUIPMENT:** Glasstech 9 ft<sup>2</sup> Stirred Melter

**LOCATION:** Savannah River Site  
Aiken, SC

**CONTACT:** Technical -- Dennis Bickford: (803) 648-6317  
Westinghouse Savannah River Company  
Commercial -- Ray Richards (419) 536-8828  
Kenneth Wetmore (419) 661-9500  
Stir-Melter, Inc.

**CAPABILITIES:** Thermal Capacity -- 600 kW  
Temperature -- 1,200°C  
Maximum Throughput -- 228 lb/hr slurry; 500 lb/hr dry feed  
Treatable Waste -- Simulated wastes, wastes slurries, and hazardous materials

**VERSATILITY:** Highest throughput per unit size of any joule heated melter. Especially designed for low cost, rapid and inexpensive disassembly and reconfiguration (one month turn around time). Is being used to investigate less expensive and more easily repaired melters for high level waste. Is available part time for non-HLW programs. Will be used for direct vitrification of wastes under a range of oxidation reduction conditions, and for development of melter sensors.

**DESCRIPTION:** The melter was especially designed for hazardous waste vitrification in cooperation with Savannah River Site. The manufacturer can supply similar units with capacities of over 6 tons/day. Stirring provides exceptionally high melting rates while minimizing the temperature and surface area available for volatilization of waste components. Can be operated batch-wise to partially maintain the low pollution characteristics of a cold top melter. Necessary permits are in place to operate for Savannah River Site waste simulants, including actual hazardous components.

**OFFGAS SYSTEM:**

A \$16,000,000 fully integrated melter system duplicating DWPF process. All major feed preparation and offgas scrubbing equipment is run using similar instrumental controls as the DWPF. Most feed preparation and offgas system components are made of Hastelloy C-276, permitting full flowsheet amounts of corrosive acids, halides, sulfates, etc. Extensive instrumentation of offgas system and feed preparation offgas allows monitoring for CO, CO<sub>2</sub>, H<sub>2</sub>, NO, NO<sub>2</sub>, N<sub>2</sub>O, SO<sub>2</sub>, and organics. Normally operates with Hg, Ag, Se, Cd, Pb, Cr, Cu, H<sub>2</sub>SO<sub>4</sub>, HNO<sub>3</sub>, etc.

**STATUS:**

Finalizing purchasing contract.

C-32

GLASS

**TITLE:** Lab Scale Rotary Calciner/Melter

**EQUIPMENT:** Custom Laboratory Rotary Calciner/Melter

**LOCATION:** Savannah River Site  
Aiken, SC

**CONTACT:** Dennis Bickford: (803) 648-6317  
Westinghouse Savannah River Company

**CAPABILITIES:** Thermal Capacity -- 3 kW  
Temperature -- 1,300°C  
Maximum Throughput -- 0.5 lb/hr  
Treatable Waste -- LLMW, waste slurries, and hazardous materials

**VERSATILITY:** Very small unit for chemical characterization of waste and offgas. Especially designed for low cost operation and inexpensive disassembly and reconfiguration. It is being used to calcine wastes prior to vitrification in crucibles, and for offgas sampling.

**DESCRIPTION:** The melter was especially designed for hazardous waste oxidation and vitrification by Savannah River Site.

**OFFGAS SYSTEM:** It is normally fitted with condenser and vented to hood. Can be fitted with glass pouring, offgas sampling as required.

**STATUS:** Built and ready for installation.

GLASS

**TITLE:** WV-0.25 Stir-Melter, Inc.

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Stir-Melter, Inc.  
Perrysburg, OH

**CONTACT:** Technical: Ray Richards -- (419) 536-8828  
Commercial: Kenneth Wetmore -- (419) 661-9500

**CAPABILITIES:** Thermal Capacity -- 10 kW  
Temperature -- 1,050°C  
Maximum Throughput<sup>1</sup> -- 120 kg/d (glass output from dry feed); 80 kg/d (glass output from 60% slurry)  
Treatable Waste -- RCRA hazardous waste

**VERSATILITY:** Stir-Melters are highly driven refractory or metal lined Joule-heated glass melters. They are much smaller in size than most other melters; quickly started and stopped, and offgases can be limited to those emanating from the feed materials. Feed materials tested include 60% water slurries, dry feed including modest amounts of organics, and light materials such as fiberglass scrap. The operating temperature limits for the metal lined melters in 1,050°C. A second stage non-stirred superheater operating at 1,150°C can be provided. The refractory lined units can operate at over 1,600°C.

**DESCRIPTION:** This is a 0.25 ft<sup>2</sup> area melter, designed specifically for simulated radioactive and hazardous wastes. It is an all metal, partially sealed unit with an upper temperature limit of 1,050°C.

**OFFGAS SYSTEM:** The size of this unit lends itself easily to glove-box enclosure for offgas containment. Stir-Melter, Inc. can perform melting trials. Stack sampling and testing can be obtained locally. Other testing and characterization is provided by other vendors.

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<sup>1</sup>Lab EPA permit limits 250 kg of waste per day and a total limit of 1000 kg for any one waste stream. Radioactive materials cannot be accepted.

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**STATUS:**

The piece of equipment will be available in the fall of 1992 for demonstration through Stir-Melter, Inc. in its laboratory facility.

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GLASS

**TITLE:** WV-1 Stir Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Stir Melter, Inc.  
Perrysburg, OH

**CONTACT:** Technical -- Ray Richards: (419) 536-8828  
Commercial -- Kenneth Wetmore: (419) 661-9500

**CAPABILITIES:** Thermal Capacity -- 60 kW  
Temperature -- 1,050°C  
Maximum Throughput<sup>1</sup> -- 500 kg/d (glass output from dry feed); 350 kg/d ( glass output from 60% slurry)  
Treatable Waste -- RCRA hazardous waste

**VERSATILITY:** Stir-Melters are highly driven refractory or metal lined Joule-heated glass melters. They are much smaller in size than most other melters; quickly started and stopped, and offgases can be limited to those emanating from the feed materials. Feed materials tested include 60% water slurries, dry feed including modest amounts of organics, and light materials such as fiberglass scrap. The operating temperature limits for the metal lined melters in 1,050°C. A second stage non-stirred superheater operating at 1,150°C can be provided. The refractory lined units can operate at over 1,600°C.

**DESCRIPTION:** This is a 1 ft<sup>2</sup> area melter, designed specifically for simulated radioactive and hazardous wastes. It is an all metal, sealed unit with an upper temperature limit of 1,050°C. It has been tested for over 3,700 hrs. Much of this time was spent on simulated high-level waste from the Savannah River Site. This was a 60 weight % water/solids slurry and the melter produced over 14.5 kg/hr of glass output. A dry feed of similar composition was also tested and produced a melting rate of over 21.4 kg/hr.

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<sup>1</sup>Lab EPA permit limits 250 kg of waste per day and a total limit of 1000 kg for any one waste stream. Radioactive materials cannot be accepted.

**OFFGAS SYSTEM:**

Stir-Melter, Inc. units are equipped with an offgas port which is typically connected to offgas handling equipment specifically designed for each application and supplied by the client or other vendors. Stir-Melter, Inc. can perform melting trials. Stack

sampling and testing can be obtained locally. Other testing and characterization is provided by other vendors.

**STATUS:**

This piece of equipment is available for demonstration through Stir-Melter, Inc. in its laboratory facility.



C-37

GLASS

**TITLE:** MiniMelter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Vitreous State Laboratory of Catholic University of America  
Washington, DC

**CONTACT:** Pedro B. Macedo: (202) 319-5329  
Vitreous State Laboratory

**CAPABILITIES:** Thermal Capacity -- 8 kW  
Temperature -- 1,150°C  
Throughput -- 10 kg/d to 30 kg/d  
Treatable Waste -- RCRA hazardous and low-level  
radioactive waste

**VERSATILITY:** Waste feeds can be dry, semi-dry, slurry, or liquid.

**DESCRIPTION:** MiniMelter is the smallest vitrification furnace with nominal capacity of 10 kg/day. This is an electrically heated furnace allowing small scale studies on various feed materials. The MiniMelter operates at temperatures not higher than 1,200°C. It is equipped with a system for continuous feeding of dry and semi-dry materials as well as wet sludges. It possesses an off-gas system for full capture of evaporation products. The MiniMelter operates in the batch mode.

For some typical waste processed at Vitreous State Laboratory, this furnace could process up to 30 kg/day.

**OFFGAS SYSTEM:** Oil bucket particulate collector  
HEPA filtration system

**STATUS:** --

C-38

GLASS

**TITLE:** MicroMelter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Vitreous State Laboratory of Catholic University of America  
Washington, DC

**CONTACT:** Pedro B. Macedo: (202) 319-5329  
Vitreous State Laboratory

**CAPABILITIES:** Thermal Capacity -- 12 kW  
Temperature -- 1,150°C  
Throughput -- 30 kg/d to 100 kg/d  
Treatable Waste -- RCRA hazardous and low-level  
radioactive waste

**VERSATILITY:** Waste feeds can be dry, semi-dry, slurry, or liquid.

**DESCRIPTION:** MicroMelter is a larger unit with all characteristics of a pilot-plant installation. Its nominal capacity is 30 kg/day and it allows quick, inexpensive vitrification studies on all kinds of feed materials at temperatures not higher than 1,200°C. The MicroMelter is also an electrically-heated furnace. It is equipped with a proprietary mixing system which ensures rapid and uniform mixing of waste materials and glass formers. The MicroMelter can safely handle low-level radioactive wastes. It operates in the continuous mode and possesses a relatively sophisticated offgas system.

For some typical wastes processed at Vitreous State Laboratory, this furnace could process up to 100 kg/day of waste.

**OFFGAS SYSTEM:** Quencher  
Scrubber  
Bag filter  
HEPA filtration system

**STATUS:** --

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GLASS

**TITLE:** 100 kg/day Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Vitreous State Laboratory of Catholic University of America  
Washington, DC

**CONTACT:** Pedro B. Macedo: (202) 319-5329  
Vitreous State Laboratory

**CAPABILITIES:** Thermal Capacity -- 42 kW  
Temperature -- 1,150°C  
Throughput -- 100 kg/day to 900 kg/day  
Treatable Waste -- RCRA hazardous and low-level  
radioactive waste

**VERSATILITY:** Waste feeds can be dry, semi-dry, slurry, or liquid.

**DESCRIPTION:** When operational, the 100 kg/day Melter will possess all of the advanced features of the MiniMelter, but will allow vitrification studies at a larger scale. It will be capable of handling low-level radioactive wastes.

For some typical wastes processed at the Vitreous State Laboratory, this furnace is capable of processing up to 900 kg/day of waste.

**OFFGAS SYSTEM:** The offgas system consists of a quencher, packed bed scrubber, mist eliminator, bag filter, and HEPA filtration unit. Both liquid and gaseous waste streams are fully recyclable.

**STATUS:** 100 kg/day Melter is currently in the construction stage and will become operational at the beginning of October 1992.

C-40

GLASS

**TITLE:** 1,000 kg/day Melter

**EQUIPMENT:** Joule-Heated Melter

**LOCATION:** Vitreous State Laboratory of Catholic University of America  
Washington, DC

**CONTACT:** Pedro B. Macedo: (202) 319-5329  
Vitreous State Laboratory

**CAPABILITIES:** Thermal Capacity -- 125 kW  
Temperature -- 1,150°C  
Throughput -- 1,000 kg/day to 5,000 kg/day  
Treatable Waste -- Asbestos-contaminated and simulated  
radioactive wastes

**VERSATILITY:** Waste feeds can be dry, semi-dry, slurry, or liquid.

**DESCRIPTION:** The 1,000 kg/day Melter will provide yet another intermediate step in the cost-effective scale-up studies which will lead to the construction and utilization of the full-scale vitrification plants. The 1,000 kg/d unit will be able to process asbestos-containing waste streams and simulated radioactive wastes.

It is expected that for the typical wastes processed at Vitreous State Laboratory that the furnace would be capable of processing up to 5,000 kg/day of waste.

**OFFGAS SYSTEM:** The offgas system consists of the following major components connected in series: film cooler, quencher, packed bed spray tower, horizontal mist eliminator, reheater, bag filter, and HEPA unit. Both gaseous and liquid wastes streams are generated. The gaseous stream is either recycled or disposed of directly or after ion exchange depending on composition. The offgas is constantly monitored on the operational, safety and environmental levels. Sensors are wired through the A&B PLC controller with safety level being interlocked to the system feed.

**STATUS:** 1,000 kg/day Melter is in the final stage of design and will become operational in the late fall of 1992.

HYDRAULIC CEMENT

**TITLE:** Ashcrete Process

**EQUIPMENT:** End-over-end drum tumbler from Stock Equipment Company

**LOCATION:** Savannah River Laboratory

**CONTACT:** Don Fisher (803/725-6428)

**CAPABILITIES:** The Ashcrete Process solidifies ash generated by the Beta Gamma Incinerator at Savannah River Plant and produces a cement waste form acceptable for burial. It processes ash within the same drum received from the incinerator.

**VERSATILITY:** The in-container mixing system used makes the equipment versatile and amenable to also solidify liquid and slurry wastes with minor modifications.

**DESCRIPTION:** At Station 1, the ash-filled drum is weighed and two iron mixing bars are placed inside it. The requisite water addition is made at Station 2 and cement and sand are added at Station 3. The cap is replaced at Station 4 and the end-over-end tumbler mechanism at Station 5 clamps the drum and rotates it at 18 rpm. This mixes the ashcrete without any physical contact between the process equipment and the contaminated ash.

**OFFGAS SYSTEM:** Other than the routine air-balance control, there is no offgas system required.

**STATUS:** Operational

HYDRAULIC CEMENT

**TITLE:** Transportable Grout Equipment Facility

**EQUIPMENT:** Twin-screw variable speed in-line mixer and progressive cavity grout pump

**LOCATION:** Hanford Site

**CONTACT:** J. A. Voogd (509) 373-5642

**CAPABILITIES:** Provides for the remotely-operated mixing of selected liquid wastes with the dry grout solids and delivery at 30-70 GPM of the resulting slurry to the disposal vaults where solidification occurs.

**VERSATILITY:** Like most cement systems, this Grout Facility can adjust its dry grout-forming solids ratios to match the needs of changing input waste streams. All process control is performed by a programmable logic controller. There is an optimum balance of automatic control and operator initiated control. The operator interface with the control system consists of a set of keyboards and CRT monitors. The CRTs provide menus to assist the operator in performing the control actions and in selecting different screens for monitoring different portions of the process.

**DESCRIPTION:** The grout facility was constructed in eight modules which provide for the remote waste mixing and pumping and all necessary process support and control. Inside the Mixer Module the dry blend material is mixed with the radioactive waste stream in a twin-screw variable speed in-line mixer. Liquid process additives are metered into the waste stream just prior to its entry into the mixer. The mixed grout slurry is deposited in a surge tank which provides a constant supply of material for the grout pump.

**OFFGAS SYSTEM:** The exhaust system draws air at 850 scfm from the Mixer Module and the baghouse dust collection system which ventilates the grout surge tank. After passing through dual HEPA filtration trains, the filtered air is sent to a stack for monitoring and release.

**STATUS:** Operational

SULFUR POLYMER CEMENT

**TITLE:** Modified Sulfur Cement Encapsulation Process

**EQUIPMENT:** Modified Sulfur Cement Process Equipment

**LOCATION:** Brookhaven National Laboratory  
Upton, NY

**CONTACT:** Paul Kalb: (516) 282-7644  
Brookhaven National Laboratory

**CAPABILITIES:** Thermal Capacity -- N/A  
Temperature -- 120 to 130°C  
Maximum Throughput -- 1 to 2 kg/hr for bench-scale process  
Treatable Waste -- RCRA hazardous and low-level  
radioactive waste

**VERSATILITY:** Treat a wide variety of waste streams including salts and  
other aqueous wastes, sludges, incinerator ash, evaporator  
concentrates, dry solids, and ion exchange resins. Can be  
easily scaled to pilot or production scale.

**DESCRIPTION:** Bench-scale processing equipment for the modified sulfur  
cement encapsulation process include a double planetary  
orbital mixer with jacketed heating/mixing vessel, vacuum  
hood, oil bath circulation heater, and hydraulic platen  
discharge unit.

**OFFGAS SYSTEM:** A HEPA filter enclosure has been constructed to enhance  
capabilities to process radioactive and hazardous materials.

**STATUS:** Bench-scale process operational.

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CERAMIC

**TITLE:** Ceramic Fabrication Equipment

**EQUIPMENT:**

**LOCATION:** Lawrence Livermore National Laboratory  
Livermore, CA

**CONTACT:** Virginia Oversby:  
Lawrence Livermore National Laboratory

**CAPABILITIES:** Thermal Capacity -- N/A  
Temperature -- N/A  
Nominal Throughput -- N/A  
Treatable Waste -- RCRA Hazardous and Radioactive Waste

**VERSATILITY:** N/A

**DESCRIPTION:** The equipment available at Lawrence Livermore National Laboratory for waste form production and testing includes controlled atmosphere glove boxes for handling radioactive and hazardous materials, high temperature furnaces for sintering operations, presses of various sizes and pressure capabilities for both hot and cold pressing, and the full range of chemical analysis capabilities required to support EPA test protocols.

**OFFGAS SYSTEM:** N/A

**STATUS:** Operational



ORGANIC BINDER

**TITLE:** Polyethylene Encapsulation Process Equipment

**EQUIPMENT:** Polyethylene Screw Extruder

**LOCATION:** Brookhaven National Laboratory  
Upton, NY

**CONTACT:** Paul Kalb: (516) 282-7644  
Brookhaven National Laboratory

**CAPABILITIES:** Thermal Capacity -- N/A  
Temperature -- N/A  
Maximum Throughput -- N/A  
Treatable Waste -- RCRA hazardous and low-level  
radioactive waste

**VERSATILITY:** Treat a wide variety of waste streams including salts and  
other aqueous wastes, sludges, incinerator ash, evaporator  
concentrates, dry solids, and ion exchange resins.

**DESCRIPTION:** Polyethylene encapsulation process equipment include two  
bench-scale single screw extruders, several volumetric  
feeders for precise metering of waste and binder, waste form  
specimen molds and miscellaneous pre-treatment equipment.  
A production-scale extruder has been procured and will be  
sintalled at BNL for the purpose of scale-up studies and full-  
scale process demonstration.

**OFFGAS SYSTEM:** A HEPA filter enclosure has been constructed to enhance  
capabilities to process radioactive and hazardous materials.

**STATUS:** Bench-scale operational. A full scale test facility is planned  
for FY-93.

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**APPENDIX D**  
**SUMMARIES FOR FINAL WASTE FORMS**



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**GLASS SUMMARIES**

D-4

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**GLASS SUMMARIES INDEX**

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- |       |                     |   |
|-------|---------------------|---|
| NO. 1 | <b>AUTHOR:</b>      |   |
|       | <b>TITLE:</b>       | Summary of Properties of Borosilicate Waste Glasses |
|       | <b>PUBLICATION:</b> |   |
|       | <b>DATE:</b>        |   |
- 
- |       |                     |  |
|-------|---------------------|--|
| NO. 2 | <b>AUTHOR:</b>      |  |
|       | <b>TITLE:</b>       | A State-of-the-Art Review of Materials Properties of Nuclear Waste Forms |
|       | <b>PUBLICATION:</b> | USDOE, PNL-3802 UC-70  |
|       | <b>DATE:</b>        | 4/1/81   |
- 
- |       |                     |  |
|-------|---------------------|--|
| NO. 3 | <b>AUTHOR:</b>      |  |
|       | <b>TITLE:</b>       | Vitrification Technologies for Treatment of Hazardous and Radioactive Waste                    |
|       | <b>PUBLICATION:</b> | EPA/625/R-92/002 U.S. EPA, Center for Environmental Research Information, Cincinnati, OH 45268 |
|       | <b>DATE:</b>        |  |
- 
- |       |                     |   |
|-------|---------------------|---|
| NO. 4 | <b>AUTHOR:</b>      | Andrews, M.K.; Bibler, N.E.; Jantzen, C.M.; Beam, D.C.  |
|       | <b>TITLE:</b>       | Initial Demonstration of the DWPF Vitrification Process and Product Control Strategy Using Actual Radioactive Waste |
|       | <b>PUBLICATION:</b> | Ceramic Transactions, Nuclear Waste Management IV, Vol 23, p. 569-576   |
|       | <b>DATE:</b>        | 4/29/91   |
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- |       |                     |   |
|-------|---------------------|---|
| NO. 5 | <b>AUTHOR:</b>      | Bates, S.O.; Bowen, W.M.  |
|       | <b>TITLE:</b>       | Interim Milestone HWVP-86-V1122C-Report on Composition Variability Testing Conducted for the HWVP |
|       | <b>PUBLICATION:</b> | HWVP-86-V1122C Final Report   |
|       | <b>DATE:</b>        | 3/1/87  |
- 
- |       |                     |   |
|-------|---------------------|---|
| NO. 6 | <b>AUTHOR:</b>      | Bates, J.K.; Jardine, L.J.; Steindler, M.J.                     |
|       | <b>TITLE:</b>       | The Hydration Process of Nuclear Waste Glass: An Interim Report |
|       | <b>PUBLICATION:</b> | Argonne National Lab Report ANL 82-11                           |
|       | <b>DATE:</b>        | 1/1/82  |
-

- NO. 7    AUTHOR:        Bates, J.K.; Jardine, L.J.; Flynn, K.F.; Steindler, M.J.  
          TITLE:        The Application of Neutron Activation Analysis to the  
                          Determination of Leach Rates of Simulated Nuclear Waste  
                          Forms  
          PUBLICATION: Argonne National Lab Report  
          DATE:        1/1/81
- 
- NO. 8    AUTHOR:        Bergsman, T.M.; Shade, J.W.; Farnsworth, R.K.  
          TITLE:        Fifth In Situ Vitrification Engineering-Scale Test of Simulated  
                          INEL Buried Waste Sites  
          PUBLICATION: Battelle, PNL-8152, Prepared for USDOE,  
          DATE:        June 1992
- 
- NO. 9    AUTHOR:        Bibler, N.E.; Jantzen, C.M.  
          TITLE:        The Product Consistency Test and its Role in the Waste Glass  
                          Acceptance Process for DWPF Glass  
          PUBLICATION: Waste Management '89, Vol. 2, R.G. Post Ed. p 743-749  
          DATE:        1/1/89
- 
- NO. 10    AUTHOR:        Bibler, N.E.; Jantzen, C.M.  
          TITLE:        Materials Interactions Relating to Long-Term Geologic  
                          Disposal of Nuclear Waste Glass  
          PUBLICATION: Scientific Basis for Nuclear Waste Management, X J.K. Bates  
                          (Ed), P. 47-66, Materials Research Society, Pittsburg  
          DATE:        1/1/89
- 
- NO. 11    AUTHOR:        Bickford, D.F.; Smith, M.E.; Allen, P.M.; Faraci, J.P.;  
                          Langton, C.A.; Wolf, K.Z.  
          TITLE:        Application of High Level Waste-Glass Technology to the  
                          Volume Reduction and Immobilization of TRU, Low Level,  
                          and Mixed Wastes  
          PUBLICATION: Proceedings "Waste Management '91" RG Post Ed. ANS  
                          Tucson, AZ p. 537-545  
          DATE:        1/1/91
- 
- NO. 12    AUTHOR:        Bickford, D.F.; Ramsey, A.A.; Jantzen, C.M.; Brown, K.G.  
          TITLE:        Control of Radioactive Waste Glass Melters: I, Preliminary  
                          General Limits at Savannah River  
          PUBLICATION: Journal of American Ceramic Society, Vol. 73, 10, p. 2896-  
                          2902  
          DATE:        10/1/90
-



- NO. 13    AUTHOR:        Bickford, D.F.; Hrma, P.; Bowan, B.W.  
          TITLE:        Control of Radioactive Waste Glass Melters: II, Residence  
                          Time and Melt Rate Limitations  
          PUBLICATION: Journal American Ceramic Society, Vol. 73, 10 p. 2903-2915  
          DATE:        10/1/90
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- NO. 14    AUTHOR:        Bickford, D. F.  
          TITLE:        Advanced Radioactive Waste-Glass Melters  
          PUBLICATION: Presentation to American Ceramic Society, Tutorial Lecture  
                          on Novel Ceramic Processing/Nuclear Division, WSRC-RP-  
                          89-1174  
          DATE:        4/22/90
- 
- NO. 15    AUTHOR:        Bickford, D.F.; Probst, R.C.; Plodinec, M.J.  
          TITLE:        Control of Radioactive Waste-Glass Melters: Part 3 - Glass  
                          Electrical Stability  
          PUBLICATION: Advances in the Fusion of Glass, D.F. Bickford, ed., American  
                          Ceramic Society Westerville, OH, June 1988  
          DATE:        6/1/88
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- NO. 16    AUTHOR:        Bickford, D.F., Pellarin, D.J.  
          TITLE:        Large Scale Leach Testing of DWPF Canister Sections  
          PUBLICATION: Savannah River Laboratory, DP-MS-86-72  
          DATE:        12/1/86
- 
- NO. 17    AUTHOR:        Cantale, C.; Castelli, S.; Donato A.; Traverso, D.M.;  
                          Colombo, P.; Scarinci, G.  
          TITLE:        A Borosilicate Glass for the Italian High-Level Waste -  
                          Characterization and Behavior  
          PUBLICATION: Radioactive Waste Management and the Nuclear Fuel Cycle,  
                          16 (1), pp. 25-47  
          DATE:        1/1/91
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- NO. 18    AUTHOR:        Chandler, G.T.; Wicks, G.G.; Wallace, R.M.  
          TITLE:        Chemical Durability of SRP Waste Glass - Saturation Effects  
                          and Influence of SA/V  
          PUBLICATION: Savannah River Laboratory, DP-MS-84-37  
          DATE:        4/29/84
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- NO. 19    AUTHOR:        Chapman, C.C.  
          TITLE:        State-of-the-Art of Waste Glass Melters  
          PUBLICATION: 3rd International Symposium, Advances in Fusion and  
                          Processing of Glass 1992, New Orleans  
          DATE:        1992

- NO. 20    AUTHOR:        Chapman, C.C.  
 TITLE:            Evaluation of Vitrifying Municipal Incinerator Ash  
 PUBLICATION:    Ceramic Transactions, Nuclear Waste Management IV, G.G. Wicks, Ed., p. 223-231  
 DATE:            4/22/91
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- NO. 21    AUTHOR:        Chapman, C.C.; Pope, J.M.; Barnes, S.M.  
 TITLE:            Electric Melting of Nuclear Waste Glasses, State of the Art  
 PUBLICATION:    Journal Non-Crystalline Solids, No. 84, p. 226-240  
 DATE:            1/1/86
- 
- NO. 22    AUTHOR:        Coordinated Research Programme of the Evaluation of  
 TITLE:            Solidified High-Level Waste Form  
                       Chemical Durability and Related Properties of Solidified  
                       High-Level Waste Forms  
 PUBLICATION:    IAEA Technical Report Series No. 257  
 DATE:            1/1/85
- 
- NO. 23    AUTHOR:        Covington, J.F.; Wicks, G.G.; Molecke, M.A.  
 TITLE:            WIPP/SRL In-Situ Tests: MIT Program - The Effects of  
                       Metal Package Components  
 PUBLICATION:    Ceramic Transactions, Nuclear Waste Management, Vol. 23,  
                       pp. 723-732  
 DATE:            4/29/91
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- NO. 24    AUTHOR:        Curti, E.  
 TITLE:            Modeling the Dissolution of Borosilicate Glasses for  
                       Radioactive Waste Disposal with the PHREEQE/GLASSOL  
                       Code: Theory and Practice  
 PUBLICATION:    PSI-Bericht Nr. 86  
 DATE:            1/1/91
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- NO. 25    AUTHOR:        Czuczwa, J.M.; Farzan, H.; Vecci, S.J.; Warchol, J.J.  
 TITLE:            Cyclone Furnace for Vitrification of Contaminated Soil and  
                       Wastes  
 PUBLICATION:    1991 Incineration Conference Proceedings, pp. 613-620,  
                       Knoxville, TN  
 DATE:            May 1991
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- NO. 26    AUTHOR:        Delege, F.; Dussossoy, J.L.  
          TITLE:        R7T7 Glass Initial Dissolution Rate Measurements Using a  
                          High-Temperature Soxhlet Device  
          PUBLICATION: Material Resource Society Symposium Proceedings 212, pp.  
                          41-47  
          DATE:        1/1/91
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- NO. 27    AUTHOR:        Ewing, R.C.; Jercinovic, J.J.  
          TITLE:        Natural Analogues: Their Application to the Prediction of the  
                          Long-Term Behavior of Nuclear Waste Forms  
          PUBLICATION: Material Research Society Symposium Proceedings, 84, pp.  
                          67-93  
          DATE:        1/1/87
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- NO. 28    AUTHOR:        Gafney, J.  
          TITLE:        Chemistry  
          PUBLICATION: Science News, p. 173  
          DATE:        3/14/92
- 
- NO. 29    AUTHOR:        Grambow, B.; Lutze, W; Ewing, R.C.; Werme, L.O.  
          TITLE:        Performance Assessment of Glass as a Long-Term Barrier to  
                          the Release of Radionuclides into the Environment  
          PUBLICATION: Material Research Society Symposium Proceedings, 112  
          DATE:        1/1/88
- 
- NO. 30    AUTHOR:        Harbour, J.R.  
          TITLE:        Demonstrating Compliance with the Waste Acceptance  
                          Preliminary Specifications on Foreign Materials within DWPF  
                          Canistered Waste Forms  
          PUBLICATION: Proceedings of the 2nd International Seminar, Radioactive  
                          Waste Products, E. Warnecke, et al., ed., KFA Research  
                          Julich, FRG, WSRC-MS-90-98, 1990.  
          DATE:        5/28/90
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- NO. 31    AUTHOR:        Harbour, J.R.; Miller, T.J.; Whitaker, M.J.  
          TITLE:        The Determination of Pressure, Dewpoint, and Composition of  
                          the Gas Within the Free Volume of Canistered Waste Forms  
          PUBLICATION: WSRC-RP-90-1167  
          DATE:        1/1/90
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- NO. 32    **AUTHOR:**        Hazelton, R.F.; Thornhill, C.K.; Ross, W.A.  
          **TITLE:**            Evaluation of the Potential for Gas Pressurization and Free  
                                 Liquid Accumulation in a Canister from the West Valley  
                                 Demonstration Project  
**PUBLICATION:**    Ceramic Transactions, Nuclear Waste Management, IV, Vol.  
                                 23, pp. 491-500  
**DATE:**                4/29/91
- 
- NO. 33    **AUTHOR:**        Janke, D.S.; Chapman, C.C.; Vogel, R.A.  
          **TITLE:**            Results of Vitrifying Fernald K-65 Residue  
**PUBLICATION:**    Ceramic Transactions, Nuclear Waste Management, Vol. 23,  
                                 pp. 53-61  
**DATE:**                4/29/91
- 
- NO. 34    **AUTHOR:**        Jantzen, C.M.  
          **TITLE:**            Solidification of Consolidated Incinerator Facility (CIF)  
                                 Wastes in Soda-Lime-Silica (SLS) Glass: Use of Reactive  
                                 Additives to Retain Hazardous and Heavy Metal  
**PUBLICATION:**    Savannah River Lab, WSRC-TR-92-214, Rev. 0  
**DATE:**                4/30/92
- 
- NO. 35    **AUTHOR:**        Jantzen, C.M.  
          **TITLE:**            Thermodynamic Approach to Glass Corrosion  
**PUBLICATION:**    Corrosion of Glass, Ceramics, and Ceramic Superconductors,  
                                 Principles, Testing, Characterization and Applications, D.E.  
                                 Clark and B.K. Zaitos, eds., pp. 153-215, Noves Publications,  
                                 Park Ridge, NJ  
**DATE:**                1/1/92
- 
- NO. 36    **AUTHOR:**        Jantzen, C.M.  
          **TITLE:**            First Principles Process-Product Models for Vitrification of  
                                 Nuclear Waste: Relationship of Glass Composition to Glass  
                                 Viscosity, Resistivity, Liquidus Temperature  
**PUBLICATION:**    Ceramic Transactions, Vol. 23, pp. 37-51, Nuclear Waste  
                                 Management IV, G.G. Wicks, et al., eds.  
**DATE:**                4/29/91
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- NO. 37    **AUTHOR:**        Jantzen, C.M.  
          **TITLE:**            Systems Approach to Nuclear Waste Glass Development  
**PUBLICATION:**    Journal of Non-Crystalline Solids, 84, No. 1-3, pp. 215-225  
**DATE:**                1/1/86
-

- NO. 38    **AUTHOR:**        Jantzen, C.M.; Bibler, N.E.  
**TITLE:**                Nuclear Waste Glass Product Consistency Test (PCT)  
**PUBLICATION:**      Savannah River Site  
**DATE:**                 1/1/90
- 
- NO. 39    **AUTHOR:**        Jantzen, C.M.; Bibler, N.E.  
**TITLE:**                The Product Consistency Test for the DWPF Wasteform  
**PUBLICATION:**      Proceedings of the 2nd International Seminar on Radioactive  
                                 Waste Products, E. Odoj, et al., eds., pp. 609-622  
**DATE:**                 6/1/90
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- NO. 40    **AUTHOR:**        Jantzen, C.M.; Clarke, D.R.; Morgan, P.E.D.; Harker, A.B.  
**TITLE:**                Leaching of Polyphase Nuclear Waste Ceramics:  
                                 Microstructural and Phase Characterization  
**PUBLICATION:**      Journal American Ceramic Society, Vol. 76, No.6, pp. 292-  
                                 300  
**DATE:**                 6/1/82
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- NO. 41    **AUTHOR:**        Jantzen, C.M.; Plodinec, M.J.  
**TITLE:**                Thermodynamic Model of Natural, Medieval and Nuclear  
                                 Waste Glass Durability  
**PUBLICATION:**      Journal of Non-Crystalline Solids, 67, pp. 107-223  
**DATE:**                 1/1/84
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- NO. 42    **AUTHOR:**        Jercinovic, M.J.; Kaser, S.A.; Ewing, R.C.; Lutze, W.  
**TITLE:**                Comparison of Surface Layers Formed on Synthetic Basaltic  
                                 Glass, French R7T7 and HMI Borosilicate Nuclear Waste  
                                 Form Glasses-Materials Interface Interactions Tests  
**PUBLICATION:**      Material Research Society Symposium Proceedings, 176,  
                                 pp.355-362, 1990  
**DATE:**                 1/1/90
- 
- NO. 43    **AUTHOR:**        Jercinovic, M. J.; Ewing, R.C.  
**TITLE:**                Basaltic Glasses From Iceland and the Deep Sea: Natural  
                                 Analogues to Borosilicate Nuclear Waste-Form Glass  
**PUBLICATION:**      JSS 88-01  
**DATE:**                 1/1/88
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- NO. 44    **AUTHOR:**        Lutze, W.  
**TITLE:**                Silicate Glasses  
**PUBLICATION:**      Radioactive Waste Forms for the Future, W. Lutze and R.C.  
                                 Ewing, Elsevier Science Publishers, B.V.  
**DATE:**                 1/1/88
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- NO. 45    AUTHOR:        Lutze, W.; Ewing, R.C.  
          TITLE:        Radioactive Waste Forms for the Future  
          PUBLICATION: Book, Elsevier Science Publishers, B.V., pp.699-739  
          DATE:         1/1/88
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- NO. 46    AUTHOR:        Lutze, W.; Grambow, B.  
          TITLE:        Chemical Corrosion of Lead-Iron Phosphate Glass  
          PUBLICATION: In-process  
          DATE:
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- NO. 47    AUTHOR:        Marples, J.A.C.; Lutze, W.; Kawanishi, M; Van Iseghem, P.  
          TITLE:        A Comparison of the Behavior of Vitrified HLW in  
                          Repositories in Salt, Clay, and Granite, I: Experimental  
          PUBLICATION: Material Research Society Symposium Proceedings, 176, pp.  
                          267-274  
          DATE:         1/1/90
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- NO. 48    AUTHOR:        McDonell, W. R.  
          TITLE:        Comparison of SRP High-Level Waste Disposal Costs for  
                          Borosilicate Glass and Crystalline Ceramic Waste Forms  
          PUBLICATION: DuPont, Savannah River Laboratory, DPST-82-346  
          DATE:         4/1/82
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- NO. 49    AUTHOR:        Mertens, L.A.; Lutze, W.; Marples, J.A.C.; Van Iseghem, P.;  
                          Vernaz, E.  
          TITLE:        A Comparison of the Behavior of Vitrified HLW In  
                          Repositories In Salt, Clay, and Granite, I: Experimental  
          PUBLICATION: Material Research Society Symposium Proceedings, 176, pp.  
                          267-274  
          DATE:         1/1/90
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- NO. 50    AUTHOR:        Namboodri, C.G.; Wicks, G.G.; Lodding, A.R.; Hench, L.L.;  
                          Newton, R.G.  
          TITLE:        Surface Analyses of SRS Waste Glass Buried For Up to Two  
                          Years in Limestone in the United Kingdom  
          PUBLICATION: Ceramic Transactions, Vol. 23, Nuclear Waste Management  
                          IV, G.G. Wicks, ed., American Ceramic Society, Westerville,  
                          OH, p. 653-662  
          DATE:
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- NO. 51    AUTHOR:        Pegg, I.L.; Greenman, W.; Guo, Y.; Muller, I.S.; Mohr, R.K.;  
Macedo, P.B.; Grant, D.C.; Mullik, P.R.  
TITLE:                Development of a Combined Soil Wash/In-Furnace  
Vitrification System for Soil Remediation at DOE Sites  
PUBLICATION:        First International Mixed Waste Symposium, ASME  
DATE:                 8/1/91
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- NO. 52    AUTHOR:        Petit, J.C.; Mea, Della G.; Dran, J.C.; Magnathier, M.C.;  
Mando, P.A.; Paccagnella, A.  
TITLE:                Hydrated-Layer Formation During Dissolution of Complex  
Silicate Glasses and Minerals  
PUBLICATION:        Geochim-Cosmochim. Acta, 54, pp. 1941-1955  
DATE:                 1/1/90
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- NO. 53    AUTHOR:        Phinney, D.L.; Ryerson, F.J.; Oversby, V.M.; Lanford, W.A.;  
Aines, R.D.; Bates, J.K.  
TITLE:                Integrated Testing of the SRL-165 Glass Waste Form  
PUBLICATION:        Material Research Society Symposium Proceedings, 84, pp.  
433-446  
DATE:                 1/1/86
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- NO. 54    AUTHOR:        Plodinec, M.J.  
TITLE:                Viscosity of Glasses Containing Simulated Savannah River  
Plant Waste  
PUBLICATION:        Savannah River Laboratory, DP-1507  
DATE:                 8/1/78
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- NO. 55    AUTHOR:        Pye, L.D.  
TITLE:                The Physical and Thermal Properties of Simulated Nuclear  
Waste Glasses and Their Melts  
PUBLICATION:        Savannah River Lab DPST-85-397  
DATE:                 2/1/85
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- NO. 56    AUTHOR:        Ramsey, A.A.  
TITLE:                EPA Tests of Simulated DWPF Waste Glass-U  
PUBLICATION:        WSRC-TR-90-22, InterOffice Memo to M.J. Plodinec  
DATE:
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- NO. 57    AUTHOR:        Ramsey, W.G.; Taylor, T.D.; Jantzen, C.M.  
TITLE:                Predictive Modeling of Leachate pH for Simulated High Level  
Waste Glass  
PUBLICATION:        Ceramic Transactions, Vol. 23, Nuclear Waste Management,  
pp. 105-114  
DATE:                 4/29/91

- NO. 58    AUTHOR:        Roberts, F.P.; Turcotte, R.P.; Weber, W.J.  
          TITLE:        Materials Characterization Center, Workshop on the  
                          Irradiation Effects in Nuclear Waste Forms - Summary Report  
          PUBLICATION: US DOE Report, PNL-3588, Pacific Northwest Laboratory,  
                          Richland, WA  
          DATE:        1/1/81
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- NO. 59    AUTHOR:        Ross, W.A.; Lokken, R.O.; May, R.P.; Roberts, F.P.;  
                          Timmerman, C.L.; Treat, R.L.; Westik, J.H.  
          TITLE:        Comparative Assessment of TRU Waste Forms and Processes.  
                          Vol. I Waste Form and Process Evaluations  
          PUBLICATION: US DOE Report PNL-4428 Vol. 1  
          DATE:        9/1/81
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- NO. 60    AUTHOR:        Savannah River Laboratory  
          TITLE:        Environmental Assessment, Waste Form Selection for SRP  
                          High-Level Waste  
          PUBLICATION: US DOE, DOE/EA-0179  
          DATE:        7/1/82
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- NO. 61    AUTHOR:        Schreiber, H.D.; Sisk, E.D.; Schreiber, C.W.; Burns, J.K.  
          TITLE:        Solubilities of Nickel and Cobalt Chalcogenides in a Nuclear  
                          Waste Glass  
          PUBLICATION: Ceramic Transactions, Vol. 23, Nuclear Waste Management,  
                          pp. 213-222  
          DATE:        4/29/91
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- NO. 62    AUTHOR:        Schumacher, R. F.  
          TITLE:        DWPF Batch 2, Waste Glass Investigations  
          PUBLICATION: Ceramic Transactions, Vol. 23, Nuclear Waste Management,  
                          G.G. Wicks, ed., pp. 453-463  
          DATE:        5/1/91
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- NO. 63    AUTHOR:        Shade, J.W.; Thompson, L.E.; Kindle, C.H.  
          TITLE:        In Situ Vitrification of Buried Waste Sites  
          PUBLICATION: Ceramic Transactions, Vol. 23, Nuclear Waste Management,  
                          G.G. Wicks, ed., American Ceramic Society, Westerville, OH,  
                          pp. 633-640  
          DATE:
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- NO. 64 AUTHOR: Shenkler, E.S.; Graham, S.; Greenhut, V.A.  
TITLE: Secondary Lead Smelter Slags: Minimizing Lead Release Levels  
PUBLICATION: Ceramic Transactions, Vol. 23, Nuclear Waste Management IV, G.G. Wicks, ed., American Ceramic Society, Westerville, OH, pp. 75-84  
DATE: 4/29/91
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- NO. 65 AUTHOR: Slate, S.C.; Ross, W.A.; Partain, W.L.  
TITLE: Impact Testing of Vitreous Simulated High-Level Waste in Canisters  
PUBLICATION: USDOE BNWL-1903  
DATE: 5/1/75
- 
- NO. 66 AUTHOR: Smith, D.K.  
TITLE: Mineralogical, Textural and Compositional Data on the Alteration of Basaltic Glass From Kilauea, Hawaii to 300°C: Insights to the Corrosion of Borosilicate Glass  
PUBLICATION: Material Research Society Symposium Proceedings, 212, pp. 115-121  
DATE: 1991
- 
- NO. 67 AUTHOR: Smith, M.E.  
TITLE: Travel Report - Summary of Talk Given at the SAIC OTD Thermal Working Meeting in Salt Lake City  
PUBLICATION: Savannah River Lab, WSRC-RP-91-1092  
DATE: 10/30/91
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- NO. 68 AUTHOR: Smith, T.H.; Ross, W.A.  
TITLE: Impact Testing of Vitreous Simulated High-Level Waste in Canisters  
PUBLICATION: USDOE BNWL-1903  
DATE: 5/1/75
- 
- NO. 69 AUTHOR: Soper, P.D.; Walker D.D.; Plodinec, M.J.; Roberts, G.J.; Lightner, L.F.  
TITLE: Optimization of Glass Composition for the Vitrification of Nuclear Waste  
PUBLICATION: American Ceramic Society, Bulletin, Vol. 62, No. 9, pp. 1013-1028, DP-MS-81-108  
DATE: 9/1/83
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- NO. 70    AUTHOR:        Stone, J.A.  
          TITLE:        An Experimental Comparison of Alternative Solid Forms for Savannah River High-Level Wastes  
          PUBLICATION: Scientific Basis for Nuclear Waste Management., Annual Meeting of the Materials Research Society, DDP-MS-81-102  
          DATE:        11/16/81
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- NO. 71    AUTHOR:        Stone, J.A.; Allender, J.S.; Gould, T.H.  
          TITLE:        Comparison of Properties of Borosilicate Glass and Crystalline Ceramic Forms for Immobilization of Savannah River Plant Waste  
          PUBLICATION: USDOE Savannah River Lab., DP-1627  
          DATE:        4/1/82
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- NO. 72    AUTHOR:        Taylor, R.F.  
          TITLE:        Chemical Engineering Problems of Radioactive Waste Fixation by Vitrification  
          PUBLICATION: Chemical Engineering Science, Vol. 40, No. 4, pp. 541-569  
          DATE:        1985
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- NO. 73    AUTHOR:        Van Iseghem, P.; Grambow, B.  
          TITLE:        The Long-Term Corrosion and Modeling of Two Simulated Belgian Reference High-Level Waste Glasses  
          PUBLICATION: Material Research Society Symposium Proceedings, 112 , pp. 631-639  
          DATE:        1/1/88
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- NO. 74    AUTHOR:        Vernaz, E.Y.; Gordon, N.  
          TITLE:        Key Parameters of Glass Dissolution in Integrated Systems  
          PUBLICATION: Material Research Society Symposium Proceedings, 212, pp. 19-30  
          DATE:        1/1/91
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- NO. 75    AUTHOR:        Volf, M.B.  
          TITLE:        Chemical Approach to Glass  
          PUBLICATION: Book, Glass Science and Technology, 7 Elsevier, New York  
          DATE:        1/1/84
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- NO. 76    AUTHOR:        Weber, W.J.; Turcotte, R.P.  
          TITLE:        Materials Characterization Center, Second Workshop on Irradiation Effects in Nuclear Waste Forms, Summary Report  
          PUBLICATION: USDOE, PNL-4121  
          DATE:        1/1/82



**GLASS SUMMARIES****NO. 1****TITLE:** Summary of Properties for Borosilicate Waste Glasses**AUTHOR:****PUBLICATION:****DATE:****ORGANIZATION:****SUMMARY:**

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Glass, borosilicate  |
| <b>Waste Type:</b>                | High level radioactive defense waste   |
| <b>Waste Loading:</b>             | Nominal 28 wt% from insoluble waste, 8 wt% from soluble  |
| <b>Development Status:</b>        | Mature after extensive testing   |
| <b>Compressive Strength:</b>      | 550 MPa (80,000 psi)   |
| <b>Thermal Stability:</b>         | Control cooling to limit fracture, little change at T<500°C.   |
| <b>Radiation Stability:</b>       | No change due to self irradiation for 1 million years  |
| <b>Biological Stability:</b>      | Inert  |
| <b>Leach Resistance:</b>          | See conclusions  |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | No free liquids  |
| <b>Chemical Durability:</b>       | Excellent  |
| <b>Compositional Flexibility:</b> | Limits on certain oxides, metals, redox, and salts   |
| <b>Gas Generation:</b>            | No gas release-simulated waste   |
| <b>RCRA Compliance:</b>           | Passes TCLP - all species below detection limit  |
| <b>Conclusions:</b>               | MCC-1, < 1 g/m <sup>2</sup> day PCT norm. Release boron <1 g/l;<br>leach resistance influenced by glass composition, pH of<br>leachate, temperature, SA/V and environment. Not affected<br>by canister. Identical results for radioactive glasses. |

## GLASS SUMMARIES

NO. 2

**TITLE:** A State-of-the-Art Review of Materials Properties of Nuclear Waste Forms

**AUTHOR:**

**PUBLICATION:** USDOE, PNL-3802

**DATE:** April 1, 1981

**ORGANIZATION:** Pacific Northwest Laboratory, Battelle

**SUMMARY:** Report is a state-of-the-art review of materials properties of nuclear waste forms for HLW. Physical properties include density, thermal conductivity and expansion, mechanical, chemical durability, radiation effects, thermal stability, etc. Waste forms include cement, ceramics, and glass.

|                            |   |
|----------------------------|---|
| Waste Form:                | Glasses, high and low fire ceramics, composites,  |
| Waste Type:                | High level radioactive waste  |
| Waste Loading:             | --  |
| Development Status:        | --  |
| Compressive Strength:      | Tensile: cement ~8, ceramic 8-30, glass 50 MPa  |
| Thermal Stability:         | --  |
| Radiation Stability:       | Density change: glass $\pm 1$ %; no change in leach rate  |
| Biological Stability:      | --  |
| Leach Resistance:          | Cement $10^{-3}$ , ceramics $10^{-7}$ to $10^{-8}$ , glass $10^{-4}$ to $10^{-7}$ g/cm <sup>2</sup> day |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | A review of literature on waste forms prior to 1981.  |

**GLASS SUMMARIES****NO. 3**

**TITLE:** Vitrification Technologies for Treatment of Hazardous and Radioactive Waste

**AUTHOR:**

**PUBLICATION:** EPA/625/R-92/002 U.S. EPA, Center for Environmental Research Information, Cincinnati, OH 45268

**DATE:** May 1, 1992

**ORGANIZATION:** U.S. Environmental Protection Agency, Office of Research and Development

**SUMMARY:** This is a handbook on the theory of vitrification processing and provides an overview of the applications and limitations of vitrification waste treatment. Chapters cover glass structure, vitrification processes, waste types, product characteristics, offgas treatment, capabilities and limitations, physical and chemical testing, and process evaluation.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Glass   |
| <b>Waste Type:</b>                | Hazardous and radioactive waste   |
| <b>Waste Loading:</b>             | Varied  |
| <b>Development Status:</b>        | Varied  |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | --  |
| <b>Biological Stability:</b>      | --  |
| <b>Leach Resistance:</b>          | --  |
| <b>Immersion Stability:</b>       | --  |
| <b>Free Liquids:</b>              | --  |
| <b>Chemical Durability:</b>       | --  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | A good review of vitrification technology and products for hazardous and radioactive wastes. Strong on ISV. |

## GLASS SUMMARIES

NO. 4

**TITLE:** Initial Demonstration of the DWPF Vitrification Process and Product Control Strategy Using Actual Radioactive Waste

**AUTHOR:** M.K. Andrews; N.E. Bibler; C.M. Jantzen; D.C. Beam

**PUBLICATION:** Ceramic Transactions, Nuclear Waste Management IV, Vol. 23, pp. 569-576

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** To test the DWPF control algorithms, actual radioactive waste was melted in a shielded joule-heated melter facility at SRS and the resulting glass evaluated. Algorithms were used to control glass properties. Predicted values agreed reasonably well with measured values.

|                            |   |                  |                 |
|----------------------------|---|------------------|-----------------|
| Waste Form:                | Glass, borosilicate                           |                  |                 |
| Waste Type:                | High level radioactive defense waste (actual) |                  |                 |
| Waste Loading:             | Nominal 28 wt%                                |                  |                 |
| Development Status:        | Mature  |                  |                 |
| Compressive Strength:      | --  |                  |                 |
| Thermal Stability:         | --  |                  |                 |
| Radiation Stability:       | --  |                  |                 |
| Biological Stability:      | --  |                  |                 |
| Leach Resistance:          | See summary                                   |                  |                 |
| Immersion Stability:       | --  |                  |                 |
| Free Liquids:              | See summary                                   |                  |                 |
| Chemical Durability:       | --  |                  |                 |
| Compositional Flexibility: | --  |                  |                 |
| Gas Generation:            | --  |                  |                 |
| RCRA Compliance:           | --  |                  |                 |
| Conclusions:               | <u>Properties</u>                             | <u>Predicted</u> | <u>Measured</u> |
|                            | Viscosity                                     | 64.8 P           | <100 POISE      |
|                            | Liquidus                                      | 1005°C           | <1050 °C.       |
|                            | Boron Rel.                                    | 26 ppm           | 69-48 ppm       |
|                            | Sodium Rel                                    | 50 ppm           | 122-88 ppm      |

The glass algorithms can be used to control actual radioactive glass vitrification. The glass was pourable indicating its viscosity was < 100 Poise as predicted. There were few spinel crystals indicating that the glass liquidus was less than 10,500°C as predicted. The durability of the glass was less than that predicted by the free energy of hydration model

## GLASS SUMMARIES

NO. 5

**TITLE:** Interim Milestone HWVP-86-VII22C - Report on Composition Variability Testing Conducted for the HWVP

**AUTHOR:** S.O. Bates; W. M. Bowen

**PUBLICATION:** HWVP-86-VI 122C Final Report

**DATE:** March 1, 1987

**ORGANIZATION:** Pacific Northwest Laboratory

**SUMMARY:** Scoping studies were conducted on the variation of certain oxides within a reference HWVP borosilicate glass.

**Waste Form:** Glass, borosilicate

**Waste Type:** High level radioactive defense waste

**Waste Loading:** 20 to 30 wt%

**Development Status:** Intermediate

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** MCC-1 @ 90°C 28 day < 1 g/m<sup>2</sup> day

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Scoping tests confirmed that wide variations in composition can be accommodated in the reference glass. The following components will be included in future glass studies: waste level, zirconium, sodium, iron, aluminum and silicon.



## GLASS SUMMARIES

NO. 6

**TITLE:** The Hydration Process of Nuclear Waste Glass: An Interim Report

**AUTHOR:** J. K. Bates; L. J. Jardine; M.J. Steindler

**PUBLICATION:** Argonne National Lab Report ANL 82-11

**DATE:** January 1, 1982

**ORGANIZATION:** Argonne National Laboratory

**SUMMARY:** Reaction of glass in water vapor at 120-240°C results in formation of secondary phases.

**Waste Form:** Borosilicate glass (SRL 211, SRL 131)

**Waste Type:** --

**Waste Loading:** --

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Decreases as secondary phases form

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Hydration - aging of glass in repository important prior to possible contact with liquid water. Aging will decrease glass durability.

## GLASS SUMMARIES

NO. 7

**TITLE:** The Application of Neutron Activation Analysis to the Title:  
Determination of Leach Rates of Simulated Nuclear Waste Forms

**AUTHOR:** J.K. Bates; L. J. Jardine; K.F. Flynn; M.J. Steindler

**PUBLICATION:** Argonne National Lab Report

**DATE:** January 1, 1981

**ORGANIZATION:** Argonne National Laboratory

**SUMMARY:** Leach tests on borosilicate glasses that were neutron-activated and analysis of activation products by gamma spectroscopy (plus alpha-beta spectroscopy)

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Borosilicate glass (PNL 76-68 SRL 211)       |
| <b>Waste Type:</b>                | --   |
| <b>Waste Loading:</b>             | --   |
| <b>Development Status:</b>        | --   |
| <b>Compressive Strength:</b>      | --   |
| <b>Thermal Stability:</b>         | --   |
| <b>Radiation Stability:</b>       | Stable after neutron activation              |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | --   |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | --   |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | Technique not suitable for B or Si analysis. |

## GLASS SUMMARIES

NO. 8

**TITLE:** Fifth In Situ Vitrification Engineering-Scale Test of Simulated INEL Buried Waste Sites

**AUTHOR:** T.M. Bergsman; J.W. Shade; R.K. Farnsworth

**PUBLICATION:** Battelle, PNL-8152, Prepared for USDOE

**DATE:** June 1, 1992

**ORGANIZATION:** Battelle, PNL

**SUMMARY:** An in-situ vitrification of assorted sealed cans containing hazardous materials was carried out. The vitrified glass product passed the TCLP. Soil samples were also analyzed. No soil sample contained hazardous levels of organics and only one soil sample contained mercury above TCLP limits.

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Glass  |
| <b>Waste Type:</b>                | Soil and cans of hazardous waste   |
| <b>Waste Loading:</b>             | --   |
| <b>Development Status:</b>        | --   |
| <b>Compressive Strength:</b>      | --   |
| <b>Thermal Stability:</b>         | --   |
| <b>Radiation Stability:</b>       | --   |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | --   |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | --   |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | Samples of glass and the surrounding soil were tested by TCLP. Only one sample contained toxic material above TCLP limits. |

## GLASS SUMMARIES

NO. 9

**TITLE:** The Product Consistency Test and its Role in the Waste Glass Acceptance Process for DWPF Glass

**AUTHOR:** N.E. Bibler; C. M. Jantzen

**PUBLICATION:** Waste Management '89, Vol.1, R.G. Post, ed. pp. 743-749

**DATE:** January 1, 1989

**ORGANIZATION:** University of Arizona, Tucson, AZ

**SUMMARY:** Development of the Product Consistency Test (PCT) for radioactive waste glass. Reproducibility 2 - 3% for a single investigator and ~8% for different investigators.

|                            |  |      |          |          |
|----------------------------|--|------|----------|----------|
| Waste Form:                | Glass, borosilicate                            |      |          |          |
| Waste Type:                | Simulated high level radioactive defense waste |      |          |          |
| Waste Loading:             | --   |      |          |          |
| Development Status:        | Submitted to ASTM                              |      |          |          |
| Compressive Strength:      | --   |      |          |          |
| Thermal Stability:         | --   |      |          |          |
| Radiation Stability:       | --   |      |          |          |
| Biological Stability:      | --   |      |          |          |
| Leach Resistance:          | --   |      |          |          |
| Immersion Stability:       | --   |      |          |          |
| Free Liquids:              | None   |      |          |          |
| Chemical Durability:       | --   |      |          |          |
| Compositional Flexibility: | --   |      |          |          |
| Gas Generation:            | --   |      |          |          |
| RCRA Compliance:           | --   |      |          |          |
| Conclusions:               | Leach Test:                                    | PCT  | Glass    | DWPF     |
|                            | Leach Results:                                 | 131  | Rad 200R | Start-up |
|                            | NL (gm/l)*                                     | 1.10 | 1.05     | 0.95     |

\* Normalized Log Boron Concentration

## GLASS SUMMARIES

NO. 10

**TITLE:** Materials Interactions Relating to Long-Term Geologic Disposal of Nuclear Waste Glass

**AUTHOR:** N.E. Bibler; C.M. Jantzen

**PUBLICATION:** Scientific Basis for Nuclear Waste Management X, J.K. Bates, ed., pp. 47-66, Materials Research Society, Pittsburg

**DATE:** January 1, 1987

**ORGANIZATION:** Materials Research Society

**SUMMARY:** Interactions between various materials in canister, overpack, etc.; the ground water; and the geologic mineral in oxic and anoxic conditions. The geologies reviewed are tuff, salt, basalt, and granite.

|                                   |  |             |                             |
|-----------------------------------|--|-------------|-----------------------------|
| <b>Waste Form:</b>                | Glass, borosilicate  |             |                             |
| <b>Waste Type:</b>                | Simulated high level radioactive defense waste                 |             |                             |
| <b>Waste Loading:</b>             | --   |             |                             |
| <b>Development Status:</b>        | Mature   |             |                             |
| <b>Compressive Strength:</b>      | --   |             |                             |
| <b>Thermal Stability:</b>         | --   |             |                             |
| <b>Radiation Stability:</b>       | Leach resistance influenced by changes in pH due to radiation. |             |                             |
| <b>Biological Stability:</b>      | --   |             |                             |
| <b>Leach Resistance:</b>          | Influenced by certain metals, Eh, pH, and mineral system       |             |                             |
| <b>Immersion Stability:</b>       | --   |             |                             |
| <b>Free Liquids:</b>              | --   |             |                             |
| <b>Chemical Durability:</b>       | --   |             |                             |
| <b>Compositional Flexibility:</b> | --   |             |                             |
| <b>Gas Generation:</b>            | --   |             |                             |
| <b>RCRA Compliance:</b>           | --   |             |                             |
| <b>Conclusions:</b>               | <b>Tuff</b>  | <b>Salt</b> | <b>Basalt &amp; Granite</b> |
| Rock                              | None   | None        | Changes Eh                  |
| S. Steel                          | None   | None        | None                        |
| Iron overpack                     | -----  | Inc. Leach. | May Inc. Leach.             |
| Radiation                         | Changes pH   | Changes Eh  | None                        |

## GLASS SUMMARIES

NO. 11

**TITLE:** Application of High Level Waste-Glass Technology to the Volume Reduction and Immobilization of TRU, Low Level, and Mixed Wastes

**AUTHOR:** D.F. Bickford; M.E. Smith; P.M. Allen; J.P. Faraci; C.A. Langton; K.Z. Wolf

**PUBLICATION:** Proceedings of Waste Management '91, R.G. Post, ed., Tucson, AZ, pp. 537-545

**DATE:** January 1, 1991

**ORGANIZATION:**

**SUMMARY:** The EPA has designated vitrification as the "Best Developed Available Technology" for immobilization of high-level nuclear waste. Recent EPA announcements indicate the agency is considering long-term waste immobilization as a preferred mode of treatment for many types of hazardous waste. It was concluded that a program for adaptation of HLW vitrification to other wastes is clearly needed, and would have a high confidence of making major impact.

|                            |   |
|----------------------------|---|
| Waste Form:                | Glass, borosilicate   |
| Waste Type:                | Low level mixed wastes, mixed TRU waste   |
| Waste Loading:             | --  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Description of stirred glass melter. Low level mixed wastes which require permanent isolation are good candidates for vitrification. Examples of waste types are presented. |

## GLASS SUMMARIES

NO. 12

**TITLE:** Control of Radioactive Waste Glass Melters: I, Preliminary General Limits at Savannah River

**AUTHOR:** D.F. Bickford; A.A. Ramsey; C.M. Jantzen; K.G. Brown

**PUBLICATION:** Journal of the American Ceramic Society, Vol. 73, 10, pp. 2896-2902

**DATE:** October 1, 1990

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** Discusses DWPF melter control considerations, e.g., melter temperature, glass composition, product durability, waste loading limits, glass redox control, and glass cooling requirements.

**Waste Form:** Glass, borosilicate

**Waste Type:** Simulated high level radioactive defense waste

**Waste Loading:** Nominal loading 28 wt%, (< 30 wt%)

**Development Status:** Mature, extensive testing scale melters, simulated and radioactive feed

**Compressive Strength:** --

**Thermal Stability:** Cooling of canister controlled to prevent devitrification and cracking.

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Greater than environmental assessment glass (131/TDS)

**Immersion Stability:** --

**Free Liquids:** None

**Chemical Durability:** --

**Compositional Flexibility:** Certain oxides and salts will be limited by control system

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:**  $\text{TiO}_2$  <1.0 wt% of glass,  $\text{SO}_4$  <0.4 wt%,  $\text{Cr}_2\text{O}_3$  <0.3 wt%,  $\text{PO}_4$  < 3.0 wt%,  $\text{NaF}$  < 0.1 wt%,  $\text{NaCl}$  < 0.6 wt%,  $\text{Fe(II)/Fe(III)}$  < 0.5

## GLASS SUMMARIES

NO. 13

**TITLE:** Control of Radioactive Waste Glass Melters: II, Residence Time and Melt Rate Limitations

**AUTHOR:** D.F. Bickford; P. Hrma; B.W. Bowan

**PUBLICATION:** Journal of the American Ceramic Society, Vol. 73, 10, pp. 2903-2915

**DATE:** October 1, 1990

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** Investigated melting behavior of simulated feeds with respect to foaming, homogenization, and glass quality. Introduces potential for stirred melting.

**Waste Form:** Glass, borosilicate

**Waste Type:** Simulated high level radioactive defense waste

**Waste Loading:** --

**Development Status:** Mature

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** None

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Waste glass melter size and residence times required by glass quality are much smaller than those used in current practice. Shorter residence times and lower melt temperatures are sufficient to develop durability similar to that of existing melters operating at 1,150°C.



**GLASS SUMMARIES****NO. 14**

**TITLE:** Advanced Radioactive Waste-Glass Melters

**AUTHOR:** D.F. Bickford

**PUBLICATION:** Presentation to American Ceramic Society, Tutorial Lecture on Novel Ceramic Processing/Nuclear Division, WSRC-RP-89-1174

**DATE:** April 22, 1990

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** Discusses how technology developed for HLW might be used for other types of waste. A new type of stirred glass melter is introduced.

**Waste Form:** Glass

**Waste Type:** Simulated HLW, TRU, <sup>248</sup>Pu, U, RCRA, and mixed

**Waste Loading:** --

**Development Status:** Preliminary trial

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** A new class of waste glass melters has been designed and proof of concept tests completed on simulated HLRW slurry. Melt rates have exceeded 155 kg/m<sup>2</sup>/hr.

## GLASS SUMMARIES

NO. 15

**TITLE:** Control of Radioactive Waste-Glass Melters: Part 3 - Glass Electrical Stability

**AUTHOR:** D.F. Bickford; R.C. Probst; M.J.Plodinec

**PUBLICATION:** Advances in the Fusion of Glass, D.F. Bickford, ed., American Ceramic Society, Westerville, OH.

**DATE:** June 1, 1988

**ORGANIZATION:** American Ceramic Society

**SUMMARY:** Pilot waste glass melters have shown a tendency for noble-metal accumulation on melter floors. These build-ups can affect joule heating of the melters.

**Waste Form:** Glass, borosilicate

**Waste Type:** High level radioactive defense waste

**Waste Loading:** 28 wt%

**Development Status:** Mature

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Build-up of noble metals in the DWPF melter would not be sufficient to disrupt the joule heating in the melter. Discusses other elements, including mercury, sulfides, and silver.

**GLASS SUMMARIES**

NO. 16

**TITLE:** Large Scale Leach Testing of DWPF Canister Sections

**AUTHOR:** D.F. Bickford; D.J. Pellarin

**PUBLICATION:** Savannah River Laboratory, DP-MS-86-72

**DATE:** December 1, 1986

**ORGANIZATION:** International Symposium on the Scientific Basis for Nuclear Waste Management, Monitored Retrievable Storage

**SUMMARY:** Leach testing of 24-in. diameter canister sections. MCC-1 leach conditions. Less than a factor of 3 increase in leachability resulted from the scale-up conditions of the test.

**Waste Form:** Glass, borosilicate

**Waste Type:** Simulated high level radioactive defense waste

**Waste Loading:** Nominal 28 wt%

**Development Status:** Mature

**Compressive Strength:** --

**Thermal Stability:** Glass was fractured on cooling

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Less than a factor of 3 increase in leachability results from combined scale-up, glass cracking, leached surface area estimation and surface roughness effects.

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** 28-day large scale test compared to typical MCC-1 test. Large scale within factor of 3 to MCC-1. Difference due to saw cutting of canister glass. Comparison of lab scale test to full size conditions.

## GLASS SUMMARIES

NO. 17

**TITLE:** A Borosilicate Glass for the Italian High-Level Waste: Characterization and Behavior

**AUTHOR:** C. Cantale; S. Castelli; A. Donato; D.M. Traverso; P. Colombo; G. Scarinci

**PUBLICATION:** Radioactive Waste Management and the Nuclear Fuel Cycle, 16 (1), pp. 25-47.

**DATE:** January 1, 1991

**ORGANIZATION:** ENEA, CRE-Casaccia, and Universita'di Padova

**SUMMARY:** Description of processing, compositions, and initial testing of candidate waste forms for Italian radioactive waste

**Waste Form:** Borosilicate glass

**Waste Type:** --

**Waste Loading:** Waste/glass=57%

**Development Status:** Intermediate stage of development

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** MCC-1 NL(B) 28d=8g/m<sup>2</sup> 91d=9g/m<sup>2</sup> for "BAZ-R" glass

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Describes testing of candidate waste forms, including physical characteristics and chemical leaching of laboratory glass and plant-produced BAZ glass.

## GLASS SUMMARIES

NO. 18

**TITLE:** Chemical Durability of SRP Waste Glass - Saturation Effects and Influence of SA/V

**AUTHOR:** G.T. Chandler; G.G. Wicks; R.M. Wallace

**PUBLICATION:** Savannah River Laboratory, DP-MS-84-37

**DATE:** April 24, 1984

**ORGANIZATION:** American Ceramic Society

**SUMMARY:** Investigated the ratio of the glass surface area (SA) to the volume of leachant (V). MCC-1 used as test.

**Waste Form:** Glass, borosilicate

**Waste Type:** Simulated high level radioactive defense waste

**Waste Loading:** --

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** See summary

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Discusses the role of pH, saturation of chemical species, and SA/V on glass leach resistance.

## GLASS SUMMARIES

NO. 19

TITLE: State-of-the-Art of Waste Glass Melters

AUTHOR: C. Chapman

PUBLICATION: 3rd International Symposium, Advances in Fusion and Processing of Glass, New Orleans, Louisiana

DATE: May 1, 1992

ORGANIZATION: Battelle, Pacific Northwest Laboratories

SUMMARY: A review of types of melters under consideration for vitrification for both high level and low level mixed wastes.

|                            |   |
|----------------------------|---|
| Waste Form:                | Glass   |
| Waste Type:                | High and low level mixed waste  |
| Waste Loading:             | 25 to 50 wt%  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | The volumes of hazardous, nonhazardous, radioactive, and mixed wastes produced each year are large. Waste vitrification or transformation of waste into a nonleaching and useful secondary glass material is an environmentally attractive process. |

## GLASS SUMMARIES

NO. 20

**TITLE:** Evaluation of Vitrifying Municipal Incinerator Ash

**AUTHOR:** C.C. Chapman

**PUBLICATION:** Ceramic Transactions, Nuclear Waste Management IV, G.G. Wicks, ed., pp. 223-231.

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** Vitrifying incinerator ash is an attractive means of stabilizing the toxic materials. Vitrification provides a further volume reduction and produces a construction material. Vitrification provides an economic treatment method while providing an environmentally sound solution to a troublesome waste stream.

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Glass calcium aluminosilicate  |
| <b>Waste Type:</b>                | Incinerator ash, municipal   |
| <b>Waste Loading:</b>             | 100% to 80% volume reduction   |
| <b>Development Status:</b>        | Initial  |
| <b>Compressive Strength:</b>      | --   |
| <b>Thermal Stability:</b>         | --   |
| <b>Radiation Stability:</b>       | --   |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | --   |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | --   |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | Vitrification of municipal incinerator ash provides an economic treatment method, while providing an environmentally sound solution to disposal of the waste stream. |

## GLASS SUMMARIES

NO. 21

TITLE: Electric Melting of Nuclear Waste Glasses, State of the Art

AUTHOR: C.C. Chapman; J.M. Pope; S.M. Barnes

PUBLICATION: Journal of Non-Crystalline Solids, No. 84, pp. 226-240

DATE: January 1, 1986

ORGANIZATION: West Valley Nuclear Services Co.

SUMMARY: The paper reviews the design of melters used throughout the world. Includes the lessons learned and the problems associated with corrosion, crystalline sludge, reboil, and metal precipitation.

|                            |   |
|----------------------------|---|
| Waste Form:                | --  |
| Waste Type:                | --  |
| Waste Loading:             | --  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Discusses the liquid-fed ceramic, West Valley Demonstration, DWPF, PAMELA melter, Japanese vitrification, Hanford, and the U.K. Harwell melters. Problems with draining, corrosion, foaming, etc., are discussed. |



## GLASS SUMMARIES

NO. 22

**TITLE:** Chemical Durability and Related Properties of Solidified High-Level Waste Forms

**AUTHOR:** Coordinated Research Programme of the Evaluation of Solidified High-Level Waste Form.

**PUBLICATION:** IAEA Technical Report Series No. 257

**DATE:** January 1, 1985

**ORGANIZATION:** International Atomic Energy Agency

**SUMMARY:** Review of waste glass/ceramic compositions, leach tests, and multicomponent effects (radiation, backfill, etc.) for international application.

|                            |  |
|----------------------------|--|
| Waste Form:                | Borosilicate glass                                       |
| Waste Type:                | --   |
| Waste Loading:             | Effects of 10-30% on leaching are composition-dependent. |
| Development Status:        | --   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | --   |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | --   |

## GLASS SUMMARIES

NO. 23

**TITLE:** WIPP/SRL In Situ Tests: MIT Program - The Effects of Metal Package Components

**AUTHOR:** J.F. Covington; G.G. Wicks; M.A. Molecke

**PUBLICATION:** Ceramic Transactions, Nuclear Waste Management, Vol. 23, pp. 723-732

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** There were no significant effects observed on leaching of SRL Y waste glass due to the presence of 304L stainless steel.

|                            |   |
|----------------------------|---|
| Waste Form:                | Glass, borosilicate   |
| Waste Type:                | Simulated high level radioactive defense waste  |
| Waste Loading:             | Nominal 28 wt%  |
| Development Status:        | Mature  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | See summary   |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | See summary   |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | No large effects observed on the leaching of SRL Y waste glass when leached in the presence of metals, including titanium, copper, nickel alloys, and several carbon steels. Metals such as A216 and lead corroded significantly. |

## GLASS SUMMARIES

NO. 24

**TITLE:** Modeling the Dissolution of Borosilicate Glasses for Radioactive Waste Disposal with the PHREEQE/GLASSOL Code: Theory and Practice

**AUTHOR:** E. Curti

**PUBLICATION:** PSI-Bericht Nr. 86

**DATE:** January 1, 1991

**ORGANIZATION:** Paul Scherrer Institute, Wurenlingen, Germany

**SUMMARY:** Discusses Granbow model of silicic acid control of glass reaction rate and applies model to measured reactivity of British MW-glass at 90°C. Finds model requires too much curve fitting to define parameters to be acceptable.

**Waste Form:** Borosilicate glass (MW) and ABS-118

**Waste Type:** --

**Waste Loading:** --

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Long term MW =  $6 \times 10^{-3}$ , ABS-118 =  $1.31 \times 10^{-3}$  g/m<sup>2</sup>/day, deionized water, 90°C

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Current model (Granbow's version) requires too many ad hoc adjustments based on data to be modeled to be acceptable for long-term predictions.

## GLASS SUMMARIES

NO. 25

**TITLE:** Cyclone Furnace for Vitrification of Contaminated Soil and Wastes

**AUTHOR:** J.M. Czuczwa; H. Farzan; S.J. Vecci; J.J. Warchol

**PUBLICATION:** 1991 Incineration Conference Proceedings, pp. 613-620, Knoxville, TN

**DATE:** May 1, 1991

**ORGANIZATION:** Babcock & Wilcox Co.

**SUMMARY:** Describes cyclone furnace which might be well-suited for treating high inorganic content hazardous wastes, e.g., soils.

|                            |  |
|----------------------------|--|
| Waste Form:                | Slag   |
| Waste Type:                | Contaminated soils   |
| Waste Loading:             | 100%   |
| Development Status:        | Early  |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | Passes TCLP  |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | Passes TCLP  |
| Conclusions:               | Furnace was able to melt soil at 150 lb/hr. Slag passed TCLP. Approximately 95% of soil was captured as slag. Had a severe problem with toxic metal volatility. This was probably due to dispersion of waste into cyclonic burner. Not a good approach to trying to encapsulate inorganic toxic waste. |

## GLASS SUMMARIES

NO. 26

**TITLE:** R7T7 Glass Initial Dissolution Rate Measurements Using a High-Temperature Soxhlet Device

**AUTHOR:** F. Delege; J.L. Dussossoy

**PUBLICATION:** Materials Research Society Symposium Proceedings, 212, pp. 41-47

**DATE:** January 1, 1991

**ORGANIZATION:** CEA-CEN France

**SUMMARY:** Measure forward rate coefficient as function of temperature using Soxhlet from 90-250°C. Follows Arrhenius behavior with  $E_a=58-60$  kJ/mol

|                            |  |
|----------------------------|--|
| Waste Form:                | Borosilicate glass, R7T7   |
| Waste Type:                | --   |
| Waste Loading:             | --   |
| Development Status:        | --   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | $K_t = A \exp((-59k)/\text{mol}/RT)$   |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | Rate coefficient follows Arrhenius behavior between 90 to 250°C with $E_a=58$ to 60 kJ/mol. Release of Li, Na, B, & Si non-stoichiometric but all linear in reaction time. Interpret $E_a$ to indicate surface reaction control of rate. |

## GLASS SUMMARIES

NO. 27

**TITLE:** Natural Analogues: Their Application to the Prediction of the Long-Term Behavior of Nuclear Waste Forms

**AUTHOR:** R.C. Ewing; M.J. Jercinovic

**PUBLICATION:** Materials Research Society Symposium Proceedings, 84, pp. 67-83

**DATE:** January 1, 1987

**ORGANIZATION:** Department of Geology, University of New Mexico

**SUMMARY:** Review of natural analogue approach to HLW problem and summary of experimental results through 1986

|                            |   |
|----------------------------|---|
| Waste Form:                | Analogues for borosilicate glass  |
| Waste Type:                | --  |
| Waste Loading:             | --  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Discusses tektite, rhyolitic, and basaltic glass analogies based on silica contents, comparisons of experimentally altered glass and naturally altered glass, primarily through secondary solids generated. |

## GLASS SUMMARIES

NO. 28

TITLE: Chemistry

AUTHOR: J. Gafney

PUBLICATION: Science News, pg. 172

DATE: March 14, 1992

ORGANIZATION: Argonne National Laboratory

SUMMARY: Observed on a qualitative basis that humic acid may interact with borosilicate labware.

Waste Form: Glass

Waste Type: Humic acid

Waste Loading: --

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: --

Immersion Stability: --

Free Liquids: --

Chemical Durability: Humic acid may interact with borosilicate labware

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Humic acid may interact with borosilicate labware. Possible complexing agent.

## GLASS SUMMARIES

NO. 29

**TITLE:** Performance Assessment of Glass as a Long-Term Barrier to the Release of Radionuclides into the Environment

**AUTHOR:** B. Grambow; W. Lutze; R.C. Ewing; L.O. Werme

**PUBLICATION:** Material Research Society Symposium Proceedings, 112

**DATE:** January 1, 1988

**ORGANIZATION:** HMI, UNM, SKB

**SUMMARY:** Summarizes leaching test interpretation of JSS project based on Grambow's reaction model.

|                            |  |
|----------------------------|--|
| Waste Form:                | Borosilicate glass JSS-A ABS-118   |
| Waste Type:                | --   |
| Waste Loading:             | --   |
| Development Status:        | --   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | Forward rate 1.5g/m <sup>2</sup> /d final rate 0.0025 g/m <sup>2</sup> /d at 90°C  |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | COGEMA glass sufficiently durable at flow rates <100 l/yr; glassol code (and his model) validated; long-term 'final rate' not well understood; backfill not effective barrier. |



## GLASS SUMMARIES

NO. 30

**TITLE:** Demonstrating Compliance with the Waste Acceptance Preliminary Specifications on Foreign Materials within DWPF Canistered Waste Forms

**AUTHOR:** J.R. Harbour

**PUBLICATION:** Proceedings of the 2nd International Seminar on Radioactive Waste Products, E. Warnecke, et al., eds., KFA Research Julich, FRG, WSRC-MS-90-98

**DATE:** May 28, 1990

**ORGANIZATION:** Westinghouse Savannah River Co.

**SUMMARY:** Study of volatility of waste glass combined with new results of thermogravimetric analysis (TGA) experiments.

|                            |  |
|----------------------------|--|
| Waste Form:                | Glass, borosilicate  |
| Waste Type:                | Simulated high level radioactive defense waste   |
| Waste Loading:             | Nominal 28 wt%   |
| Development Status:        | Mature   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | --   |
| Immersion Stability:       | --   |
| Free Liquids:              | No organics or free liquid found   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | Highly reduced glass powder may absorb oxygen > 400°C  |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | TGA results demonstrate that powdered waste glass samples can gain or loose weight upon heating. The weight loss is associated with adsorbed water loss while weight gain at 400°C may be associated with oxygen uptake of reduced species, e.g., iron. The volatility of these glass samples provides evidence that no free liquids, free gases, organics, or explosives are released upon heating the waste glass to its glass transition temperature. |

## GLASS SUMMARIES

NO. 31

**TITLE:** The Determination of Pressure, Dewpoint, and Composition of the Gas Within the Free Volume of Canistered Waste Forms

**AUTHOR:** J.R. Harbour; T.J. Miller; M.J. Whitaker

**PUBLICATION:** WSRC-RP-90-1167

**DATE:** January 1, 1990

**ORGANIZATION:** Westinghouse Savannah River Co. Aiken, SC

**SUMMARY:** The free volume in four simulated canistered waste forms produced during Scale Glass Melter Campaigns was examined after being hermetically sealed for 2 to 3 years. The internal gas pressure, dewpoint temperature and gas composition were determined.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Glass, borosilicate   |
| <b>Waste Type:</b>                | Simulated high level radioactive defense waste  |
| <b>Waste Loading:</b>             | Nominal 28 wt%  |
| <b>Development Status:</b>        | Mature  |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | --  |
| <b>Biological Stability:</b>      | --  |
| <b>Leach Resistance:</b>          | --  |
| <b>Immersion Stability:</b>       | --  |
| <b>Free Liquids:</b>              | Relative humidities in canister 5-20 %, no free water.  |
| <b>Chemical Durability:</b>       | No unexpected chemical compound(s) found within sealed containers.  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | No free liquids or unexpected chemical compounds were identified within the sealed containers after several years of storage. The simulated waste form does not volatilize or emit chemical compounds, including water, during storage. |

## GLASS SUMMARIES

NO. 32

**TITLE:** Evaluation of the Potential for Gas Pressurization and Free Liquid Accumulation in a Canister from the West Valley Demonstration Project

**AUTHOR:** R.F. Hazelton; C.K. Thornhill; W.A. Ross

**PUBLICATION:** Ceramic Transactions, Nuclear Waste Management, IV, Vol. 23, pp. 491-500

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** A full-scale canister from WVDP was tested to determine potential for gas generation (non-radiolytic) and liquid accumulation. Heated for eight weeks above glass transition temperature.

|                            |  |
|----------------------------|--|
| Waste Form:                | --   |
| Waste Type:                | --   |
| Waste Loading:             | --   |
| Development Status:        | --   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | --   |
| Immersion Stability:       | --   |
| Free Liquid:               | No free liquid detected (increase in H <sub>2</sub> O of 0.00004 wt%)  |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | No unusual gasses detected nor generation of pressure.   |
| RCRA Compliance:           | --   |
| Conclusions:               | Test did not generate any gas pressurization. Oxygen depleted. Carbon dioxide increased. No free water detected. |

## GLASS SUMMARIES

NO. 33

**TITLE:** Results of Vitrifying Fernald K-65 Residue

**AUTHOR:** D.S. Janke; C.C. Chapman; R.A. Vogel

**PUBLICATION:** Ceramic Transactions, Nuclear Waste Management, Vol. 23, pp. 53-61

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society, Westerville, OH

**SUMMARY:** K-65 residue, which contains radium, uranium, daughter products, and heavy metals (lead), was vitrified on a bench-scale system which permitted analysis of the off-gas. TCLP, EP were measured.

**Waste Form:** Glass, soda iron barium silicate

**Waste Type:** Pitchblende residue, treated as TRU waste

**Waste Loading:** Approximately 80 wt%; volume reduction 60%

**Development Status:** Initial

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** See conclusions

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** Reduced by factor of 33,000; limited to amount from surface of glass.

**RCRA Compliance:** --

**Conclusions:** K-65 residue tested "hazardous" by EP toxicity test, with radon emanation rate (52,400 pCi/m<sup>2</sup>/s) after vitrification tested "non-hazardous" radon gas reduced by factor of 33,000 to 1.56 pCi/m<sup>2</sup>/s.

## GLASS SUMMARIES

NO. 34

**TITLE:** Solidification of Consolidated Incinerator Facility (CIF) Wastes in Soda-Lime-Silica (SLS) Glass: Use of Reactive Additives to Retain Hazardous and Heavy Metal Constituents

**AUTHOR:** C.M. Jantzen

**PUBLICATION:** Savannah River Lab, WSRC-TR-92-214, Rev. 0

**DATE:** April 30, 1992

**ORGANIZATION:** Westinghouse Savannah River Co.

**SUMMARY:** Discusses the vitrification of CIF blowdown and bottom kiln ash waste with available silica sources to form a stable soda lime glass waste form. The stirred electric melter is recommended as the melter system.

|                            |  |
|----------------------------|--|
| Waste Form:                | Glass, soda-lime   |
| Waste Type:                | Incinerator blow-down and ash, low level radioactive, hazardous, mixed   |
| Waste Loading:             | 45 to 50 wt%   |
| Development Status:        | Preliminary  |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | Expected to pass TCLP  |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | Wide variation in feed material  |
| Gas Generation:            | --   |
| RCRA Compliance:           | Expected to pass TCLP  |
| Conclusions:               | Vitrification suggested as a viable process for solidification of CIF blowdown and kiln ash waste. Only one additive, SiO <sub>2</sub> , is necessary for vitrification. |

## GLASS SUMMARIES

NO. 35

**TITLE:** Thermodynamic Approach to Glass Corrosion

**AUTHOR:** C.M. Jantzen

**PUBLICATION:** Corrosion of Glass, Ceramics, and Ceramic Superconductors, Principles, Testing, Characterization and Applications, D.E. Clark and B.K. Zaitos, eds., pp. 153-215, Noves Publications, Park Ridge, NJ

**DATE:** January 1, 1992

**ORGANIZATION:** Savannah River Site

**SUMMARY:** A review of theory of glass corrosion. Approaches glass corrosion from thermodynamic point of view. Discusses various parameter effects, e.g., SA/V, Eh, and pH.

Waste Form: --

Waste Type: --

Waste Loading: --

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: --

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

**Conclusions:** Hydration thermodynamics have wide applicability to predict the durability of natural, ancient, modern, and nuclear waste glass. The predicted durabilities correlate with those observed in nature and give a means for interpolation of the long-term durability of nuclear waste glasses.

**GLASS SUMMARIES****NO. 36**

**TITLE:** First Principles Process-Product Models for Vitrification of Nuclear Waste: Relationship of Glass Composition to Glass Viscosity, Resistivity, Liquidus Temperature, and Durability

**AUTHOR:** C.M. Jantzen

**PUBLICATION:** Ceramic Transactions, Vol. 23, pp. 37-51, Nuclear Waste Management IV, G.G. Wicks, et al., eds.

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** Development of first principle models for viscosity, electrical conductivity, liquidus temperature, and durability.

**Waste Form:** Glass, borosilicate

**Waste Type:** Simulated high level radioactive defense waste

**Waste Loading:** --

**Development Status:** Mature

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** None

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** The process/product models developed for the DWPF have been developed on first principle concepts of glass chemistry, solubility, precipitation, and thermodynamics. The models have been successfully used to control several Research Melter Campaigns at SRS (simulated and radioactive waste).

## GLASS SUMMARIES

NO. 37

**TITLE:** Systems Approach to Nuclear Waste Glass Development

**AUTHOR:** C.M. Jantzen

**PUBLICATION:** Journal of Non-Crystalline Solids, 84, No. 1-3, pp. 215-225

**DATE:** January 1, 1986

**ORGANIZATION:**

**SUMMARY:** A review and comparison among borosilicate, high-silica, and phosphate glasses indicated that borosilicate glass was most desirable for the immobilization of nuclear waste.

|                                   |               |
|-----------------------------------|---------------|
| <b>Waste Form:</b>                | Glass         |
| <b>Waste Type:</b>                | Nuclear waste |
| <b>Waste Loading:</b>             | --            |
| <b>Development Status:</b>        | --            |
| <b>Compressive Strength:</b>      | --            |
| <b>Thermal Stability:</b>         | --            |
| <b>Radiation Stability:</b>       | --            |
| <b>Biological Stability:</b>      | --            |
| <b>Leach Resistance:</b>          | --            |
| <b>Immersion Stability:</b>       | --            |
| <b>Free Liquids:</b>              | --            |
| <b>Chemical Durability:</b>       | --            |
| <b>Compositional Flexibility:</b> | --            |
| <b>Gas Generation:</b>            | --            |
| <b>RCRA Compliance:</b>           | --            |

**Conclusions:** A system evaluation of aluminosilicate glasses indicates that they have superior product characteristics, but are difficult to process. Evaluation of phosphate glasses indicates that they have adequate to poor product characteristics and are difficult to process. Borosilicate glass exhibits favorable product performance and ease of processability.



**GLASS SUMMARIES****NO. 38**

**TITLE:** Nuclear Waste Glass Product Consistency Test (PCT)

**AUTHOR:** C.M. Jantzen; N.E. Bibler

**PUBLICATION:** Savannah River Site

**DATE:** November 1, 1990

**ORGANIZATION:** Westinghouse Savannah River Co. Aiken, SC

**SUMMARY:** The PCT procedure as submitted to ASTM C26.13. The PCT is a method of quickly determining the durability and consistency of glasses. Proposed for use at DWPF.

**Waste Form:** Glass

**Waste Type:** --

**Waste Loading:** --

**Development Status:** Mature

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** 7 day 90°C deionized water.

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** A durability test, designated the Product Consistency Test (PCT) was developed for DWPF glass in order meet to the applicable waste acceptance specifications. The response of the PCT procedure was based on extensive testing with glasses of widely different compositions. The PCT was determined to be very reproducible, to yield reliable results rapidly and to be easily performed in shielded cell facilities with radioactive samples.

## GLASS SUMMARIES

NO. 39

**TITLE:** The Product Consistency Test for the DWPF Wasteform

**AUTHOR:** C.M. Jantzen; N.E. Bibler

**PUBLICATION:** Proceedings of the 2nd International Seminar on Radioactive Waste Products, E. Odoj, et al., eds., pp. 609-622

**DATE:** June 1, 1990

**ORGANIZATION:** Kernforschunganlanger (KFA) Julich, FRG

**SUMMARY:** Development of Product Consistency Test (PCT). Comparison to MCC-1 as well as other tests, results of round-robin evaluations.

**Waste Form:** Glass, borosilicate

**Waste Type:** Simulated high level radioactive defense waste

**Waste Loading:** --

**Development Status:** Mature

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Log norm. Boron compared to free energy of hydration, results of round-robin evaluation. Approximately 10% difference between laboratories.

**GLASS SUMMARIES****NO. 40**

**TITLE:** Leaching of Polyphase Nuclear Waste Ceramics: Microstructural and Phase Characterization

**AUTHOR:** C.M. Jantzen; D.R. Clarke; P.E.D. Morgan; A.B. Harker

**PUBLICATION:** Journal of the American Ceramic Society, Vol. 65, No. 6, pp. 292-300

**DATE:** June 1, 1982

**ORGANIZATION:** Rockwell International Science, Ctr.

**SUMMARY:** The leaching of complex polyphase nuclear waste ceramics is described in the context of the geochemically established dissolution behavior of the constituent phases.

**Waste Form:** Ceramic, tailored for SRS defense waste

**Waste Type:** High-level defense waste

**Waste Loading:** 30 - 60 wt%

**Development Status:** Initial

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Dissolution was incongruent and controlled by the solubilities of the individual phases. Necessary to avoid intergranular glass phases due largely to trace amounts of silica and microcracking of the principle phases during processing.

## GLASS SUMMARIES

NO. 41

**TITLE:** Thermodynamic Model of Natural, Medieval and Nuclear Waste Glass Durability

**AUTHOR:** C.M. Jantzen; M.J. Plodinec

**PUBLICATION:** Journal of Non-Crystalline Solids, 67, pp. 207-223

**DATE:** January 1, 1984

**ORGANIZATION:** Elsevier Science Publishers, North Holland, Amsterdam

**SUMMARY:** A thermodynamic model of glass durability based on the hydration of "structural units" has been applied to natural glass, medieval window glasses, and glasses containing nuclear waste.

**Waste Form:** Glass, general

**Waste Type:** --

**Waste Loading:** --

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** See summary

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** A method of calculating glass durability based on the free energy of hydration is presented. The contribution of the redox level and alumina coordination must also be considered. Glass compositions which raise the pH of leachate above 10 must be corrected with an additional free energy term. Waste glass is compared to known glasses which were formed 500 to 0.5 million years ago.

## GLASS SUMMARIES

NO. 42

**TITLE:** Comparison of Surface Layers Formed on Synthetic Basaltic Glass, French R7T7 and HMI Borosilicate Nuclear Waste Form Glasses-Materials Interface Interactions Tests-Waste Isolation Pilot Plant

**AUTHOR:** M.J. Jercinovic; S.A. Kaser; R.C. Ewing; W. Lutze

**PUBLICATION:** Material Research Society Symposium Proceedings, 176, pp. 355-362

**DATE:** January 1, 1990

**ORGANIZATION:** Department of Geology, University of New Mexico; HMI, Berlin

**SUMMARY:** Analysis of secondary phases formed on basalt, R7T7, and HMI glasses reacted up to 2 years.

|                            |   |
|----------------------------|---|
| Waste Form:                | Borosilicate glass, basalt as natural analogue  |
| Waste Type:                | --  |
| Waste Loading:             | --  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Layer compositions reflect glass compositions, although some elements are incorporated from leachant. |

## GLASS SUMMARIES

NO. 43

**TITLE:** Basaltic Glasses From Iceland and the Deep Sea: Natural Analogues to Borosilicate Nuclear Waste-Form

**AUTHOR:** M.J. Jercinovic; R.C. Ewing

**PUBLICATION:** JSS 88-01

**DATE:** January 1, 1988

**ORGANIZATION:** Department of Geology, University of New Mexico

**SUMMARY:** Provides characterization of reaction processes and products for reaction of natural basaltic glasses.

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Basalt glass as analogue for borosilicate glass  |
| <b>Waste Type:</b>                | --   |
| <b>Waste Loading:</b>             | --   |
| <b>Development Status:</b>        | --   |
| <b>Compressive Strength:</b>      | --   |
| <b>Thermal Stability:</b>         | --   |
| <b>Radiation Stability:</b>       | --   |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | --   |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | --   |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | Amorphous and crystalline products form on glass surface but do not protect underlying glass from further attack. Supports Grambow's hypothesis of two characteristic rate coefficients for initial rate in dilute solutions and final rate in "silica-saturated" solutions. |

## GLASS SUMMARIES

NO. 44

**TITLE:** Silicate Glasses

**AUTHOR:** W. Lutze

**PUBLICATION:** Radioactive Waste Forms for the Future, W. Lutze, R.C. Ewing, eds., Elsevier Science Publishers, B.V.

**DATE:** January 1, 1988

**ORGANIZATION:** Hahn-Meitner-Institute, Berlin

**SUMMARY:** Review of silicate glasses as used in high-level waste disposal. Review of mechanical and chemical properties of U.S. and foreign glasses.

**Waste Form:** Borosilicate glass, many different compositions discussed

**Waste Type:** Reprocessed high-level waste

**Waste Loading:** --

**Development Status:** Advanced

**Compressive Strength:** --

**Thermal Stability:** Glass will crack as it cools

**Radiation Stability:** Radiation effects in glass are small, in leachate lead to modest increase

**Biological Stability:** --

**Leach Resistance:** In general 0.5-5 g/m<sup>2</sup>/d at 90° deionized water initially

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** Nitric acids from radiolysis increase rate slightly; phase separation possible at high waste loadings

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Glass composition determined primarily by processing techniques; overloading with waste may lead to phase separation; glass will fracture as it cools; radiation does not significantly affect glass durability (enhancement less than 5 times); different glasses have different corrosion rates, which decrease with time; solid alteration products may affect reaction.

## GLASS SUMMARIES

NO. 45

**TITLE:** Radioactive Waste Forms for the Future

**AUTHOR:** W. Lutze; R.C. Ewing

**PUBLICATION:** Book, Elsevier Science Publishers, B.V., pp. 699-739

**DATE:** January 1, 1988

**ORGANIZATION:**

**SUMMARY:** A review of high level waste forms. Compares borosilicate glass and other glasses, Synroc and other ceramics, and FUETAP cement. Provides a summary of physical and chemical properties.

**Waste Form:** Glass, borosilicate, Synroc, tailored ceramics, FUETAP

**Waste Type:** High-level defense waste

**Waste Loading:** 10-60 wt%

**Development Status:** Varied

**Compressive Strength:** Ceramic~550 MPa; FUETAP ~20 MPa

**Thermal Stability:** --

**Radiation Stability:** %vol change: borosilicate  $\pm 1.5$ , Synroc +8

**Biological Stability:** --

**Leach Resistance:** Borosilicate  $< 1 \text{ g/m}^2/\text{d}$ ; difficult to compare forms due to radiation effects

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Borosilicate: large data base, corrosion mechanism well understood, long term corrosion validated by natural glasses, industrial scale fabrication has been demonstrated.

Synroc: well-characterized, more stable at elevated temperatures in hydrothermal environment better mechanical integrity, no flow rate dependence for leaching, natural analogues available.



## GLASS SUMMARIES

NO. 46

TITLE: Chemical Corrosion of Lead-Iron Phosphate Glass

AUTHOR: W. Lutze; B. Grambow

PUBLICATION: In-process

DATE:

ORGANIZATION: Hahn-Meitner Institute, Berlin

## SUMMARY:

Waste Form: Phosphate glass, lead-iron  
Waste Type: LWR commercial waste  
Waste Loading: 6.4 wt%  
Development Status: Initial  
Compressive Strength: --  
Thermal Stability: --  
Radiation Stability: --  
Biological Stability: --  
Leach Resistance: Max rate 0.05 g/m<sup>2</sup>/day (MCC-1, 90°C, deionized water)  
Immersion Stability: --  
Free Liquids: --  
Chemical Durability: --  
Compositional Flexibility: --  
Gas Generation: --  
RCRA Compliance: --  
Conclusions: Leach rate controlled by Pb<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>. Lead-iron phosphate glass more stable than borosilicate due to lower solubility of Pb<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> compared to H<sub>4</sub>SiO<sub>4</sub>, which controls rate of borosilicate reaction.

**GLASS SUMMARIES**

NO. 47

**TITLE:** A Comparison of the Behavior of Vitrified HLW in Repositories in Salt, Clay, and Granite. Part II: Results

**AUTHOR:** J.A.C. Marples; W. Lutze; M. Kawanishi; P. Van Iseghem

**PUBLICATION:** Material Research Society Symposium Proceedings, 176, pp. 275-282

**DATE:** January 1, 1990

**ORGANIZATION:**

**SUMMARY:** Results of round-robin with R7T7 glass at 900°C in salt, clay, or granite at 90°C.

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Borosilicate glass   |
| <b>Waste Type:</b>                | --   |
| <b>Waste Loading:</b>             | --   |
| <b>Development Status:</b>        | --   |
| <b>Compressive Strength:</b>      | --   |
| <b>Thermal Stability:</b>         | --   |
| <b>Radiation Stability:</b>       | --   |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | --   |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | --   |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | Glass corroded at about the same rate in clay and granite systems, but about 10 times faster in salt. Agreement between labs on absolute releases is poor. |

## GLASS SUMMARIES

NO. 48

**TITLE:** Comparison of SRP High-Level Waste Disposal Costs for Borosilicate Glass and Crystalline Ceramic Waste Forms

**AUTHOR:** W.R. McDonell

**PUBLICATION:** DuPont, Savannah River Laboratory, DPST-82-346

**DATE:** April 1, 1982

**ORGANIZATION:** DuPont, Savannah River Laboratory

**SUMMARY:** Comparison of costs for immobilization and disposal HLW indicates that the borosilicate waste form is less costly than the crystalline waste form. Waste disposal for ceramic less costly due to fewer canisters, however not sufficient to offset the higher development and processing costs.

|                            |   |
|----------------------------|---|
| Waste Form:                | Glass/borosilicate - ceramic/Synroc-D   |
| Waste Type:                | High level radioactive defense waste  |
| Waste Loading:             | Glass - 28 wt%; ceramic - 65 wt% (52% after Al wash)  |
| Development Status:        | Glass - mature; ceramic - intermediate  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Costs dependent on size of ceramic body, development costs. Must consider all costs, intermediate storage, shipping, final storage, and monitoring. The cost reductions due to lesser number of ceramic canisters did not offset the higher development and processing costs. |

## GLASS SUMMARIES

NO. 49

**TITLE:** A Comparison of the Behavior of Vitrified HLW in Repositories in Salt, Clay, and Granite, I: Experimental

**AUTHOR:** L.A. Mertens; W. Lutze; J.A.C. Marples; P. Van Iseghem; E. Vernaz

**PUBLICATION:** Material Research Society Symposium Proceedings, 176, pp. 267-274

**DATE:** January 1, 1990

**ORGANIZATION:**

**SUMMARY:** Round-robin results for leach tests with R7T7 glass at 90°C.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Borosilicate glass  |
| <b>Waste Type:</b>                | --  |
| <b>Waste Loading:</b>             | --  |
| <b>Development Status:</b>        | --  |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | --  |
| <b>Biological Stability:</b>      | --  |
| <b>Leach Resistance:</b>          | --  |
| <b>Immersion Stability:</b>       | --  |
| <b>Free Liquids:</b>              | --  |
| <b>Chemical Durability:</b>       | --  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | Comparison of results from different labs shows large differences in highly soluble species of B, Li, and Mo. |

**GLASS SUMMARIES**

NO. 50

**TITLE:** Surface Analyses of SRS Waste Glass Buried for up to Two Years in Limestone in the United Kingdom

**AUTHOR:** C.G. Namboodri; G.G. Wicks; A.R. Ledding; L.L. Hench; R.G. Newton

**PUBLICATION:** Ceramic Transactions, Vol. 23, Nuclear Waste Management IV, G.G. Wicks, ed., American Ceramic Society, Westerville, OH, pp. 653-662

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society

**SUMMARY:** Evaluation of borosilicate glass after one and two years at ambient temperature. For all samples, the glass interaction zone was less than 1 micron.

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Glass, borosilicate  |
| <b>Waste Type:</b>                | Simulated high-level radioactive waste.  |
| <b>Waste Loading:</b>             | 28 wt%   |
| <b>Development Status:</b>        | --   |
| <b>Compressive Strength:</b>      | --   |
| <b>Thermal Stability:</b>         | --   |
| <b>Radiation Stability:</b>       | --   |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | --   |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | --   |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | Less than one micron interaction layer after two years at ambient temperature. |

## GLASS SUMMARIES

NO. 51

**TITLE:** Development of a Combined Soil Wash/In-Furnace Vitrification System for Soil Remediation at DOE Sites

**AUTHOR:** I.L. Pegg; W. Greenman; Y. Guo; I.S. Muller; R. K. Mohr; P.B. Macedo; D.C. Grant; P.R. Mullik

**PUBLICATION:** First International Mixed Waste Symposium, ASME

**DATE:** August 1, 1991

**ORGANIZATION:** Duratek Corp, CUA, Westinghouse

**SUMMARY:** Vitrification and soil washing technologies are briefly reviewed and compared to other technologies for remediation of contaminated soils and sludges.

|                            |  |
|----------------------------|--|
| Waste Form:                | Glass, borosilicate  |
| Waste Type:                | Low-level, mixed, and TRU  |
| Waste Loading:             | 60-80 % loading blended streams  |
| Development Status:        | Initial  |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | --   |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | While the costs associated with waste treatment are important components in a comparison of treatment technologies, they represent only a part of the total costs for site remediation. Disposal and long-term monitoring must also be considered. Recommends washing of contaminated soils and slurries, capture of radioactive and hazardous materials, followed by vitrification. |

## GLASS SUMMARIES

NO. 52

**TITLE:** Hydrated-Layer Formation During Dissolution of Complex Silicate Glasses and Minerals

**AUTHOR:** J.C. Petit; D.G. Mea; J.C. Dran; M.C. Maganthier; P.A. Mando; A. Paccagnelia

**PUBLICATION:** Geochim-Cosmochim. Acta, 54, pp. 1941-1955

**DATE:** January 1, 1990

**ORGANIZATION:** SESD/LECALT, CEN-FAR, France

**SUMMARY:** Study of the dissolution behavior of three complex silicate glasses and five crystalline silicates after implantation of Pb ions to produce dislocations.

|                            |   |
|----------------------------|---|
| Waste Form:                | Borosilicate glass  |
| Waste Type:                | Simulated   |
| Waste Loading:             | None  |
| Development Status:        | Experimental  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Hydrated layers form on all samples in which heavy elements accumulate. |

## GLASS SUMMARIES

NO. 53

TITLE: Integrated Testing of the SRL-165 Glass Waste Form

AUTHOR: D.L. Phinney; F.J. Ryerson; V.M. Oversby; W.A. Lanford; R.D. Aines; J.K. Bates

PUBLICATION: Material Research Society Symposium Proceedings, 84, pp. 433-446

DATE: January 1, 1986

ORGANIZATION: Lawrence Livermore National Laboratory, Argonne National Laboratory

SUMMARY: Mass-balance analyses of reacted SRL-165 glass to account for alteration layer composition, tuff rock and steel vessel surfaces, and leachate.

|                            |  |
|----------------------------|--|
| Waste Form:                | Borosilicate glass, SRL-165  |
| Waste Type:                | Doped with $^{237}\text{Np}$ , $^{239}\text{Pu}$ and $^{242}\text{Am}$   |
| Waste Loading:             | --   |
| Development Status:        | --   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | 1.5 to 3.0 micron/yr at 90°C at S.A.=1/10m in J-13 solution  |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | Glass reacts via ion exchange, followed by surface dissolution. Actinides sorbed onto teflon and steel vessel surfaces. U and Li diffused into tuff. |



## GLASS SUMMARIES

NO. 54

**TITLE:** Viscosity of Glasses Containing Simulated Savannah River Plant Waste

**AUTHOR:** M.J. Plodinec

**PUBLICATION:** Savannah River Laboratory, DP-1507

**DATE:** August 11, 1978

**ORGANIZATION:** E.I. DuPont, Savannah River Laboratory, Aiken, SC

**SUMMARY:** The viscosity of glass melts containing four simulated sludge types and two frits were measured over the temperature range 750-1200°C.

**Waste Form:** Glass, borosilicate

**Waste Type:** Simulated high level radioactive defense waste

**Waste Loading:** 25 to 30 wt%

**Development Status:** Preliminary (1978)

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** None

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Limits for viscosity, sludge loading, and devitrification were defined. Melts containing devitrification were found to be non-Newtonian.

## GLASS SUMMARIES

NO. 55

**TITLE:** The Physical and Thermal Properties of Simulated Nuclear Waste Glasses and Their Melts

**AUTHOR:** L.D. Pye

**PUBLICATION:** Savannah River Laboratory, DPST-85-397

**DATE:** February 1, 1985

**ORGANIZATION:** Alfred University

**SUMMARY:** The physical and thermal properties of three simulated nuclear waste glasses (165), high-iron, TDS, and high-alumina were measured.

|                            |  |
|----------------------------|--|
| Waste Form:                | Glass, borosilicate  |
| Waste Type:                | Simulated high level radioactive defense waste   |
| Waste Loading              | Nominal 28 wt%   |
| Development Status:        | Mature   |
| Compressive Strength:      | M.O.R. 3700 to 6200 psi  |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | --   |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | Density @ 23°C, 2.6 to 2.8; coefficient of linear expansion <300°C 9.9 to 10 ppm; also, values for thermal diffusivity, Young's modulus, shear modulus, Poisson's ratio, specific heat, and viscosity. |

## GLASS SUMMARIES

NO. 56

**TITLE:** EPA Tests of Simulated DWPF Waste Glass-U

**AUTHOR:** A.A. Ramsey

**PUBLICATION:** WSRC-TR-90-22, Inter Office Memo to M.J. Plodinec.

**DATE:** January 10, 1990

**ORGANIZATION:** Westinghouse Savannah River Co.

**SUMMARY:** The simulated DWPF waste glasses tested passed both the EP Toxicity and the TCLP tests. The glasses were prepared by doping two to three times the expected amounts of Cr, Pb, Ba, Ag, Se, and Cd. The resulting material is not a hazardous waste.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Simulated high-level waste borosilicate glass.            |
| <b>Waste Type:</b>                | Simulated radioactive, actual hazardous elements          |
| <b>Waste Loading:</b>             | --  |
| <b>Development Status:</b>        | --  |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | --  |
| <b>Biological Stability:</b>      | --  |
| <b>Leach Resistance:</b>          | Passed TCLP and EP Tox tests                              |
| <b>Immersion Stability:</b>       | --  |
| <b>Free Liquids:</b>              | --  |
| <b>Chemical Durability:</b>       | --  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | Non-hazardous   |
| <b>Conclusions:</b>               | Simulated high level waste form is not a hazardous waste. |

## GLASS SUMMARIES

NO. 57

**TITLE:** Predictive Modeling of Leachate pH for Simulated High Level Waste Glass

**AUTHOR:** W.G. Ramsey; T.D. Taylor; C.M. Jantzen

**PUBLICATION:** Ceramic Transactions, Vol. 23, Nuclear Waste Management, pp. 105-114

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** A methodology is proposed for predicting the equilibrium pH of leachates from PCT experiments in deionized water. Glasses in the  $\text{SiO}_2\text{-Al}_2\text{O}_3\text{-B}_2\text{O}_3\text{-Fe}_2\text{O}_3\text{-CaO-Na}_2\text{O}$  system were tested.

|                            |  |
|----------------------------|--|
| Waste Form:                | Glass  |
| Waste Type:                | --   |
| Waste Loading:             | --   |
| Development Status:        | --   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | Leach resistance largely dependent on the pH of leachate                             |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | It was possible to predict the equilibrium pH based solely on the glass composition. |

## GLASS SUMMARIES

NO. 58

**TITLE:** Materials Characterization Center, Workshop on the Irradiation Effects in Nuclear Waste Forms - Summary Report

**AUTHOR:** F.P. Roberts; R.P. Turcotte; W.J. Weber

**PUBLICATION:** US DOE Report, PNL-3588

**DATE:** January 1, 1981

**ORGANIZATION:** Pacific Northwest Laboratory, Richland, Washington

**SUMMARY:** The workshop considered the utility of the proposed MCC-6 irradiation test and the effect of various types of radiation (alpha, beta, gamma, and ionizing) and transmutation effects on the waste form. Considered actinide-doping the waste form. Among the important property changes caused by irradiation are those that lead to leachability.

|                            |   |
|----------------------------|---|
| Waste Form:                | Glass and ceramic/fuel  |
| Waste Type:                | Doping of $^{238}\text{Pu}$ or $^{244}\text{Cm}$  |
| Waste Loading:             | --  |
| Development Status:        | Initial   |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | Density change < 1% for saturation doses. Damage slight.  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Alpha decay is probably the principal contributor to structural damage. MCC-6 considers approximately $3 \times 10^{18}$ alpha decays/cc while HLW would only receive $10^{17}$ to $5 \times 10^{17}$ . Should consider dose equivalent to that received in one million years. Density changes for glass waste generally do not exceed 1%. Studies reported that structural changes induced by alpha decay do not adversely affect leach rates in glass waste forms. Leaching increased less than 2 times. Studies indicated that radiolysis can produce nitric acid in air-saturated water. This can affect leaching. Ceramics showed larger volume changes. |

## GLASS SUMMARIES

NO. 59

**TITLE:** Comparative Assessment of TRU Waste Forms and Processes, Vol. I: Waste Form and Process Evaluations

**AUTHOR:** W.A. Ross; R.O. Lokken; R.P. May; F.P. Roberts; C.L. Timmerman; R.L. Treat; J.H. Westik

**PUBLICATION:** US DOE Report PNL-4428, Vol. I

**DATE:** September 1, 1982

**ORGANIZATION:** Pacific Northwest Laboratory, Richland Washington

**SUMMARY:** This study provides an assessment of seven waste forms and eight processes for immobilizing TRU wastes. Includes preparation and characterization of TRU-containing waste forms. Waste forms were cast cement, cold pressed cement, FUETAP cement, borosilicate glass, aluminosilicate glass, basalt glass-ceramic, and pressed and sintered ceramic. Properties, processes, and costs were compared.

|                            |  |
|----------------------------|--|
| Waste Form:                | Eight types noted in summary   |
| Waste Type:                | TRU waste, 3 parts sludge, 1 part incinerator ash                        |
| Waste Loading:             | >20 %  |
| Development Status:        | --   |
| Compressive Strength:      | Cement 3-8; glass 35-40; ceramic 48-60 MPa. MCC-11 (Tensile)             |
| Thermal Stability:         | Glass and ceramic-no weight loss to 800°C; cement-substantial loss       |
| Radiation Stability:       | Cumulative radiation doses for time >10 <sup>5</sup> yrs not significant |
| Biological Stability:      | --   |
| Leach Resistance:          | --   |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | Cements produce gases due to radiation or heat                           |
| RCRA Compliance:           | --   |

## GLASS SUMMARIES

NO. 59 (Cont.)

## Conclusions:

Cement had the lowest release of Pu, but was otherwise less durable than the glasses or ceramics. Leachate pH appeared to play an important role in the releases from the waste forms. Differences in waste forms are not more than two orders of magnitude. All seven waste forms may exceed proposed performance objectives. The maximum annual release of radionuclides should be less than  $10^{-5}$  of the maximum inventory of that radionuclide. The cement waste forms had the lowest strength, followed by glass, then ceramics. Overall rating: ceramics, glasses, cements. All would be acceptable for WIPP or NRC requirements. Also evaluated processes.

## GLASS SUMMARIES

NO. 60

TITLE: Environmental Assessment, Waste Form Selection for SRP High-Level Waste

AUTHOR: Savannah River Laboratory

PUBLICATION: U.S. Department of Energy, DOE/EA-0179

DATE: July 1, 1982

ORGANIZATION: Department of Energy

SUMMARY: Discusses physical characteristics of borosilicate glass as the waste form for high-level radioactive waste and compares borosilicate glass with crystalline ceramic.

Waste Form: Glass, borosilicate

Waste Type: High level radioactive defense waste

Waste Loading: 28 wt%

Development Status: Mature

Compressive Strength: 550 MPa. Fraction of fines from 10 J/cm<sup>3</sup> - 0.14 to 0.18% < 10 $\mu$

Thermal Stability: Will crack on cooling from melt, may increase leach by 5 times

Radiation Stability: Performance not affected by self-irradiation for periods of 10<sup>6</sup> yrs

Biological Stability: Inert

Leach Resistance: Steady state at 25 to 55°C, 10<sup>-3</sup> to 10<sup>-4</sup> g/m<sup>2</sup>/day

Immersion Stability: Generally forms protective layer

Free Liquids: No free liquids.

Chemical Durability: Little effect between pH 5 to 9

Compositional Flexibility: Stable/inert

Gas Generation: Should be maintained in structure of glass

RCRA Compliance: --

Conclusions: Initial leach, 10<sup>-1</sup> to 10<sup>-3</sup> g/m<sup>2</sup>/day, steady state 10<sup>-3</sup> to 10<sup>-4</sup>. Discusses influence of temperature, pH, water, etc. Discusses impact resistance. Compares borosilicate glass to Synroc-D



## GLASS SUMMARIES

NO. 61

**TITLE:** Solubilities of Nickel and Cobalt Chalcogenides in a Nuclear Waste Glass

**AUTHOR:** H.D. Schreiber; E.D. Sisk; C.W. Schreiber; J.K. Burns

**PUBLICATION:** Ceramic Transactions, Vol.23, Nuclear Waste Management, pp. 213-222

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** Measured the solubilities of NiS, NiSe, NiTe, CoS, and CoSe in SRL Frit 131 as a function of temperature and oxygen fugacity.

**Waste Form:** Glass, borosilicate

**Waste Type:** Simulated high level radioactive defense waste

**Waste Loading:** Nominal 28 wt%

**Development Status:** Mature

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** Nickel sulfide and nickel selenide are both relatively insoluble in the glass

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** The solubilities decreased with decreasing oxygen fugacity. In general, Ni<sub>3</sub>S<sub>2</sub> is the least soluble of the compounds in SRL-131.

## GLASS SUMMARIES

NO. 62

**TITLE:** DWPF Batch 1, Waste Glass Investigations

**AUTHOR:** R.F. Schumacher

**PUBLICATION:** Ceramic Transactions, Vol. 23, Nuclear Waste Management, G.G.Wicks, ed., pp. 453-463

**DATE:** May 1, 1991

**ORGANIZATION:** American Ceramic Society

**SUMMARY:** Discusses the effects of variability in the DWPF feed mixture on glass properties, including solubility of sludge and copper-nickel precipitation.

**Waste Form:** Glass, borosilicate

**Waste Type:** High level radioactive defense waste

**Waste Loading:** 10 to 40 wt%

**Development Status:** Mature

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** Limits on chrome, nickel, and copper

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Limited to 30 wt% sludge, copper limited to about 0.30 wt%.

## GLASS SUMMARIES

NO. 63

**TITLE:** In Situ Vitrification of Buried Waste Sites

**AUTHOR:** J.W. Shade; L.E. Thompson; C.H. Kindle

**PUBLICATION:** Ceramic Transactions, Vol. 23, Nuclear Waste Management, G.G. Wicks, ed., American Ceramic Society, Westerville, OH, pp. 633-640

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society

**SUMMARY:** Discusses ISV of buried wastes. Stabilizes or volatilizes organic chemicals, stabilizes toxic elements.

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Glass, ISV   |
| <b>Waste Type:</b>                | Mixed waste  |
| <b>Waste Loading:</b>             | --   |
| <b>Development Status:</b>        | --   |
| <b>Compressive Strength:</b>      | --   |
| <b>Thermal Stability:</b>         | --   |
| <b>Radiation Stability:</b>       | --   |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | --   |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | --   |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | Testing to date indicates that ISV is a promising remediation technology for application to buried wastes. It is capable of processing a wide variety of waste types. The resulting product has been shown to meet EPA requirements. |

## GLASS SUMMARIES

NO. 64

**TITLE:** Secondary Lead Smelter Slags: Minimizing Lead Release Levels

**AUTHOR:** E.S. Shenkler; S. Graham; V.A. Greenhut

**PUBLICATION:** Ceramic Transactions, Vol. 23, Nuclear Waste Management IV, G.G. Wicks, ed., American Ceramics Society, Westerville, Ohio, pg. 75-84

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society

**SUMMARY:** Secondary lead slags and mattes were formulated into glass compositions in order to stabilize the material with regard to the lead release levels as established by the EPA TCLP.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Glass   |
| <b>Waste Type:</b>                | Lead waste  |
| <b>Waste Loading:</b>             | --  |
| <b>Development Status:</b>        | --  |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | --  |
| <b>Biological Stability:</b>      | --  |
| <b>Leach Resistance:</b>          | Approximately 50 times improvement  |
| <b>Immersion Stability:</b>       | --  |
| <b>Free Liquids:</b>              | --  |
| <b>Chemical Durability:</b>       | --  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | An approximate 50 times improvement in leach resistance is possible. Varied percentage of network former. |

## GLASS SUMMARIES

NO. 65

**TITLE:** Reference Commercial High-Level Waste Glass and Canister Definition

**AUTHOR:** S.C. Slate; W.A. Ross; W.L. Partain

**PUBLICATION:** USDOE, PNL-3838

**DATE:** September 1, 1981

**ORGANIZATION:** Pacific Northwest Laboratory

**SUMMARY:** Presents technical data and performance characteristics of a high-level waste glass and canister intended for use in the design of a complete waste encapsulation package suitable for disposal of a waste product which would be produced in a commercial nuclear-fuel reprocessing plant.

**Waste Form:** Glass, borosilicate

**Waste Type:** High level radioactive commercial waste/intermediate level liquid

**Waste Loading:** 31 wt%

**Development Status:** Initial

**Compressive Strength:** Estimated 37 to 51 MPa diametric compression; also discusses impact tests

**Thermal Stability:** Below 500°C, diffusion processes very slow

**Radiation Stability:** Radiation effects discounted; no change in leach  $3 \times 10^{17}$  d/cc

**Biological Stability:** --

**Leach Resistance:**  $2 \times 10^{-6}$  to  $45 \times 10^{-6}$  g/cm<sup>2</sup> day Cs release (simulated and radioactive)

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** Very slow diffusion release of helium

**RCRA Compliance:** --

**Conclusions:** Laboratory research has indicated the practicality of reprocessing fuel for the efficient recovery of uranium and plutonium, along with a minimal addition of non-radioactive chemicals. The HLW composition is based on the Allied General Nuclear Services plant.

## GLASS SUMMARIES

NO. 66

**TITLE:** Mineralogical, Textural and Compositional Data on the Alteration of Basaltic Glass From Kilauea, Hawaii to 300°C: Insights to the Corrosion of a Borosilicate Glass

**AUTHOR:** D.K. Smith

**PUBLICATION:** Material Research Society Symposium Proceedings, 212, pp. 115-121

**DATE:** 1991

**ORGANIZATION:** Lawrence Livermore National Laboratory

**SUMMARY:** Analysis of solids in palagonite layer of naturally altered basalt glass shows that smectite, chlorites, and actinolite form. Adopts dissolution/precipitation mechanism to describe banding of phases in layer.

|                            |   |
|----------------------------|---|
| Waste Form:                | Basalt glass as analogue of borosilicate glass (SRL 165)  |
| Waste Type:                | --  |
| Waste Loading:             | --  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Discusses secondary minerals found in palagonite layer. Critically compares basalt as analogue for HLW. |

**GLASS SUMMARIES****NO. 67**

**TITLE:** Travel Report - Summary of Talk given at the SAIC OTD Thermal Working Meeting in Salt Lake City

**AUTHOR:** M. E. Smith

**PUBLICATION:** Savannah River Lab, WSRC-RP-91-1092

**DATE:** October 30, 1991

**ORGANIZATION:** Westinghouse Savannah River Co, Aiken, SC

**SUMMARY:** Comparison of total costs of two waste forms, cement and glass, for the treatment of consolidated incinerator facility blow down. Vitrification can be as cost effective as cementation when the long term maintenance costs of the waste form are considered.

**Waste Form:** Glass, cementitious

**Waste Type:** Incinerator blow-down

**Waste Loading:** --

**Development Status:** Conceptual

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** The treatment of low level waste by mixing with cement has, in the past, been considered more cost effective than the vitrification process. Vitrification can be as cost effective as cementation for this particular case.

## GLASS SUMMARIES

NO. 68

TITLE: Impact Testing of Vitreous Simulated High-Level Waste in Canisters

AUTHOR: T.H. Smith; W.A. Ross

PUBLICATION: USDOE BNWL-1903

DATE: May 1, 1975

ORGANIZATION: Battelle, Pacific Northwest Laboratories

SUMMARY: A portion of a risk analysis for accidental release of radioisotopes. Investigated the impact breakup characteristics of encapsulated waste glass. Tests were of a simulated waste glass in cylindrical 304L stainless steel canisters. Ten cylinders impacted at velocities of 25 and 44 fps. Twenty two smaller specimens were tested at velocities of 25, 44, 66 and 117 fps. Tested some of small canisters at 800°F.

Waste Form: --

Waste Type: --

Waste Loading: --

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: --

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Canisters breached only at the two highest velocities 66 and 117 fps. Breaches were all very small cracks. The inventory fraction smaller than 10 micron typically ranged from  $10^{-8}$  to  $10^{-4}$  for the 80 mph impact. The surface area typically increased by only a few percent of the initial surface area, but by a factor of 40 for the 117 fps impact (80 mph).



## GLASS SUMMARIES

NO. 69

**TITLE:** Optimization of Glass Composition for the Vitrification of Nuclear Waste

**AUTHOR:** P.D. Soper; D.D. Walker; M.J. Plodinec; G.J. Roberts; L.F. Lightner

**PUBLICATION:** American Ceramic Society Bulletin, Vol. 62, No. 9, pp. 1013-1028, DP-MS-81-108

**DATE:** September 1, 1983

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** Development of the early DWPF sludge only glass frits, (Types 131-166) Twenty-five trials leading to optimization of Frit 165, basis for selection of composition. A statistical approach to glass development.

|                            |   |          |          |
|----------------------------|---|----------|----------|
| Waste Form:                | Glass, borosilicate   |          |          |
| Waste Type:                | Simulated high level radioactive defense                          |          |          |
| Waste Loading:             | Nominal 28 wt%  |          |          |
| Development Status:        | Early   |          |          |
| Compressive Strength:      | --  |          |          |
| Thermal Stability:         | --  |          |          |
| Radiation Stability:       | --  |          |          |
| Biological Stability:      | --  |          |          |
| Leach Resistance:          | 0.125 to 0.586 g/m <sup>2</sup> /day                              |          |          |
| Immersion Stability:       | --  |          |          |
| Free Liquids:              | None  |          |          |
| Chemical Durability:       | --  |          |          |
| Compositional Flexibility: | --  |          |          |
| Gas Generation:            | --  |          |          |
| RCRA Compliance:           | --  |          |          |
| Conclusions:               | Leach Test: Static 90°C- -40/+60 mesh, pH buffer solution, 4,7,10 |          |          |
|                            | S.A. 0.007  | Frit 131 | Frit 165 |
|                            | (g/m <sup>2</sup> /day)   |          |          |
|                            | pH 4.00   | 0.586    | 0.165    |
|                            | pH 7.00   | 0.161    | 0.125    |
|                            | pH 10.00  | 0.458    | 0.477    |

## GLASS SUMMARIES

NO. 70

**TITLE:** An Experimental Comparison of Alternative Solid Forms for Savannah River High-Level Wastes

**AUTHOR:** J.A. Stone

**PUBLICATION:** Scientific Basis for Nuclear Waste Management, Annual Meeting of the Material Research Society, DP-MS-81-102

**DATE:** November 16, 1981

**ORGANIZATION:** Material Research Society, 4th International Symposium, Scientific Basis for Nuclear Waste Management

**SUMMARY:** Compares leachability of alternative waste forms: borosilicate glass, high silica glass, tailored ceramic, and Synroc-D ceramic.  
 Uranium SR<TC<BS<HS  
 Cesium HS<BS<SR

**Waste Form:** Borosilicate and high silica glass, tailored and Synroc-D ceramic

**Waste Type:** High level radioactive defense waste

**Waste Loading:** Glasses ~28 wt%; ceramics ~59 wt%

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** Irradiated samples

**Biological Stability:** --

**Leach Resistance:** MCC-1 28 day @ 90°C, deionized water, silicate water, brine

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** No single waste form had a clear advantage for leach resistance. Ceramic retained U best, while high silica retained Cs best. Risk analyses indicated that all the forms tested should be satisfactory.

## GLASS SUMMARIES

NO. 71

**TITLE:** Comparison of Properties of Borosilicate Glass and Crystalline Ceramic Forms for Immobilization of Savannah River Plant Waste

**AUTHOR:** J.A. Stone; J.S. Allender; T.H. Gould

**PUBLICATION:** USDOE Savannah River Laboratory, DP-1627

**DATE:** April 1, 1982

**ORGANIZATION:** E.I. DuPont, SRL, Aiken, SC

**SUMMARY:** Properties of borosilicate glass are compared to the crystalline ceramic Synroc-D for immobilization of SRP defense high-level waste.

|                            |   |                     |
|----------------------------|---|---------------------|
| Waste Form:                | Glass, borosilicate   |                     |
| Waste Type:                | Simulated high level radioactive defense waste  |                     |
| Waste Loading:             | Nominal 28 wt% (Synroc-D ~50%)  |                     |
| Development Status:        | Early evaluation  |                     |
| Compressive Strength:      | 550 MPa (Synroc-D 280 MPa)  |                     |
| Thermal Stability:         | Both forms may fracture on cooling; little change < 500°C                             |                     |
| Radiation Stability:       | Little change due to self-irradiation for periods of 1 million yrs. (m)               |                     |
| Biological Stability:      | --  |                     |
| Leach Resistance:          | MCC-1, similar to Synroc-D, g/m <sup>2</sup> /day; Cs~1, Sr<10 <sup>-3</sup> , U~0.10 |                     |
| Immersion Stability:       | --  |                     |
| Free Liquids:              | None  |                     |
| Chemical Durability:       | --  |                     |
| Compositional Flexibility: | Change in glass structure due to irradiation is slight                                |                     |
| Gas Generation:            | Little release of helium  |                     |
| RCRA Compliance:           | --  |                     |
| Conclusions:               | B.S. Glass  | Synroc-D            |
|                            | Waste Loading   | 28 wt%      -50 wt% |
|                            | Leach Rate g/m <sup>2</sup> /day  | Cs~1        ~1      |
|                            |   | Sr <.001    ~0.1    |
|                            |   | U -0.1      ~0.0001 |

## GLASS SUMMARIES

NO. 72

TITLE: Chemical Engineering Problems of Radioactive Waste Fixation by Vitrification

AUTHOR: R.F. Taylor

PUBLICATION: Chemical Engineering Science, Vol. 40, No.4, pp. 541-569

DATE: 1985

ORGANIZATION: U.K. Atomic Energy Authority

SUMMARY: A review of the chemical engineering problems faced in the vitrification of high-level radioactive liquid wastes resulting from the reprocessing of nuclear fuel. A general chemical engineering view of waste vitrification.

|                            |  |
|----------------------------|--|
| Waste Form:                | Glass and glass-ceramic  |
| Waste Type:                | High level radioactive liquid waste  |
| Waste Loading:             | --   |
| Development Status:        | --   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | Self-irradiation does not reduce mechanical properties.  |
| Biological Stability:      | --   |
| Leach Resistance:          | < 1 g/m <sup>2</sup> /day  |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | Volatility of Ru and Cs during melting   |
| RCRA Compliance:           | --   |
| Conclusions:               | Detailed discussion of volatility of Ru and Cs problems. Describes in-can melting, pot melting, continuous metallic melters, and joule heated melters. Concluded that vitrification processes have developed to the stage where a route may reasonably be chosen. Discussed the problems which are central to the vitrification process. |

## GLASS SUMMARIES

NO. 73

**TITLE:** The Long-Term Corrosion and Modelling of Two Simulated Belgian Reference High-Level Waste Glasses

**AUTHOR:** P. Van Iseghem; B. Grambow

**PUBLICATION:** Material Research Society Symposium Proceedings, 112, pp. 631-639.

**DATE:** January 1, 1988

**ORGANIZATION:** SCK/CEN Belgium and H-MI, Berlin

**SUMMARY:** Comparison of leach test results to computer modelling for two glass compositions shows effects of SAN to reaccelerate corrosion at long times probably due to formation of secondary phases, such as analcime.

**Waste Form:** Borosilicate glass

**Waste Type:** Reprocessed high level waste

**Waste Loading:** None

**Development Status:** Intermediate

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Long-term rates around  $5 \times 10^{-3}$  and  $3 \times 10^{-3}$  g/m<sup>2</sup>/d prior to reacceleration

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** --

## GLASS SUMMARIES

NO. 74

TITLE: Key Parameters of Glass Dissolution in Integrated Systems

AUTHOR: E.Y. Vernaz; N. Gordon

PUBLICATION: Material Research Society Symposium Proceedings, 212, pp. 19-30

DATE: January 1, 1991

ORGANIZATION: CEA/CEN Valrho, France

SUMMARY: Summarizes test results of RFTF glass under potential French storage conditions

Waste Form: Borosilicate glass

Waste Type: Simulated

Waste Loading: None

Development Status: Advanced

Compressive Strength: --

Thermal Stability: --

Radiation Stability: Glasses doped with Am showed no difference in leachability

Biological Stability: --

Leach Resistance: Long term (3 yr) leach rate=  $2 \times 10^{-3}$  g/m<sup>2</sup>/day.

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Presence of clay leads to higher long-term leach rates, bentonite has no effect on leaching.

## GLASS SUMMARIES

NO. 75

TITLE: Chemical Approach to Glass

AUTHOR: M.B. Volf

PUBLICATION: Book, Glass Science and Technology, 7 Elsevier, New York

DATE: January 1, 1984

ORGANIZATION:

SUMMARY: Discusses the influence of the various elements present in glass. The book is generally limited to silicate glasses, which are most widely used in commercial glassmaking. Borate, phosphate, and germanate glasses are dealt with only briefly.

Waste Form: Glass general

Waste Type: --

Waste Loading: --

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: --

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Used to discuss solubility of chlorides, noble metals, etc., in glass.

## GLASS SUMMARIES

NO. 76

**TITLE:** Materials Characterization Center, Second Workshop on Irradiation Effects In Nuclear Waste Forms, Summary Report

**AUTHOR:** W.J. Weber; R.P. Turcotte

**PUBLICATION:** USDOE, PNL-4121

**DATE:** January 1, 1982

**ORGANIZATION:** Pacific Northwest Laboratory, Battelle

**SUMMARY:** Discusses results of the second workshop on irradiation effects in nuclear waste forms. See Ref. No. 58.

Waste Form: --  
Waste Type: --  
Waste Loading: --  
Development Status: --  
Compressive Strength: --  
Thermal Stability: --  
Radiation Stability: --  
Biological Stability: --  
Leach Resistance: --  
Immersion Stability: --  
Free Liquids: --  
Chemical Durability: --  
Compositional Flexibility: --  
Gas Generation: --  
RCRA Compliance: --  
Conclusions:

Ion or neutron irradiations are not substitutes for the actinide doping technique, as described by the MCC-6. Ion or neutron irradiations may be useful for screening tests or more fundamental studies.



## GLASS SUMMARIES

NO. 77

**TITLE:** Nuclear Waste Glasses

**AUTHOR:** G.G. Wicks

**PUBLICATION:** Treatise on Materials Science and Technology, Vol. 26, Glass IV, pp. 57-118

**DATE:** January 1, 1985

**ORGANIZATION:** Savannah River Laboratory

**SUMMARY:** A review of vitrification and storage of high level glass waste forms.

**Waste Form:** Glass, borosilicate

**Waste Type:** High-level radioactive defense wastes

**Waste Loading:** 5 to 40 wt%

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** Increase in glass surface area was less than 10 times with increased cooling rate. The effect of radiation on glass properties is small.

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Durability improves with increased waste loading

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** Generation of helium in closed canister is about 80 psi in  $10^6$  years.

**RCRA Compliance:** --

**Conclusions:** Describes DWPF, West Valley, French AVM, and ceramic melters. Describes the performance of the waste glass form. Briefly reviews some of the facets of glass leaching. Effects of waste loading, temperature, time, and pH on leaching are considered. Mechanical properties, thermal stability, and radiation effects are considered.

**GLASS SUMMARIES****NO. 78**

**TITLE:** High Level Radioactive Waste - Doing Something About it

**AUTHOR:** G.G. Wicks; D.F. Bickford

**PUBLICATION:** DuPont Savannah River Laboratory DP-1777

**DATE:** March 1, 1989

**ORGANIZATION:** Savannah River Laboratory

**SUMMARY:** A history of the high level waste vitrification program at Savannah River. Includes the involvement of E.I. DuPont Inc.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Glass, borosilicate   |
| <b>Waste Type:</b>                | High level radioactive defense waste                                |
| <b>Waste Loading:</b>             | --  |
| <b>Development Status:</b>        | --  |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | --  |
| <b>Biological Stability:</b>      | --  |
| <b>Leach Resistance:</b>          | --  |
| <b>Immersion Stability:</b>       | --  |
| <b>Free Liquids:</b>              | --  |
| <b>Chemical Durability:</b>       | --  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | Discusses background of Savannah River and development of the DWPF. |

## GLASS SUMMARIES

NO. 79

**TITLE:** Long-Term Leaching Behavior of Simulated Savannah River Plant Waste Glass

**AUTHOR:** G.G. Wicks; J.A. Stone; G.T. Chandler; S. Williams

**PUBLICATION:** Savannah River Lab, DP-1728

**DATE:** August 1, 1986

**ORGANIZATION:** E.I. DuPont, Aiken, SC

**SUMMARY:** Long-term leaching data were obtained for SRP 131 type waste glasses. Experiments were conducted for time periods of up to 4 years using MCC-1 type procedures.

|                            |  |
|----------------------------|--|
| Waste Form:                | Glass, borosilicate  |
| Waste Type:                | Simulated high level radioactive defense waste   |
| Waste Loading:             | 35 wt%   |
| Development Status:        | Intermediate glasses   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | < 1 g/m <sup>2</sup> /day  |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | The data obtained represent a worst case for release of radionuclides from glass matrices. Observed leach rates of less than 1 g/m <sup>2</sup> /d after about 2 to 3 months and lower at longer times. Measured species, such as Sr, after leaching for 3.5 years at 90°C are about 0.006 g/m <sup>2</sup> /d and for sodium are about 0.1 g/m <sup>2</sup> /d. |

## GLASS SUMMARIES

NO. 80

**TITLE:** Analyses of SRS Waste Glass Buried in Granite in Sweden and Salt in the United States

**AUTHOR:** J.P. Williams; G.G. Wicks; D.E. Clark; A.R. Lodding

**PUBLICATION:** Ceramic Transactions, Vol. 23, Nuclear Waste Management IV, G.G. Wicks, et al., eds., pp. 663-674

**DATE:** April 29, 1991

**ORGANIZATION:** American Ceramic Society, Westerville, Ohio

**SUMMARY:** The chemical durability of SRL waste glass compositions buried for up to two years in granite at Stipa and WIPP was excellent and the leach rate decreased with time. These glasses were characterized by the formation of two main surface layers. Quantitative analyses of the glass interaction zone showed that less than one micron of the glass had interacted with the environment after testing at 90°C for 2 years.

**Waste Form:** Glass, borosilicate

**Waste Type:** Simulated, high-level radioactive waste

**Waste Loading:** 28 wt%

**Development Status:** Advanced

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** The chemical durability of SRL waste glass compositions buried for up to two years in granite at Stripa easily passed the leaching requirements and also improved with increasing time.

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** One micron interaction after 2 years in accelerated conditions at 90°C in salt and granite repositories.

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**HYDRAULIC CEMENT SUMMARIES**

## HYDRAULIC CEMENT SUMMARIES INDEX

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|-------|--------------|--|
| NO. 1 | AUTHOR:      | Boehmer, A.M.; Larsen, M.M.  |
|       | TITLE:       | Solidification of Hazardous and Mixed Radioactive Waste at the Idaho National Engineering Laboratory |
|       | PUBLICATION: | Waste Management '86   |
|       | DATE:        | 3/2-6/86   |
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| NO. 2 | AUTHOR:      | Bostik, W. D., et al.   |
|       | TITLE:       | Grout-Based Waste Forms for the Solidification of Anion Exchange Resins |
|       | PUBLICATION: | Ceramic Transactions, Vol. 23   |
|       | DATE:        | 1991  |
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|-------|--------------|--|
| NO. 3 | AUTHOR:      | Cohen, S.; Crouzet, P.   |
|       | TITLE:       | Syncrete (a highly efficient polymer cement embedding matrix for waste processing) |
|       | PUBLICATION: | Waste Management '86   |
|       | DATE:        | 3/2-4/86   |
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| NO. 4 | AUTHOR:      | Cowgill, M.G.  |
|       | TITLE:       | A Comparison of Solidification Media for the Stabilization of Low-Level Radioactive Wastes |
|       | PUBLICATION: | BNL-52304  |
|       | DATE:        | 10/91  |
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|-------|--------------|---|
| NO. 5 | AUTHOR:      | Fischer, D.F.; Johnson, T.R.                |
|       | TITLE:       | Immobilization of IFR Salt Wastes in Mortar |
|       | PUBLICATION: | Spectrum '88                                |
|       | DATE:        | 9/11-15/88                                  |
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| NO. 6 | AUTHOR:      | Fischer, D.F.; et al.   |
|       | TITLE:       | Laboratory Evaluations of Mortars for Immobilizing Waste IFR Salt |
|       | PUBLICATION: | ANL-IFR-120   |
|       | DATE:        | 9/89  |
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| NO. 7 | AUTHOR:      | Funk, H.J.; Pfeiffer, R.  |
|       | TITLE:       | Solidification by Cementation at Liquid Radioactive Primary Waste Mixes |
|       | PUBLICATION: | Waste Management '90  |
|       | DATE:        | 2/25-3/1/90   |
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- NO. 16 AUTHOR: Hoyle, S.; Grutzeck, M.W.  
TITLE: Effects of Phase Composition on the Cesium Leachability  
of Cement-Based Waste Forms  
PUBLICATION:  
DATE: 1986
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- NO. 17 AUTHOR: Jaouen, C.; Vigreux, B.  
TITLE: Cement Solidification of Spent Ion Exchange Resins  
Produced by the Nuclear Industry  
PUBLICATION:  
DATE: 9/88
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- NO. 18 AUTHOR: Kalb, P.D.; Colombo, P.  
TITLE: Full Scale Leaching of Commercial Reactor Waste Forms  
PUBLICATION: BNL 35561  
DATE: 9/84
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- NO. 19 AUTHOR: Kalb, P.D.; Heiser, J.H. III; Colombo, P.  
TITLE: Encapsulation of Mixed Radioactive and Hazardous  
Waste Contaminated Incinerator Ash in Modified Sulfur  
Cement  
PUBLICATION: BNL-45163  
DATE: 1990
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- NO. 20 AUTHOR: Langton, C.A.  
TITLE: Metal Toxicity Evaluation of Savannah River Plant  
Saltstone Comparison of EP and TCLP Test Results  
PUBLICATION: Waste Management '88, Vol. 1  
DATE: 2/28-3/3/88
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- NO. 21 AUTHOR: Langton, C.A.; Dukes, M.D.; Simmons, R.V.  
TITLE: Cement-Based Waste Forms for Disposal of Savannah  
River Plant Low-Level Radioactive Salt Waste  
PUBLICATION: DP-MS-83-71  
DATE: 11/14-17/83
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- NO. 22 AUTHOR: Lerch, R. E.  
TITLE: Division of Waste Management, Production and  
Reprocessing Programs Report for January-June 1977  
PUBLICATION: HEDL-TME-77-74  
DATE: 7/77
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- NO. 23    AUTHOR:            Lerch, R. E.  
          TITLE:            Division of Waste Management, Production and  
                              Reprocessing Programs Report for January-December  
                              1976  
          PUBLICATION:    HDL-TME-77-40  
          DATE:            1977
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- NO. 24    AUTHOR:            McConnell, J.W.; Neilson, R.M.; Rogers, R.D.  
          TITLE:            Testing Waste Forms Containing High Radionuclide  
                              Loadings  
          PUBLICATION:    Waste Management '86  
          DATE:            3/2-6/86
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- NO. 25    AUTHOR:            McIsaac, C.V.; Akers, D.W.; McConnell, J.W.; Morcos,  
                              N.  
          TITLE:            Leach Studies on Cement-Solidified Ion Exchange Resins  
                              From Decontamination Processes at Operating Nuclear  
                              Power Stations  
          PUBLICATION:    EGG-M-92090  
          DATE:
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- NO. 26    AUTHOR:            Rosenstiel, T.L.; Bodett, S.P.; Lange, R.G.  
          TITLE:            Envirostone Gypsum Cement  
          PUBLICATION:    Topical Report  
          DATE:            1984
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- NO. 27    AUTHOR:            Rzycki, B.M.; Suarez, A.A.  
          TITLE:            Setting Temperature Evolution of Nitrate Radwaste  
                              Immobilized in Ordinary Portland Cement  
          PUBLICATION:     
          DATE:            9/88
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- NO. 28    AUTHOR:            Sams, T.L.; McDaniel, E.W.  
          TITLE:            Development of a Cement-Based Grout for  
                              Immobilization of a Low-Level Waste Stream Containing  
                              Sodium Sulfate  
          PUBLICATION:    Waste Management '88, Vol. 1  
          DATE:            2/28-3/3/88
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- NO. 29 AUTHOR: Sauda, K.; Todo, F.; Nakashima, T.; Kagawa, T.;  
Kuribayashi, H.  
TITLE: Advanced Cement-Solidification Process for Spent Ion-  
Exchange Resins  
PUBLICATION: Waste Management '90  
DATE: 2/25-3/1/90
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- NO. 30 AUTHOR: Schuler, T.F.; Charlesworth, D.L.  
TITLE: Solidification of Radioactive Incinerator Ash  
PUBLICATION: Waste Management '86  
DATE: 3/2-6/86
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- NO. 31 AUTHOR: Siskind, B.; Adams, J.W.; Clinton, J.H.; Piciulo, P.L.  
TITLE: The Effect of Cure Conditions on the Stability of Cement  
Waste Forms After Immersion in Water  
PUBLICATION: Waste Management '88  
DATE: 2/25-3/190
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- NO. 32 AUTHOR: Stone, J.A.; d'Entremont, P.D.  
TITLE: Measurement and Control of Cement Set Times in Waste  
Solidification  
PUBLICATION: DP-1404  
DATE: 9/76
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- NO. 33 AUTHOR: Tallent, O.K.; McDaniel, E.W.; Delcul, G.D.; Dodson,  
K.E.; Trotter, D.R.  
TITLE: Immobilization of Technetium and Nitrate in Cement-  
Based Materials  
PUBLICATION: Materials Research Society Symposium Proceedings, Vol.  
112, 1988  
DATE: 1988
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- NO. 34 AUTHOR: Taouen, C.; Vigreux, B.  
TITLE: Cement Solidification of Spent Ion Exchange Resins  
Produced by the Nuclear Industry  
PUBLICATION: Spectrum '88  
DATE: 9/11-15/88
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- NO. 35 AUTHOR: Unger, S.L.; Telles, R.W.  
TITLE: Surface Encapsulation Process for Managing Low-Level  
Radioactive Wastes  
PUBLICATION:  
DATE: 1986
-



- NO. 43    AUTHOR:            Cooley, C.R.; et al.  
          TITLE:            Alternatives for Managing Wastes from Reactors and  
                                 Post-Fission Operations in the LWR Fuel Cycle  
          PUBLICATION:    ERDA-76-42, Vol. 2 of 5  
          DATE:            May 1976
- 
- NO. 44    AUTHOR:            Place, B.G.  
          TITLE:            Treatment Technology for Transuranic Waste Streams--  
                                 Cementation, Vitrification, and Incineration Testing for  
                                 the treatment of Spent Ion Exchange Media  
          PUBLICATION:    WHC-EP-0462  
          DATE:            April 1992
- 
- NO. 45    AUTHOR:            Ganser, B.L.  
          TITLE:            Cementation--A Solution for Final Disposal?  
          PUBLICATION:    Waste Management '90  
          DATE:            March 1990
- 
- NO. 46    AUTHOR:            McConnell, Jr., J. W.; Rogers, R.D.  
          TITLE:            Results of Field Testing of Waste Forms Using Lysimeters  
          PUBLICATION:    Waste Management '90  
          DATE:            February 1990
- 
- NO. 47    AUTHOR:            Compiled by Lerch, R.E.  
          TITLE:            Division of Waste Management, Production and  
                                 Reprocessing Programs Report for January-June 1977  
          PUBLICATION:    HEDL-TME 77-74  
          DATE:            July 1977
-

**HYDRAULIC CEMENT SUMMARIES****NO. 1**

**TITLE:** Solidification of Hazardous and Mixed Radioactive Waste at the Idaho National Engineering Laboratory

**AUTHOR:** A. M. Boehmer; M. M. Larsen

**PUBLICATION:** Waste Management '86

**DATE:** March 2-6, 1986

**ORGANIZATION:** Idaho National Engineering Laboratory (INEL)

**SUMMARY:** Testing shows toxic waste materials can be solidified using cement, cement-silicate, or Envirostone binders.

**Waste Form:** Cement, cement-silicate, Envirostone  
**Waste Type:** Mixed waste  
**Waste Loading:** Up to 12,500 ppm potassium chromate can be immobilized with cement silicate

**Development Status:** --  
**Compressive Strength:** --  
**Thermal Stability:** --  
**Radiation Stability:** --  
**Biological Stability:** --  
**Leach Resistance:** --  
**Immersion Stability:** --  
**Free Liquids:** --  
**Chemical Durability:** --  
**Compositional Flexibility:** --  
**Gas Generation:** --

**RCRA Compliance:** Did testing to show toxic characteristics were done away with

**Conclusions:** Has listed treatment for 54 types of mixed wastes in table form.

## HYDRAULIC CEMENT SUMMARIES

NO. 2

**TITLE:** Grout-Based Waste Forms for the Solidification of Anion Exchange Resins

**AUTHOR:** W.D. Bostick; et al.

**PUBLICATION:** Ceramic Transactions, Vol. 23

**DATE:** 1991

**ORGANIZATION:** Oak Ridge National Laboratory

**SUMMARY:** Results on the encapsulation of beaded anion exchange resins in grout formulations are presented.

**Waste Form:** Type I-II portland cement, slag, and additives

**Waste Type:** Dowex SRB-OH resin beads in nitrate form containing  $^{99}\text{Tc}$

**Waste Loading:** 40 vol% of waste form

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Leachability index for  $^{99}\text{Tc}$  was 14 and for nitrate ion was >9.5

**Immersion Stability:** All samples passed

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** The best binder composition was 75% blast furnace slag, 10 wt% Type I-II cement and 15 wt% attapulgite clay. This formula had a high leachability index for nitrate and pertechnetate ions, the lowest weight loss during wetting/drying cycle testing and an acceptable freezing/thawing behavior.

## HYDRAULIC CEMENT SUMMARIES

NO. 3

**TITLE:** Syncrete (A Highly Efficient Polymer Cement Embedding Matrix for Waste Processing)

**AUTHOR:** S. Cohen; P. Crouzet

**PUBLICATION:** Waste Management '86

**DATE:** March 2-4, 1986

**ORGANIZATION:** STMI (France)

**SUMMARY:** Report describes characteristics and test results of Syncrete, a polymer-cement concrete.

**Waste Form:** Hydraulic concrete plus a thermosetting polymer mixture

**Waste Type:** --

**Waste Loading:** 60% for resins, 30 to 35% for sludges and concentrates

**Development Status:** --

**Compressive Strength:** >90 mPa

**Thermal Stability:** 12,000 frost/thaw cycles over 3 years resulted in weight loss of only 0.3 %.

**Radiation Stability:** No apparent damage from  $10^8$  rad exposures.

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** Stable during water immersion to 142 days

**Free Liquids:** --

**Chemical Durability:** Resists attack from 10 to 20 % acid solutions for up to a year. Also, it is readily applicable to salts such as sea salts.

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Syncrete combines the positive characteristics of hydraulic cement with the sealing characteristics of thermosetting polymer material.

**HYDRAULIC CEMENT SUMMARIES****NO. 4**

**TITLE:** A Comparison of Solidification Media for the Stabilization of Low-Level Radioactive Wastes

**AUTHOR:** M.G. Cowgill

**PUBLICATION:** BNL 52304

**DATE:** October 1991

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** This report gives comparative performance characteristics for various waste forms.

|                                   |          |
|-----------------------------------|----------|
| <b>Waste Form:</b>                | Variable |
| <b>Waste Type:</b>                | --       |
| <b>Waste Loading:</b>             | --       |
| <b>Development Status:</b>        | --       |
| <b>Compressive Strength:</b>      | --       |
| <b>Thermal Stability:</b>         | --       |
| <b>Radiation Stability:</b>       | --       |
| <b>Biological Stability:</b>      | --       |
| <b>Leach Resistance:</b>          | --       |
| <b>Immersion Stability:</b>       | --       |
| <b>Free Liquids:</b>              | --       |
| <b>Chemical Durability:</b>       | --       |
| <b>Compositional Flexibility:</b> | --       |
| <b>Gas Generation:</b>            | --       |
| <b>RCRA Compliance:</b>           | --       |
| <b>Conclusions:</b>               | --       |



## HYDRAULIC CEMENT SUMMARIES

NO. 5

**TITLE:** Immobilization of IFR Salt Wastes in Mortar

**AUTHOR:** D. F. Fischer; T. R. Johnson

**PUBLICATION:** Spectrum '88

**DATE:** September 11-15, 1988

**ORGANIZATION:** Argonne National Laboratory

**SUMMARY:** Both flyash and blast furnace slag were found to be beneficial mortar additives for immobilizing IFR waste salt.

**Waste Form:** Portland cement-base mortars

**Waste Type:** Alkaline earth chlorides

**Waste Loading:** 10%

**Development Status:** --

**Compressive Strength:** Required  $\geq 7$  mPa within 3 days; average 10 to 20 mPa after 3 days and 50 to 70 mPa after 56 days

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Leachability indexes for chloride -7 to 8.

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** --

## HYDRAULIC CEMENT SUMMARIES

NO. 6

TITLE: Laboratory Evaluations of Mortars for Immobilizing Waste IFR  
Salt

AUTHOR: D. F. Fischer; et al.

PUBLICATION: ANL-IFR-120

DATE: September 1989

ORGANIZATION: Argonne National Laboratory

## SUMMARY:

Waste Form: 25% Type I portland cement, 25% Class F flyash and 50%  
blast furnace slag  
Waste Type: Chloride salt form IFR  
Waste Loading: 3-10% salt  
Development Status: More work needed for qualification  
Compressive Strength: 7 day 10 mPa (1450 psig)  
Thermal Stability: --  
Radiation Stability: --  
Biological Stability: --  
Leach Resistance: Leachability index greater than 7 for chloride ion  
Immersion Stability: --  
Free Liquids: None  
Chemical Durability: --  
Compositional Flexibility: Yes  
Gas Generation: --  
RCRA Compliance: --  
Conclusions:  $\text{CaCl}_2$  containing salt waste produced a fluid mix that set  
extremely fast. Flyash addition increased set times  
significantly with more than 10% salt, the specimens swelled  
and cracked, but with about 4%, the mortar was reasonably  
strong and did not swell or crack. Several binder mixes were  
evaluated and reported.

**HYDRAULIC CEMENT SUMMARIES**

NO. 7

**TITLE:** Solidification by Cementation at Liquid Radioactive Primary Waste Mixes

**AUTHOR:** H. J. Funk; R. Pfeiffer

**PUBLICATION:** Waste Management '90

**DATE:** February 25-March 1, 1990

**ORGANIZATION:** Kernforschungszentrum Karlsruhe

**SUMMARY:** Shows successful cementation of evaporator concentrates containing primarily sodium nitrate.

**Waste Form:** Cement

**Waste Type:** Liquid waste evaporator concentrates

**Waste Loading:** --

**Development Status:** In use

**Compressive Strength:**  $\geq 10 \text{ N/mm}^2$

**Thermal Stability:** --

**Radiation Stability:**  $3.7 \text{ Bg/cm}^2$

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** No free liquid detected

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** --

## HYDRAULIC CEMENT SUMMARIES

NO. 8

TITLE: High Strength Cementized Dried Resins

AUTHOR: R. L. Gay; L. F. Grantham

PUBLICATION: Waste Management '88

DATE: 1988

ORGANIZATION: Rockwell International Corporation

SUMMARY: Depleted ion exchange resins dried in a high-efficiency dryer and solidified in cement are nearly impermeable to water and will not reabsorb or swell when immersed.

Waste Form: Cement

Waste Type: Depleted ion-exchange resin

Waste Loading: 28 wt% dried resin = 70 wt% of original dewatered resins

Development Status: Mature

Compressive Strength: 20 samples passed 0.48 mPa (69 psi) test

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: --

Immersion Stability: 20 samples passed 90-day immersion test, 2 samples failed

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Fresh resins did not produce high-strength cement samples, but chemically spent resin waste forms passed compressive strength and immersion tests. Cost saving figures are presented for both dewatered and dried resin waste forms, based on volume reduction.

## HYDRAULIC CEMENT SUMMARIES

NO. 9

**TITLE:** Performance Testing of Blast-Furnace Slag for Immobilization of Technetium in Grout

**AUTHOR:** T. M. Gilliam; R. D. Spence; B. S. Evans-Brown; I. L. Morgan; S. L. Shoemaker; W. D. Bostick

**PUBLICATION:** Nuclear and Hazardous Waste Management, Spectrum '88

**DATE:** September 1988

**ORGANIZATION:** Martin Marietta Energy Systems

**SUMMARY:** Grout formulas have been developed that can satisfactorily sequester  $^{99}\text{Tc}$  in low level mixed waste.

**Waste Form:** Portland cement/flyash/blast furnace slag

**Waste Type:** Heavy metals sludge and filtrate

**Waste Loading:** 38% weight

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Leachability index for  $^{99}\text{Tc}$  was 10.5 and  $\text{NO}_3$  was 7.3

**Immersion Stability:** --

**Free Liquids:** None

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Leachability indices of  $^{99}\text{Tc}$  and  $\text{NO}_3$  increased by 4.4 units (10.5 versus .1) and 1.4 units (7.3 versus 5.9), respectively, by the addition of slag in waste forms. The principal mechanisms of improved retention are a result of the redox potential of the slag and not the resulting improved physical properties of the grout.

**HYDRAULIC CEMENT SUMMARIES**

NO. 10

**TITLE:** Disposal Concepts for Waste in Underground Single-Shell Storage Tanks at the Hanford Site

**AUTHOR:** W. O. Greenhalgh

**PUBLICATION:** Waste Management '92

**DATE:** March 1-5, 1992

**ORGANIZATION:** Westinghouse Hanford Company

**SUMMARY:** Disposal concepts and waste form options for use in immobilizing liquid tank waste were discussed.

|                            |   |
|----------------------------|---|
| Waste Form:                | Variable  |
| Waste Type:                | Single-shell tank waste liquid  |
| Waste Loading:             | Variable  |
| Development Status:        | Laboratory screening of waste form candidates   |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | The goal of this program was to develop or establish disposal technology that could be used to remediate single-shell tank waste liquid. Disposal options to be tested were outlined. |

## HYDRAULIC CEMENT SUMMARIES

NO. 11

**TITLE:** The Immobilization of Organic Liquid Wastes

**AUTHOR:** W. O. Greenhalgh

**PUBLICATION:** Waste Management '86

**DATE:** March 2-6, 1986

**ORGANIZATION:** Westinghouse Hanford Company

**SUMMARY:** The report describes the cement immobilization of low-level radioactive organic liquid waste to form a non-combustible monolith solid.

**Waste Form:** Cement

**Waste Type:** Organic liquid waste such as pump oil, mineral spirits, and TBP-NPH

**Waste Loading:** Variable from 32 to 40 vol%

**Development Status:** Applied to actual waste materials

**Compressive Strength:** Variable - a 35 vol% TBP-Dodecane product exhibited a 28-day compressive strength of 250 psi.

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** Flame resistant solid

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Emulsifiers and cement accelerator additives were used to solidify organic liquids up to 45% by volume. Cement products containing greater than 40 vol% might not exhibit sufficient strength to meet the NRC minimum of 60 psi without use of a good set accelerator.

**HYDRAULIC CEMENT SUMMARIES****NO. 12**

**TITLE:** High-Impact Concrete for Fill in U.S. Department of Transportation Type Shipping Containers

**AUTHOR:** W. O. Greenhalgh; R. J. Cash

**PUBLICATION:** Waste Management '90, Tucson, AZ

**DATE:** February 25 - March 1, 1990

**ORGANIZATION:** Westinghouse Hanford Company

**SUMMARY:** The report describes a high-impact concrete product which is not brittle, but will absorb high energy impacts with little or no loss of product integrity.

**Waste Form:** Cement/concrete

**Waste Type:** --

**Waste Loading:** --

**Development Status:** --

**Compressive Strength:** This concrete does not exhibit any appreciable compressive strength but rather absorbs the compression force. The product, without containment, can be dropped 30 ft with minimal or no loss of material.

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** None

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** The high impact concrete was designed for use in filling empty tanks, vessels, etc., requiring shipment over public highways with interior parts that have radioactive contamination. The fill concrete eliminates or nearly eliminates the potential loss of containment due to normal impact accidents (traffic accidents).



**HYDRAULIC CEMENT SUMMARIES****NO. 13**

**TITLE:** Product Characteristics of TRU Waste Immobilized in Grout

**AUTHOR:** W. O. Greenhalgh; R. J. Cash; M. A. Christie

**PUBLICATION:** Waste Management '88

**DATE:** February 28-March 3, 1988

**ORGANIZATION:** Westinghouse Hanford Company

**SUMMARY:** Laboratory and drum site specimens of shred/grout (shredded waste solidified with a cement grout) will be paired and tested against NRC waste form positions and other handling and transportation type tests.

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Cement/grout monolith  |
| <b>Waste Type:</b>                | Simulated TRU combustible waste  |
| <b>Waste Loading:</b>             | Near 100%  |
| <b>Development Status:</b>        | Demonstration with simulated waste   |
| <b>Compressive Strength:</b>      | 510 psi  |
| <b>Thermal Stability:</b>         | 480 psi - no apparent damage   |
| <b>Radiation Stability:</b>       | 210/120 psi - no weight loss, no apparent damage   |
| <b>Biological Stability:</b>      | 420, 560, 690, 650, and 1010 psi - no damage   |
| <b>Leach Resistance:</b>          | Index >10  |
| <b>Immersion Stability:</b>       | >80 psi  |
| <b>Free Liquids:</b>              | None detected  |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | The shred/grout waste form produced met all DOE and NRC test requirements and exhibited the following characteristics: Free standing monolith; no free liquid; resistant to radiation; resistant to biodegradation; resistant to water leaching; stable in water; resistant to climate variations; resistant to typical fire situations; stable during cross-country transport; resistant to forklift handling punctures; stable to substantial impact-type accidents. |

## HYDRAULIC CEMENT SUMMARIES

NO. 14

**TITLE:** Detection of Free Liquid in Sealed Containers Simulating Drums of Solidified Radioactive Liquid Waste

**AUTHOR:** W. O. Greenhalgh; C. R. Green

**PUBLICATION:** Nondestructive Evaluation in the Nuclear Industry, Salt Lake City, Utah, American Society for Metals (ASM) publication

**DATE:** February 11-13, 1980

**ORGANIZATION:** Hanford Engineering Development Laboratory

**SUMMARY:** The report describes two successfully demonstrated ways of examining drums by NDT thermal methods for water (free liquid).

**Waste Form:** Cement type, urea formaldehyde, and bitumen (asphalt)

**Waste Type:** Simulated sodium sulfate radwaste

**Waste Loading:** Variable

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Infrared scanning camera and thermal imaging methods were shown to detect "free liquid" in sealed drums of simulated, solidified liquid waste. The volume of free liquid can also be estimated by the methods.

## HYDRAULIC CEMENT SUMMARIES

NO. 15

TITLE: Advanced Cementation Concepts Final Report

AUTHOR: C. G. Howard

PUBLICATION: AEEW-R2398

DATE: October 1989

ORGANIZATION: Winfrith Technology Center

SUMMARY: Improvements to existing cement formulations were sought using inorganic additives, such as microsilica and limestone flour.

Waste Form: Ordinary portland cement/blast furnace slag/pulverized fuel ash

Waste Type: --

Waste Loading: 45% to 52%

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: The addition of microsilica reduced  $^{137}\text{Cs}$  leachability from 45% to 145 days to 16%

Immersion Stability: --

Free Liquids: Samples containing 6% microsilica have no measurable bleed after 24 hours

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Microsilica and limestone flour block pore spaces in cement forms and reduced the total amount of  $^{137}\text{Cs}$  leached out. These additives caused no detrimental effects on the dimensional stability, compressive strength, or elastic modules of the waste forms.

## HYDRAULIC CEMENT SUMMARIES

NO. 16

**TITLE:** Effects of Phase Composition on the Cesium Leachability of Cement-Based Waste Forms

**AUTHOR:** S. Hoyle; M. W. Grutzeck

**PUBLICATION:** Waste Management '86

**DATE:** 1986

**ORGANIZATION:** Pennsylvania State University

**SUMMARY:** Phase relations in the system  $\text{CaO-Al}_2\text{O}_3\text{-SiO}_2\text{-H}_2\text{O}$  were investigated to enhance Cs retention by enrichment of the bulk composition of the mixture in silica and alumina.

**Waste Form:** Type I portland cement/calcium aluminate cement/condensed silica fume

**Waste Type:** Deionized water containing added cesium hydroxide

**Waste Loading:** 30% to 60%

**Development Status:** --

**Compressive Strength:** 15 mPa (2100 psi) to 73 mPa (10,500 psi)

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Improvement in Cs leachability is achieved by increasing silica or alumina content. A high calcium content is correlated with high Cs leachability.

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** Yes

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Five mixtures having varying compositions were formulated from portland cement, calcium aluminate cement, and condensed silica fume. Leaching results indicated that the enrichment of the bulk composition of a mixture in silica and for alumina-enhanced cesium retention. All samples were found to be as strong, if not stronger, than their non-cesium containing counterparts.

## HYDRAULIC CEMENT SUMMARIES

NO. 17

**TITLE:** Cement Solidification of Spent Ion Exchange Resins Produced by the Nuclear Industry

**AUTHOR:** C. Jaouen; B. Vigreux

**PUBLICATION:** Spectrum '88

**DATE:** September 1988

**ORGANIZATION:** SGN (France)

**SUMMARY:** A simple rapid pretreatment.

**Waste Form:** Cement  
**Waste Type:** Spent ion exchange resin  
**Waste Loading:** 40 to 75 vol%, 27 wt% dry residue  
**Development Status:** Mature  
**Compressive Strength:** 25 wt% resin was 9 mPa after 90 days; 10 wt% resin was 31 mPa after 90 days  
**Thermal Stability:** --  
**Radiation Stability:** --  
**Biological Stability:** --  
**Leach Resistance:** --  
**Immersion Stability:** --  
**Free Liquids:** None  
**Chemical Durability:** --  
**Compositional Flexibility:** --  
**Gas Generation:** --  
**RCRA Compliance:** --  
**Conclusions:** A mobile plant has been designed utilizing a highly efficient batch mixer for mixing PWR secondary system resins and for cement solidification of ash. Pretreatment of waste can be adapted to various cases. Resins are saturated (i.e., the water content is limited), thereby reducing the water/cement/resin interactions.

## HYDRAULIC CEMENT SUMMARIES

NO. 18

TITLE: Full Scale Leaching of Commercial Reactor Waste Forms

AUTHOR: P. D. Kalb; P. Colombo

PUBLICATION: BNL 35561

DATE: September 1984

ORGANIZATION: Brookhaven National Laboratory

SUMMARY: Full-scale leaching experiment using 55-gal drum size waste forms from commercial reactor stations was accomplished

Waste Form: Masonry cement and Type III cement

Waste Type: Sodium sulfate and boric acid concentrated ion exchange resins

Waste Loading: --

Development Status: Full-scale

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: --

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Trends observed in the leaching of laboratory-scale simulated waste forms have been confirmed for full-size commercial reactor waste forms. Leachability of  $^{60}\text{Co}$  from cement was two orders of magnitude lower than Cs isotopes. Elevated concentrations of sodium sulfate in combination with the presence of ion exchange resins can cause premature deterioration in cement waste forms.

**HYDRAULIC CEMENT SUMMARIES****NO. 19**

**TITLE:** Encapsulation of Mixed Radioactive and Hazardous Waste Contaminated Incinerator Ash in Modified Sulfur Cement

**AUTHOR:** P. D. Kalb; J. H. Heiser III; P. Colombo

**PUBLICATION:** BNL-45163

**DATE:** 1990

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Modified sulfur cement provides improved waste loadings and waste form performance over hydraulic cement.

**Waste Form:** Sulfur cement versus hydraulic cement.

**Waste Type:** LLW and mixed waste

**Waste Loading:** Sulfur cement > hydraulic cement: 6 times more sodium sulfate, 2.6 times more boric acid, 1.4 times more bottom ash, and 2.7 times more flyash

**Development Status:** --

**Compressive Strength:** Minimum compressive strength is 1800 psi for sulfur cement containing sodium flyash, sulfate, or ash wastes

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Leach Index for Co<sup>60</sup> 10.7 or higher for sulfur cement and sodium sulfate or ash. Leach Index for Cs<sup>137</sup> 9.7 or higher for sulfur cement and sodium sulfate or ash.

**Immersion Stability:** --

**Free Liquids:** --

**Chemical durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** --

## HYDRAULIC CEMENT SUMMARIES

NO. 20

TITLE: Metal Toxicity Evaluation of Savannah River Plant Saltstone  
Comparison of EP and TCLP Test Results

AUTHOR: C. A. Langton

PUBLICATION: Waste Management '88, Vol. 1

DATE: February 28-March 3, 1988

ORGANIZATION: Savannah River Laboratory

## SUMMARY:

|                            |  |
|----------------------------|--|
| Waste Form:                | Saltstone grout  |
| Waste Type:                | Sodium salt solution with pH >12.5 and Cr <sup>+6</sup> >100 ppm |
| Waste Loading:             | --   |
| Development Status:        | --   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | --   |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | --   |



## HYDRAULIC CEMENT SUMMARIES

NO. 21

**TITLE:** Cement-Based Waste Forms for Disposal of SRP Low-Level Salt Waste

**AUTHOR:** C. A. Langton; et al.

**PUBLICATION:** DP-MS-83-71

**DATE:** November 1983

**ORGANIZATION:** DuPont Company

**SUMMARY:** A cement-based waste form, "saltstone," has been developed for SRP mixed waste containing 32 wt% sodium salts.

|                            |   |
|----------------------------|---|
| Waste Form:                | Portland cement/flyash  |
| Waste Type:                | Salt solution containing 32 wt% sodium salts  |
| Waste Loading:             | 38%   |
| Development Status:        | Mature  |
| Compressive Strength:      | 3 mPa (450 psi) minimum   |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | $10^{-5}$ g/cm <sup>2</sup> -day  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Physical properties, mineralogy and microstructure of the saltstone are similar to conventional grouts. The saltstone product, in combination with design of the total landfill, meets all state and federal guidelines for contaminant release into the environment. |

**HYDRAULIC CEMENT SUMMARIES**

NO. 22

**TITLE:** Division of Waste Management, Production and Reprocessing Programs Report for January-June 1977

**AUTHOR:** Compiled by R.E. Lerch

**PUBLICATION:** HEDL-TME-77-74

**DATE:** July 1977

**ORGANIZATION:** Hanford Engineering Development Laboratory

**SUMMARY:** This report addresses immobilization of boric acid waste and provides composition phase diagrams for boric acid and ion-exchange resin. It also presents thermal stability data for asphalt and urea-formaldehyde organic binders.

**Waste Form:** Variable

**Waste Type:** Low level waste and radwaste

**Waste Loading:** Variable

**Development Status:** --

**Compressive Strength:** Data given for boric acid and ion-exchange waste forms; compression strengths up to 70 kg/cm<sup>2</sup> were reported.

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** A variety of leach rates are tabulated, cesium leach rates varied from 0.5 to 0.01, and strontium from 0.04 to 0.00090 g/cm<sup>2</sup> day.

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Three phase diagrams of cement/water/waste were determined for boric acid. Problems of cementing resins are discussed, some lower waste loading limits are stable. Thermal profiles on asphalt and urea-formaldehyde waste products are given. Leach rate data are provided for cesium and strontium tracers in test waste forms.

**HYDRAULIC CEMENT SUMMARIES****NO. 23**

**TITLE:** Division of Waste Management, Production and Reprocessing Programs Report for January-December 1976

**AUTHOR:** Compiled by R.E. Lerch

**PUBLICATION:** HEDL-TME-77-40

**DATE:** April 1977

**ORGANIZATION:** Hanford Engineering Development Laboratory

**SUMMARY:**

**Waste Form:** Cement Type I-II  
**Waste Type:** Variable-calcium sulfate, neutralized ferric sulfate, sodium sulfate, sodium nitrate, sodium chloride, ion-exchange resin (anion and cation)

**Waste Loading:** --  
**Development Status:** --  
**Compressive Strength:** Compressive strengths are tabulated, strengths up to 1700 psi are tested

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Three phase diagrams of cement/water/waste were determined for the simulated wastes listed above. Tables of product density and compression strengths are given.

**HYDRAULIC CEMENT SUMMARIES****NO. 24**

**TITLE:** Testing Waste Forms Containing High Radionuclide Loadings

**AUTHOR:** J. W. McConnell, Jr.; R. M. Neilson; R. D. Rogers

**PUBLICATION:** Waste Management '86

**DATE:** March 2-6, 1986

**ORGANIZATION:** Idaho National Engineering Laboratory

**SUMMARY:** The report addresses 10 CFR 61 testing of TMI type resin waste.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Portland cement and vinyl ester-styrene (VES) |
| <b>Waste Type:</b>                | TMI spent resin                               |
| <b>Waste Loading:</b>             | --  |
| <b>Development Status:</b>        | --  |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | --  |
| <b>Biological Stability:</b>      | Cement: no; VES: bacteria-no, fungi-yes       |
| <b>Leach Resistance:</b>          | --  |
| <b>Immersion Stability:</b>       | --  |
| <b>Free Liquids:</b>              | --  |
| <b>Chemical Durability:</b>       | --  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | --  |

## HYDRAULIC CEMENT SUMMARIES

NO. 25

**TITLE:** Leach Studies on Cement-Solidified Ion-Exchange Resins from Decontamination Processes at Operating Nuclear Power Stations

**AUTHOR:** C. V. McIsaac; et al.

**PUBLICATION:** EGG-M-9290

**DATE:**

**ORGANIZATION:** Idaho National Engineering Laboratory (INEL)

**SUMMARY:** The effects of varying pH and leachant compositions on leachability were determined for cement-solidified decontamination ion-exchange resin wastes.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Cement  |
| <b>Waste Type:</b>                | Decontamination ion-exchange resins   |
| <b>Waste Loading:</b>             | 30%   |
| <b>Development Status:</b>        | Needs further evaluation to improve strength when immersed in water   |
| <b>Compressive Strength:</b>      | $3 \times 10^3$ kPa (440 psig)  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | --  |
| <b>Biological Stability:</b>      | --  |
| <b>Leach Resistance:</b>          | Leachability index for all test samples were $>8$ for $\text{Co}^{60}$ , $\text{Ni}^{63}$ , and $\text{Fe}^{55}$  |
| <b>Immersion Stability:</b>       | Physical disintegration during leach testing  |
| <b>Free Liquids:</b>              | None  |
| <b>Chemical Durability:</b>       | Initial leachant pH does not affect leachate pH   |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | Loss of waste form stability after exposure to leachants due to swelling of solidified resin beads. However, cumulative releases of radionuclides and chelating agents for forms that disintegrated were similar to those for forms that maintained their general physical integrity. |

## HYDRAULIC CEMENT SUMMARIES

NO. 26

TITLE: Envirostone Gypsum Cement

AUTHOR: T. L. Rosenstiel; S. P. Bodett; R. G. Lange

PUBLICATION: Topical Report

DATE: 1984

ORGANIZATION: United State Gypsum

SUMMARY: Summary of the qualification of Envirostone gypsum cement to 10 CFR 61 type tests

Waste Form: --

Waste Type: --

Waste Loading: --

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: --

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Meets NRC regulatory requirements in 10 CFR 61 for the following wastes: 24% boric acid solution; mixed bed ion exchange bead resin (unexpended); powdered ion exchange resin (unexpended); lubricating oil; mixtures of oil and 24% boric acid solution; mixtures of mixed bed ion exchange bead resin and 24% boric acid solution; mixtures of powdered ion exchange resin and 24% boric acid solution; EDTA decontamination fluid; a neat aqueous mixture of ENVIROSTONE® gypsum cement to simulate encapsulation of solid objects such as used equipment, spent filter cartridges, etc.

**HYDRAULIC CEMENT SUMMARIES**

NO. 27

**TITLE:** Setting Temperature Evolution of Nitrate Radwaste Immobilized  
in Ordinary Portland Cement

**AUTHOR:** B.M. Rzycki; A.A. Suarez

**PUBLICATION:**

**DATE:** September 1988

**ORGANIZATION:** IPEN-CNEN/SP

**SUMMARY:**

|                                   |                 |
|-----------------------------------|-----------------|
| <b>Waste Form:</b>                | Portland cement |
| <b>Waste Type:</b>                | --              |
| <b>Waste Loading:</b>             | --              |
| <b>Development Status:</b>        | --              |
| <b>Compressive Strength:</b>      | --              |
| <b>Thermal Stability:</b>         | --              |
| <b>Radiation Stability:</b>       | --              |
| <b>Biological Stability:</b>      | --              |
| <b>Leach Resistance:</b>          | --              |
| <b>Immersion Stability:</b>       | --              |
| <b>Free Liquids:</b>              | --              |
| <b>Chemical Durability:</b>       | --              |
| <b>Compositional Flexibility:</b> | --              |
| <b>Gas Generation:</b>            | --              |
| <b>RCRA Compliance:</b>           | --              |
| <b>Conclusions:</b>               | --              |

## HYDRAULIC CEMENT SUMMARIES

NO. 28

**TITLE:** Development of a Cement-Based Grout for Immobilization of a Low-Level Waste Stream Containing Sodium Sulfate

**AUTHOR:** T. L. Sams; E. W. McDaniel

**PUBLICATION:** Waste Management '88, Vol. 1

**DATE:** February 1988

**ORGANIZATION:** Oak Ridge National Laboratory

**SUMMARY:** A unique approach to grout formulation using statistical methods has been presented using a low level waste stream containing sodium sulfate

|                            |  |
|----------------------------|--|
| Waste Form:                | Cement-based grout   |
| Waste Type:                | --   |
| Waste Loading:             | --   |
| Development Status:        | --   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | --   |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | This formulation development work encompassed several variables. the reference grout formula developed is cement 46.0%, flyash 32.8%, attapugite clay 1.4%, IRPC 0.8%. This should produce grouts that are processible and durable |



## HYDRAULIC CEMENT SUMMARIES

NO. 29

**TITLE:** Advanced Cement-Solidification Process for Spent Ion-Exchange Resins

**AUTHOR:** K. Sauda; et al.

**PUBLICATION:** Waste Management '90, Vol. 2

**DATE:** February 1990

**ORGANIZATION:** JGC Corporation

**SUMMARY:** A new pretreatment technique for spent ion exchange resins solidified in cement has been reported.

**Waste Form:** Portland cement

**Waste Type:** Ion exchange resins

**Waste Loading:** Resins/water/cement = 18/36/46

**Development Status:** Pilot plant demonstration

**Compressive Strength:** Passed, 1400 to 3000 psi

**Thermal Stability:** Passed, after 30 cycles compressive strength was 1200-2000 psi

**Radiation Stability:** Passed, after  $10^8$  rad compressive strength was 1080 psi

**Biological Stability:** Passed, no growth and compressive strength was 1500 psi

**Leach Resistance:** Leachability index was greater than 7

**Immersion Stability:** Passed, compressive strength after 90 days ranged from 330 to 2650 psi

**Free Liquids:** None

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** None

**RCRA Compliance:** --

**Conclusions:** Dewatered spent ion-exchange resins are agitated at high speed in a highly alkaline cement slurry. The pretreatment parameters have been optimized and 10 CFR 61 tests performed. The products met the waste form criteria.

## HYDRAULIC CEMENT SUMMARIES

NO. 30

TITLE: Solidification of Radioactive Incinerator Ash

AUTHOR: T. F. Schuler; D. L. Charlesworth

PUBLICATION: Waste Management '86

DATE: March 2-6, 1986

ORGANIZATION: Savannah River Laboratory

SUMMARY: An ash cementing process termed "Ashcrete" was set up, the waste product tested, and results reported in this paper.

Waste Form: Cement

Waste Type: Incinerator ash

Waste Loading: --

Development Status: --

Compressive Strength: A minimum of 100 psi criteria was established,  $\geq 1000$  psi was achieved.

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: --

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: --

## HYDRAULIC CEMENT SUMMARIES

NO. 31

**TITLE:** The Effect of Cure Conditions on the Stability of Cement Waste Forms After Immersion in Water

**AUTHOR:** B. Siskind; J. W. Adams; J. H. Clinton; P. L. Piciulo

**PUBLICATION:** Waste Management '88

**DATE:** February 25-March 1, 1990

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Cement-solidified ion exchange resin exhibits a non-portland cement-like behavior if the waste loading is too high.

|                                   |                                    |
|-----------------------------------|------------------------------------|
| <b>Waste Form:</b>                | Cement                             |
| <b>Waste Type:</b>                | Spent mixed-bed ion exchange resin |
| <b>Waste Loading:</b>             | --                                 |
| <b>Development Status:</b>        | --                                 |
| <b>Compressive Strength:</b>      | --                                 |
| <b>Thermal Stability:</b>         | --                                 |
| <b>Radiation Stability:</b>       | --                                 |
| <b>Biological Stability:</b>      | --                                 |
| <b>Leach Resistance:</b>          | --                                 |
| <b>Immersion Stability:</b>       | --                                 |
| <b>Free Liquids:</b>              | --                                 |
| <b>Chemical Durability:</b>       | --                                 |
| <b>Compositional Flexibility:</b> | --                                 |
| <b>Gas Generation:</b>            | --                                 |
| <b>RCRA Compliance:</b>           | --                                 |
| <b>Conclusions:</b>               | --                                 |

## HYDRAULIC CEMENT SUMMARIES

NO. 32

TITLE: Measurement and Control of Cement Set Times in Waste Solidification

AUTHOR: J. A. Stone; P. D. d'Entremont

PUBLICATION: DP-1404

DATE: September 1976

ORGANIZATION: Savannah River Laboratory

SUMMARY: A commercial retarder was added to cement-sludge formulations to achieve set times that are workable for routine handling equipment. Set times and compressive strengths were measured.

|                            |   |
|----------------------------|---|
| Waste Form:                | Cement and/or concrete  |
| Waste Type:                | Simulated or actual sludge waste  |
| Waste Loading:             | 10, 25, 40% of solids   |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | <u>For High Alumina Cement</u><br>Tank 5: 9,500 psi at 10% sludge, 9,000 psi at 25%, and 3,000 psi at 40%<br>Tank 13: 900 psi at 10% sludge, 7,000 psi at 25%, and 1,600 psi at 40%<br>Tank 15: 6,800 psi at 10% sludge, 4,800 psi at 25%, and 1,400 psi at 40% |

## HYDRAULIC CEMENT SUMMARIES

NO. 33

**TITLE:** Immobilization of Technetium and Nitrate in Cement-Based Materials

**AUTHOR:** O.I. Tallent; et al.

**PUBLICATION:**

**DATE:** 1988

**ORGANIZATION:** Martin Marietta

**SUMMARY:** The leachability of technetium and nitrate wastes immobilized in cement-based materials have been investigated.

**Waste Form:** Blends of Type I-II cement/slag/flyash

**Waste Type:** Simulated double shell slurry feed from Hanford

**Waste Loading:** --

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Leachability index for nitrate ranged from 7.0 to 7.5 and for technetium from 8.0 to 8.3

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Decreasing waste concentration by dilution with water decreased leachability to Tc ion, but did not affect nitrate ion leachability. Chemical reducing properties of the slag reduce the Tc from the pertechnetate to the dioxide, a less soluble and slower leaching species.

## HYDRAULIC CEMENT SUMMARIES

NO. 34

**TITLE:** Cement Solidification of Spent Ion Exchange Resins Produced by the Nuclear Industry

**AUTHOR:** C. Taouen; B. Vigreux

**PUBLICATION:** Spectrum '88

**DATE:** September 11-15, 1988

**ORGANIZATION:** SGN (France)

**SUMMARY:** A mobile cementation solidification was designed to treat the resin waste.

**Waste Form:** Cement solidification

**Waste Type:** Spent ion exchange resin

**Waste Loading:** 40 to 75%

**Development Status:** The SGN resin measured 8 mPa after 28 days and 9 mPa after 90 days.

**Compressive Strength:** 10% resin measured 26 mPa after 28 days and 31 mPa after 90 days.

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** --

## HYDRAULIC CEMENT SUMMARIES

NO. 35

**TITLE:** Surface Encapsulation Process for Managing Low-Level  
Radioactive Wastes

**AUTHOR:** S. L. Unger; R. W. Telles

**PUBLICATION:**

**DATE:** 1986

**ORGANIZATION:** Environmental Protection Polymers, Inc.

**SUMMARY:**

**Waste Form:** Surface encapsulated agglomerated modules - agglomerates  
of mixed LLRW with co-reacted polyethylene and  
polybutadiene

**Waste Type:** Low level radioactive waste

**Waste Loading:** 50% of total (including jacket)

**Development Status:** Prototype in development

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** --

**HYDRAULIC CEMENT SUMMARIES**

NO. 36

**TITLE:** Alternatives for Managing Wastes from Reactors and Post-Fission Operations in the LWR Fuel Cycle - Vol. 2

**AUTHOR:** United States - Energy Research & Development Administration

**PUBLICATION:** ERDA-76-43

**DATE:** May 1976

**ORGANIZATION:** United States - Energy Research & Development Administration

**SUMMARY:** Volume 2 addresses solidification methods and waste forms including comparisons.

|                            |    |
|----------------------------|----|
| Waste Form:                | -- |
| Waste Type:                | -- |
| Waste Loading:             | -- |
| Development Status:        | -- |
| Compressive Strength:      | -- |
| Thermal Stability:         | -- |
| Radiation Stability:       | -- |
| Biological Stability:      | -- |
| Leach Resistance:          | -- |
| Immersion Stability:       | -- |
| Free Liquids:              | -- |
| Chemical Durability:       | -- |
| Compositional Flexibility: | -- |
| Gas Generation:            | -- |
| RCRA Compliance:           | -- |
| Conclusions:               | -- |



**HYDRAULIC CEMENT SUMMARIES**

NO. 37

**TITLE:** The Cement Solidification Systems at LANL (1)**AUTHOR:** G. W. Veazey**PUBLICATION:****DATE:** December 1990**ORGANIZATION:** Los Alamos National Laboratory**SUMMARY:** 7A-50 infrequently used system.**Waste Form:** Portland cement**Waste Type:** TA-50 system pH-adjusted TRU-waste**Waste Loading:** 42% waste**Development Status:** --**Compressive Strength:** --**Thermal Stability:** --**Radiation Stability:** --**Biological Stability:** --**Leach Resistance:** --**Immersion Stability:** --**Free Liquids:** Some surface moisture**Chemical Durability:** --**Compositional Flexibility:** --**Gas Generation:** --**RCRA Compliance:** --**Conclusions:** Corrosion of drum can occur if moisture gains entrance into drum. Entrance may be made through carbon filters on drum vents if not protected from environment.

## HYDRAULIC CEMENT SUMMARIES

NO. 38

**TITLE:** The Cement Solidification Systems at LANL (2)

**AUTHOR:** G. W. Veazey

**PUBLICATION:** LA-UR-90-4161

**DATE:** December 1990

**ORGANIZATION:** Los Alamos National Laboratory

**SUMMARY:** Problems of drum corrosion in extended outdoor storage prior to burial and appearance of free liquid in sealed drums are addressed.

**Waste Form:** Portland cement and Envirostone

**Waste Type:** Evaporator bottoms TRU-level waste

**Waste Loading:** 23 gal sludge per 55-gal drum

**Development Status:** Existing operating system

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** Appearance of free liquid possibly linked to gamma irradiation

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** Appearance of more than 15 liters low-level free liquid in sealed drums after 8-44 weeks.

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Theories have been proposed for the appearance of free liquids, although the exact causes have not been pinpointed. Inadequate mixing, gamma irradiation, and exposure of drums to rain are likely causes.

## HYDRAULIC CEMENT SUMMARIES

NO. 39

TITLE: Solid Forms for Savannah River Plant High-Level Waste

AUTHOR: R. M. Wallace; H. L. Hull; R. F. Bradley

PUBLICATION: DP-1335

DATE: December 1973

ORGANIZATION: Savannah River Laboratory

SUMMARY: This study was made to evaluate candidate solid waste forms and solidification processes.

Waste Form: --

Waste Type: --

Waste Loading: --

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: --

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: --

## HYDRAULIC CEMENT SUMMARIES

NO. 40

TITLE: Evaluation of Solidification/Stabilization As a Best Demonstrated Available Technology for Contaminated Soils

AUTHOR: L. Weitzman; L. E. Hamel

PUBLICATION: PB89-169908

DATE: March 1989

ORGANIZATION: Acurex Corporation

SUMMARY: Demonstrated that soil could be immobilized and contents stabilized by use of cement, lime, or a lime flyash mixture.

|                            |   |
|----------------------------|---|
| Waste Form:                | Portland cement, lime kiln dust, equal parts of lime and flyash |
| Waste Type:                | Hazardous contaminated soil                                     |
| Waste Loading:             | --  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | --  |

## HYDRAULIC CEMENT SUMMARIES

NO. 41

TITLE: Saltstone Processing Startup at the Savannah River Plant

AUTHOR: E. L. Whilhite; R. L. Hooker; H. F. Sturm; C. A. Langton; E. S. Occpihti

PUBLICATION: Spectrum '88

DATE: September 11-15, 1988

ORGANIZATION: E. I. DuPont de Nemours and Company

SUMMARY: Saltstone was shown to be non-hazardous after EP and TCLP testing. The feed is hazardous because of Cr and other metals.

Waste Form: Saltstone grout

Waste Type: Tank waste (LLW)

Waste Loading: --

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: --

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Diffusion Coefficients:  $1.3 \times 10^{-9} \text{ cm}^2/\text{sec}$  for nitrate;  $4 \times 10^{-12} \text{ cm}^2/\text{sec}$  for technetium;  $4 \times 10^{-13} \text{ cm}^2/\text{sec}$  for chromium.

## HYDRAULIC CEMENT SUMMARIES

NO. 42

TITLE: Grout Disposal System for Hanford Site Mixed Waste

AUTHOR: J. E. Van Beek; D. D. Wodrich

PUBLICATION: Waste Management '90, pp 797-802

DATE: March 1990

ORGANIZATION: Westinghouse Hanford Company

SUMMARY: The Grout Treatment Facility has been constructed at Hanford for processing liquid radioactive and hazardous tank wastes into a cement-based solid for disposal in near-surface concrete vaults.

|                            |  |
|----------------------------|--|
| Waste Form:                | Grout  |
| Waste Type:                | Phosphate/Sulfate salt solution  |
| Waste Loading:             | --   |
| Development Status:        | Large-scale disposal has been performed  |
| Compressive Strength:      | 200 to 2000 psi  |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | Leach index of Cs-137 is 9 and of Sr-90 is 14  |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | The grouting process combines cementitious materials with liquid waste to form grout slurry which is pumped to large, underground concrete vaults for solidification and final disposal. The Grout Treatment Facility disposal system design is a unique concept for meeting both hazardous and low-level waste disposal criteria. |

## HYDRAULIC CEMENT SUMMARIES

NO. 43

TITLE: Alternatives for Managing Wastes from Reactors and Post-Fission Operations in the LWR Fuel Cycle

AUTHOR: C. R. Cooley, et al.

PUBLICATION: ERDA-76-42, Vol. 2 of 5

DATE: May 1976

ORGANIZATION: US ERDA

## SUMMARY:

Waste Form: Portland cement, bitumen, UF

Waste Type:  $\text{Na}_2\text{SO}_4$ , boric acid, bead resins, diatomaceous earth

Waste Loading: --

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance:  $10^{-5}$  to  $10^{-1}$  gm/ $\text{cm}^2$  day for sodium ion and  $10^{-9}$  to  $10^{-7}$  gm/ $\text{cm}^2$  day for actinides

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Typical waste/cement formulations have been presented. Solutes in waste solutions and slurries can be incorporated in cement products at salt/cement weight ratios of 0.5, although more typically from 0.15 to 0.3. The maximum water to cement weight ratio is 0.5 and the minimum is 0.25. Sodium silicate as an additive helps to lower final volume increase factor.

## HYDRAULIC CEMENT SUMMARIES

NO. 44

**TITLE:** Treatment Technology for Transuranic Waste Streams--  
Cementation, Vitrification, and Incineration Testing for the  
treatment of Spent Ion Exchange Media

**AUTHOR:** B. G. Place

**PUBLICATION:** WHC-EP-0462

**DATE:** April 1992

**ORGANIZATION:** Westinghouse Hanford Company

**SUMMARY:** Cementation pretreatment testing was conducted for ion exchange  
resins and zeolites from a number of Hanford Site facilities.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | 1. Cement 2. Glass  |
| <b>Waste Type:</b>                | Organic ion exchange resins and zeolites  |
| <b>Waste Loading:</b>             | --  |
| <b>Development Status:</b>        | --  |
| <b>Compressive Strength:</b>      | 920-3700 psi  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | --  |
| <b>Biological Stability:</b>      | --  |
| <b>Leach Resistance:</b>          | --  |
| <b>Immersion Stability:</b>       | --  |
| <b>Free Liquids:</b>              | --  |
| <b>Chemical Durability:</b>       | --  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | Attritive and size-reduction pretreatment methods proved<br>most effective in producing a stable final waste form.<br>Pretreated zeolite exhibited substantially higher compressive<br>strengths and higher waste packaging efficiencies when<br>compared to untreated zeolite. |



## HYDRAULIC CEMENT SUMMARIES

NO. 45

TITLE: Cementation--A Solution for Final Disposal?

AUTHOR: B. L. Ganser

PUBLICATION: Waste Management '90

DATE: March 1990

ORGANIZATION: NUKEM Gmb H

SUMMARY: The suitability of cemented waste forms for final disposal in Germany has been evaluated through several tests from 1979 to 1989.

Waste Form: Cement and additives

Waste Type: Evaporator concentrates, resins, filter sludges

Waste Loading: Up to 40 wt% waste content (dry substance)

Development Status: --

Compressive Strength: 10-100 MPa

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance:  $10^{-4}$  to  $7 \times 10^{-3}$  gm/cm<sup>2</sup> day

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: Blast furnace slag cement gave lower average leach rates than ordinary Portland cement. Long term leach tests showed that after about 5 years there was a sharp decrease in leach rates. Direct cementation of liquid wastes showed excellent mechanical properties.

## HYDRAULIC CEMENT SUMMARIES

NO. 46

TITLE: Results of Field Testing of Waste Forms Using Lysimeters

AUTHOR: J. W. McConnell, Jr.; R. D. Rogers

PUBLICATION: Waste Management '90

DATE: February 1990

ORGANIZATION: Idaho National Engineering Laboratory

SUMMARY: Results of lysimeter tests on ion-exchange resins are presented and the use of lysimeter data in performance assessment is discussed.

|                            |  |
|----------------------------|--|
| Waste Form:                | Portland Type I-II   |
| Waste Type:                | Ion-exchange resins from prefilters  |
| Waste Loading:             | --   |
| Development Status:        | --   |
| Compressive Strength:      | --   |
| Thermal Stability:         | --   |
| Radiation Stability:       | --   |
| Biological Stability:      | --   |
| Leach Resistance:          | --   |
| Immersion Stability:       | --   |
| Free Liquids:              | --   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | --   |
| Conclusions:               | Sr-90 is more prevalent than Cs-137, CO-60 or Sb-1215 in collected liquid samples. The limiting step in receiving Sr-90 in the leachate is not release of the nuclide from the waste forms, but rather it is the soil characteristics that limit movement. |

**HYDRAULIC CEMENT SUMMARIES****NO. 47**

**TITLE:** Division of Waste Management, Production and Reprocessing Programs Report for January-June 1977

**AUTHOR:** Compiled by R. E. Lerch

**PUBLICATION:** HEDL-TME 77-74

**DATE:** July 1977

**ORGANIZATION:** Hanford Engineering Development Laboratory

**SUMMARY:** The report addresses immobilization of boric acid waste and provides composition phase diagrams for boric acid and ion-exchange resin. It also presents thermal stability data for asphalt and urea-formaldehyde organic binders.

**Waste Form:** Variable

**Waste Type:** Low-level waste and radwaste

**Waste Loading:** Variable

**Development Status:** --

**Compressive Strength:** Data given for boric acid and ion-exchange waste forms; compression strengths up to 70 kg/cm<sup>2</sup> were reported.

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Three-phase diagrams of cement/water/waste were determined for boric acid. Problems of cementing resins are discussed, some lower waste loading limits are stable. Thermal profiles on asphalt and urea-formaldehyde waste products are given. Leach rate data is provided for cesium and strontium tracers in test waste forms.

**SULFUR POLYMER CEMENT SUMMARIES**

**SULFUR POLYMER CEMENT SUMMARIES INDEX**

- 
- |       |              |   |
|-------|--------------|---|
| NO. 1 | AUTHOR:      | Darnell, G.R.; Aldrich, W.C; Logan, J.A.  |
|       | TITLE:       | Full-Scale Tests of Sulfur Polymer Cement and Nonradioactive Waste in Heated and Unheated Prototypical Containers |
|       | PUBLICATION: | EGG-WM-10109, Idaho National Engineering Laboratory   |
|       | DATE:        | 1992  |
- 
- |       |              |   |
|-------|--------------|---|
| NO. 2 | AUTHOR:      | Johnson, E.R.   |
|       | TITLE:       | An Assessment of the Potential of Modified Sulfur Cement for the Conditioning of Low-Level Radioactive Waste for Dumping at Sea |
|       | PUBLICATION: | JAI-200   |
|       | DATE:        | 6/83  |
- 
- |       |              |   |
|-------|--------------|---|
| NO. 3 | AUTHOR:      | van Dalen, A.; Rijpkema, J.E.   |
|       | TITLE:       | Modified Sulphur Cement: A Low Porosity Encapsulation Material for Low, Medium, and Alpha Waste |
|       | PUBLICATION: | Nuclear Science and Technology (Commission of the European Communities), EUR 12303              |
|       | DATE:        | 1989  |
- 
- |       |              |  |
|-------|--------------|--|
| NO. 4 | AUTHOR:      | Kalb, P.D.; Heiser III, J.H.; Colombo, P.  |
|       | TITLE:       | Modified Sulfur Cement Encapsulation of Mixed Waste Contaminated Incinerator Fly Ash |
|       | PUBLICATION: | Waste Management, Vol. II, pp. 147-153   |
|       | DATE:        | 1991   |
- 
- |       |              |  |
|-------|--------------|--|
| NO. 5 | AUTHOR:      | Darnell, G.R.  |
|       | TITLE:       | Sulfur Polymer Cement, a Solidification and Stabilization Agent for Hazardous and Radioactive Wastes |
|       | PUBLICATION: | First International Mixed Waste Symposium, Washington, D.C.,   |
|       | DATE:        | 1991   |
- 
- |       |              |  |
|-------|--------------|--|
| NO. 6 | AUTHOR:      | Kalb, P.D.; Heiser III, J.H.; Pietrzak, R.; Colombo, P.                    |
|       | TITLE:       | Durability of Incinerator Ash Waste Encapsulated in Modified Sulfur Cement |
|       | PUBLICATION: | BNL-45292, 1991 Incineration Conference, Knoxville, TN                     |
|       | DATE:        | 1991   |
-

- NO. 7    AUTHOR:            Kalb, P.D.; Heiser III, J.H.; Pietrzak, R.; Colombo, P.  
          TITLE:                Comparison of Modified Sulfur Cement and Hydraulic  
                                 Cement for Encapsulation of Radioactive and Mixed  
                                 Wastes  
          PUBLICATION:      Twelfth Annual U.S. DOE Low-Level Waste Management  
                                 Conference  
          DATE:                1991
- 
- NO. 8    AUTHOR:            Kalb, P.D.; Heiser III, J.H.; Colombo, P.  
          TITLE:                Encapsulation of Mixed Radioactive and Hazardous  
                                 Waste Contaminated Incinerator Ash in Modified Sulfur  
                                 Cement, BNL-43691  
          PUBLICATION:      Incineration Conference 1990, San Diego, CA,  
          DATE:                1991
- 
- NO. 9    AUTHOR:            McBee, W.C.; Sullivan, T.A.; Fike, H.L.  
          TITLE:                Sulfur Construction Materials  
          PUBLICATION:      United States Department of the Interior Bureau of Mines  
                                 Bulletin 678  
          DATE:                1985
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**SULFUR POLYMER CEMENT SUMMARIES****NO. 1**

**TITLE:** Full-Scale Tests of Sulfur Polymer Cement and Nonradioactive Waste in Heated and Unheated Prototypical Containers

**AUTHORS:** G.R. Darnell; W.C. Aldrich; J.A. Logan

**PUBLICATION:** EGG-WM-10109, Idaho National Engineering Laboratory

**DATE:** 1992

**ORGANIZATION:** EG&G Idaho, Inc.

**SUMMARY:** Describes and discusses the first full-scale tests of SPC as a stabilization agent for simulated radioactive incinerator ash and steel waste. Tests were conducted with a horizontal, externally-and-internally heated, production-scale mixer. A prototypical, rectangular, 24-ft<sup>3</sup>, thin-walled, steel waste container with no appendages was used to hold 800 lb of simulated sized waste while molten SPC combined with incinerator hearth ash under 3/8-in. diameter was poured into it. One test was with an ambient-temperature container and steel waste. Another test was with the steel waste, the container, and the mold-form preheated to melt temperature in an oven. Destructive examination showed better results in the concrete mass where heating was applied. It was proposed that better heating of the container and waste be provided for the next tests.

|                                 |   |
|---------------------------------|---|
| <b>Waste Form:</b>              | Sulfur polymer cement                                   |
| <b>Waste Type:</b>              | Incinerator hearth ash and steel pipes (nonradioactive) |
| <b>Waste Loading:</b>           | 40 wt% ash (not counting 800 lb of steel pipe)          |
| <b>Development Status:</b>      | 50% established   |
| <b>Compressive Strength:</b>    | Lab tests not conducted, but resisted hammer blows      |
| <b>Thermal Stability:</b>       | --  |
| <b>Radiation Stability:</b>     | --  |
| <b>Biological Stability:</b>    | --  |
| <b>Leach Resistance:</b>        | --  |
| <b>Immersion Stability:</b>     | --  |
| <b>Free Liquids:</b>            | None  |
| <b>Chemical Durability:</b>     | --  |
| <b>Compositional Stability:</b> | --  |
| <b>Gas Generation:</b>          | --  |
| <b>RCRA Compliance:</b>         | --  |

**SULFUR POLYMER CEMENT SUMMARIES**

**NO. 1 (cont.)**

**Conclusions:**

Preheating the waste container and its 800 lb of steel "waste" to 135°C greatly improved the pour and eliminated nearly all voids experienced in the "cold test," even though the container and contents lost ~34°C during the pour. The evidence indicates that the container and its contents need to be preheated and then maintained at SPC melt temperatures throughout the pour.



**SULFUR POLYMER CEMENT SUMMARIES****NO. 2**

**TITLE:** An Assessment of the Potential of Modified Sulfur Cement for the Conditioning of Low-Level Radioactive Waste for Dumping at Sea

**AUTHORS:** E.R. Johnson

**PUBLICATION:** JAI-200

**DATE:** June 1983

**ORGANIZATION:** E. R. Johnson Associates, Inc. for Brookhaven National Laboratory

**SUMMARY:** This report presents curves on strength versus time, strength growth curve of SPC versus Portland cement concrete, stress strain, modulus of elasticity, freeze-thaw cycles to failure for SPC and Portland cement concrete, linear thermal expansion, and variation of compressive strength and specific gravity. SPC's properties for radioactive waste stabilization include rapid curing; no residual free water; little degradation (especially from radiation); resistant to chemical and physical degradation, not chemically reactive, low leaching rate, particularly by saline waters or ground; and is easily handled and controlled by operators.

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Sulfur polymer cement  |
| <b>Waste Type:</b>                | --   |
| <b>Waste Loading:</b>             | --   |
| <b>Development Status:</b>        | Initial effort for evaluation as final waste form  |
| <b>Compressive Strength:</b>      | Equal to or greater than that of hydraulic concrete  |
| <b>Thermal Stability:</b>         | Resistant to thermal cycling   |
| <b>Radiation Stability:</b>       | --   |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | --   |
| <b>Immersion Stability:</b>       | Relatively impervious to water and moisture  |
| <b>Free Liquids:</b>              | None   |
| <b>Chemical Durability:</b>       | Strong, corrosion resistant concrete; highly resistant to dissolution and chemical corrosion |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |

**SULFUR POLYMER CEMENT SUMMARIES**

**NO. 2 (cont.)**

Conclusions:

The author predicted the attributes of SPC as a radioactive waste stabilization agent that have since proven true. A genesis report for radioactive and hazardous waste stabilization.

## SULFUR POLYMER CEMENT SUMMARIES

NO. 3

**TITLE:** Modified Sulphur Cement: A Low Porosity Encapsulation Material for Low, Medium and Alpha Waste

**AUTHOR:** van Dalen, A., and Rijpkema, J. E.

**PUBLICATION:** Nuclear Science And Technology (Commission of the European Communities), EUR 12303

**DATE:** 1989

**ORGANIZATION:** Energy Research Foundation ECN, 1755 ZG Petten, The Netherlands

**SUMMARY:** This report summarizes tests conducted with modified sulfur cement, more commonly referred to as sulfur polymer cement (SPC). It confirms tests conducted in the U.S.A. at Brookhaven National Laboratory (BNL). They, like BNL, combined SPC with dried ion-exchange resins. In both cases the resins took on moisture very quickly and ruptured the concrete. The Netherlands tried a different approach that worked. They combined the ion-exchange resins with SPC, raised the temperature from the normal 135°C to 230 to 250°C and held that temperature for 3 hours. The test specimen was immersed in water for one year with no degradation. No information is provided on how they handled the gaseous emissions. All of the NRC's laboratory tests were conducted.

**Waste Form:** Sulfur polymer cement concrete

**Waste Type:** Ion-exchange resins; sludges with 5% U, 1% Ti, 3% Cu, 1% Mn, 3% Al, and 2% Na; precipitates; lead iodide (model for <sup>129</sup>I); PbI<sub>2</sub>; incinerator ashes with borate, CaCO<sub>3</sub>, CsCl, EuCl<sub>3</sub>, and 1.86% K<sub>2</sub>O, 1.04% Na<sub>2</sub>, 2.07% Fe<sub>2</sub>O<sub>3</sub>, 28.38% Al<sub>2</sub>O<sub>3</sub>, 16.09% CaO, 2.12% MgO, 43.88% SiO<sub>2</sub>, 1.72% TiO<sub>2</sub>, 0.92% BaO, 0.10% MnO, and 0.32% P<sub>2</sub>O<sub>3</sub>; <sup>137</sup>Cs; <sup>60</sup>Co; SrCO<sub>3</sub>; toxic metals; and other metals

**Waste Loading:** See above

**Development Status:** Second major effort in testing SPC for final waste form

**Compressive Strength:** 3333 psi after one day and 9275 psi after two years

**Thermal Stability:** Passed tests

**Radiation Stability:** 6232 psi before 10<sup>8</sup> rad and 8840 psi after

**Biological Stability:** --

**Leach Resistance:** --

**SULFUR POLYMER CEMENT SUMMARIES****NO. 3 (cont.)**

|                            |  |
|----------------------------|--|
| Immersion Stability:       | Passed tests   |
| Free Liquids:              | None   |
| Chemical Durability:       | No problem noted. Irradiation suggested longevity  |
| Compositional Flexibility: | Improved by the addition of $Al_2O_3$ and/or $Fe_2O_3$   |
| Gas Generation:            | Subjecting SPC with samples containing $PbI_2$ , incinerator ash, and borate waste to $10^8$ rad produced no gaseous radiolysis products             |
| RCRA Compliance:           | Not tested to EPA's TCLP   |
| Conclusions:               | If we can reproduce their excellent results with ion-exchange resins without any safety concern, that fact alone makes this report a valuable asset. |

## SULFUR POLYMER CEMENT SUMMARIES

NO. 4

**TITLE:** Modified Sulfur Cement Encapsulation of Mixed Waste Contaminated Incinerator Fly Ash

**AUTHORS:** P. D. Kalb; J.H. Heiser III; P. Colombo

**PUBLICATION:** Waste Management, Vol. II, pp. 147-153

**DATE:** 1991

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Upon cooling, SPC achieves compressive strengths averaging 4,000 psi. Due to high concentrations of zinc, lead, cadmium, sodium, and chlorine in the mixed waste fly ash obtained from the Waste Experimental Reduction Facility (WERF), it is difficult to combine with portland cement. That waste also contained approximately 1.5 B/g (pc/g) of activity made up of mixed fusion products--primarily <sup>137</sup>Cs and activation products--which were primarily <sup>57</sup>Co and <sup>125</sup>Sb. Highly soluble metal chloride salts (mostly zinc chloride) further complicate this waste.

**Waste Form:** Sulfur polymer cement

**Waste Type:** Mixed waste fly ash (37 wt% zinc, 7.5 wt% lead, 0.2 wt% cadmium)

**Waste Loading:** 43 wt%

**Development Status:** Early

**Compressive Strength:** 4,000 psi (>27.6 MPa), which is twice the binder strength alone

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** 3 to 4 times better than Portland cement

**Immersion Stability:** --

**Free Liquids:** None

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** Passed TCLP with high loadings of lead and cadmium

**Conclusions:** Presence of high concentrations of zinc, lead, cadmium, sodium, and chloride make loading in SPC more successful than in portland cement by a factor of 3.2..

**SULFUR POLYMER CEMENT SUMMARIES**

NO. 5

**TITLE:** Sulfur Polymer Cement, a Solidification and Stabilization Agent for Hazardous and Radioactive Wastes

**AUTHORS:** G.R. Darnell

**PUBLICATION:** First International Mixed Waste Symposium, Washington, D.C.

**DATE:** 1991

**ORGANIZATION:** EG&G Idaho, Inc., Idaho National Engineering Laboratory

**SUMMARY:** SPC's resistance to acids, salts, water penetration, and physical failure are discussed. It discusses the ability to remelt stabilized waste, should some remediation decree be rendered. SPC's propensity for microencapsulation of toxic metals is discussed. SPC's ability to accept higher loadings of troublesome wastes than other agents is discussed.

**Waste Form:** Sulfur polymer cement concrete

**Waste Type:** MW incinerator flyash and hearthash, boric acid, sodium sulfate

**Waste Loading:** 43 wt% mixed waste flyash (37 wt% zinc, 7.5 wt% lead, 0.2 wt% cadmium); 40 wt% boric acid; 40 wt% sodium sulfate; 40 wt% hearthash

**Development Status:** Early

**Compressive Strength:** Averages ~4,000 psi

**Thermal Stability:** --

**Radiation Stability:** Gains compressive strength

**Biological Stability:** Passed test with no visible deterioration

**Leach Resistance:** 4 to 8 orders of magnitude better than NRC requirements

**Immersion Stability:** Passes with addition of 0.5 wt% Omens Corning glass fibers

**Free Liquids:** No free liquid in final waste form

**Chemical Durability:** Passed all EPA and NRC lab-scale tests

**Compositional Flexibility:** Passed all EPA and NRC lab-scale tests

**Gas Generation:** No gas generation during  $10^8$  rad test

**RCRA Compliance:** --

**Conclusions:** Sulfur polymer cement resists attack by most acids and salts. This and other facts suggest longevity. SPC gains strength when subjected to high radiation, which also suggests longevity. SPC can pass the EPA's TCLP with heavy loadings of toxic metals and no volatilization of toxic metals in the process.

**SULFUR POLYMER CEMENT SUMMARIES**

NO. 6

**TITLE:** Durability of Incinerator Ash Waste Encapsulated in Modified Sulfur Cement

**AUTHORS:** P. D. Kalb, J. H. Heiser III, R. Pietrzak, and P. Colombo

**PUBLICATION:** BNL-45292, 1991 Incineration Conference, Knoxville, TN

**DATE:** 1991

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Brookhaven National Laboratory completed durability tests with SPC laden with mixed waste fly ash. Tests completed were those prescribed by the NRC; specifically, the tests were compressive strength, biodegradation, irritation, water immersion, thermal cycling, and leaching. Incinerator hearth ash and fly ash from the rotary-kiln incinerator at Rocky Flats Plant had no problem passing the tests with a 43 wt% loading in SPC. Fly ash from the INEL's WERF were loaded at the same ratio, but the high concentrations of soluble metal salts tended to deteriorate in the immersion test. The same stabilized waste was subjected to EPA's TCLP test with successful results.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Sulfur polymer cement                                   |
| <b>Waste Type:</b>                | RFP ash and INEL ash (mixed waste)                      |
| <b>Waste Loading:</b>             | 43 wt%  |
| <b>Development Status:</b>        | --  |
| <b>Compressive Strength:</b>      | 4,430 psi (30.5 MPa) at maximum ash loading of 43 wt%   |
| <b>Thermal Stability:</b>         | No significant change in compressive strength           |
| <b>Radiation Stability:</b>       | Passed tests  |
| <b>Biological Stability:</b>      | No visible microbial growth of either bacteria or fungi |
| <b>Leach Resistance:</b>          | 5 to 8 orders of magnitude slower than NRC leach index  |
| <b>Immersion Stability:</b>       | Glass fibers required to pass with one ash tested       |
| <b>Free Liquids:</b>              | None  |
| <b>Chemical Durability:</b>       | Passed all tests  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | Passed TCLP test with 43 wt% of mixed waste flyash      |

**SULFUR POLYMER CEMENT SUMMARIES**

**NO. 6 (cont.)**

**Conclusions:**

Compressive strength increased with increased waste loadings (4,430 psi at 43 wt% loading). Had to add glass 0.5 wt% fibers in the concrete to pass immersion test with one of the ashes tested (contained extremely high concentrations of soluble metal salts).



## SULFUR POLYMER CEMENT SUMMARIES

NO. 7

**TITLE:** Comparison of Modified Sulfur Cement and Hydraulic Cement for Encapsulation of Radioactive and Mixed Wastes

**AUTHORS:** P. D. Kalb; J. H. Heiser III; R. Pietrzak; P. Colombo

**PUBLICATION:** Twelfth Annual U.S. DOE Low-Level Waste Management Conference

**DATE:** 1991

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Modified sulfur cement (SPC) is compared to hydraulic cements in its capabilities as radioactive and mixed waste stabilization agent. SPC is a thermoplastic that melts at 120°C and requires no activator, while hydraulic cements do require an activator. At Brookhaven National Laboratory they compared the loadings of four divergent wastes that could be stabilized in the 2 cements to EPA standards. The four wastes were loaded in SPC and hydraulic cements. SPC was capable of passing the TCLP with the following waste loadings: incinerator bottom ash, 1.4 wt%; boric acid, 2.6 wt%; incinerator fly ash, 2.7 wt%; and sodium sulfate, 6.0 wt%. The tests showed that an average of 3.2 times more waste by volume could be placed in SPC than in portland cement. It was concluded that untreated nitrate salts and ion-exchange resins could not be successfully stabilized in SPC.

|                                    |  |            |             |            |
|------------------------------------|--|------------|-------------|------------|
| Waste Form:                        | Sulfur polymer cement                                  |            |             |            |
| Waste Type:                        | Sodium sulfate,  | fly ash,   | bottom ash, | boric acid |
| Waste Loading: <sup>a</sup>        | 40(9)  | 43(16)     | 43(40)      | 40(15)     |
| Development Status:                | In earliest stages of testing and development          |            |             |            |
| Compressive Strength: <sup>a</sup> | 4300(3000)   | 4200(4000) | 6500(4000)  | 2000(1400) |
| Thermal Stability: <sup>b</sup>    | 4300(3700)   | 4200(2800) | 6500(4700)  | 2000(2200) |
| Radiation Stability:               | --   |            |             |            |
| Biological Stability:              | --   |            |             |            |
| Leach Resistance:                  | 4 to 8 orders of magnitude better than NRC leach index |            |             |            |
| Immersion Stability: <sup>c</sup>  | 4300(3000)   | 4300(4000) | 6500(4000)  | 2000(1400) |

- a. Expressed in terms of weight percent. Unbracketed is for SPC and bracketed is for portland cement.
- b. Before cycling is unbracketed and after cycling is bracketed.
- c. <sup>60</sup>Co is unbracketed and <sup>137</sup>Cs is bracketed.

**SULFUR POLYMER CEMENT SUMMARIES**

**NO. 7 (cont.)**

|                            |  |
|----------------------------|--|
| Free Liquids:              | None   |
| Chemical Durability:       | --   |
| Compositional Flexibility: | --   |
| Gas Generation:            | --   |
| RCRA Compliance:           | Passed TCLP  |
| Conclusions:               | SPC will accommodate more waste than portland cement with higher compressive strength. |

**SULFUR POLYMER CEMENT SUMMARIES****NO. 8**

**TITLE:** Encapsulation of Mixed Radioactive and Hazardous Waste Contaminated Incinerator Ash in Modified Sulfur Cement

**AUTHORS:** P. D. Kalb; J. H. Heiser III; P. Colombo

**PUBLICATION:** BNL-43691, Incineration Conference 1990, San Diego, CA

**DATE:** 1990

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Modified sulfur cement (SPC) was tested with mixed waste fly ash from the INEL's WERF. Wet chemical and solid phase waste characterization analyses showed high concentrations of lead and cadmium and other soluble metal salts in the fly ash. The EPA's TCLP test was passed with a 43 wt% loading of the fly ash after adding 7 wt% sodium sulfide, which enhanced the retention of the soluble metal salts. This work was completed at Brookhaven National Laboratory.

**Waste Form:** Sulfur polymer cement

**Waste Type:** Mixed waste fly ash containing 7.5 wt% lead and 0.2 wt% cadmium

**Waste Loading:** 43 wt%

**Development Status:** Early stages of testing

**Compressive Strength:** Exceeds 4,000 psi at 43 wt% loading

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** Passed TCLP test at 43 wt% loading; well below EPA concentration criteria for delisting

**Conclusions:** Addition of 7 wt% sodium sulfide allowed more of toxic-metal-bearing ash to be added and pass the TCLP test.

**SULFUR POLYMER CEMENT SUMMARIES**

NO. 9

**TITLE:** Sulfur Construction Materials

**AUTHORS:** W.C. McBee; T.A. Sullivan; H.L. Fike

**PUBLICATION:** United States Department of the Interior Bureau of Mines Bulletin 678

**DATE:** 1985

**ORGANIZATION:** U.S. Bureau of Mines

**SUMMARY:** This report on SPC takes it from prehistoric times when elemental sulfur was used as a cement through its development as a modern cement, with some exceptional qualities. Discussed in this report are SPC's strengths and weaknesses, safety considerations, chemical reactions, and effects of aggregates. Charts present curves on viscosity dependency, strength, differential scanning calorimetry thermograms, linear thermal-expansion data, aggregate gradation, stress strain, compressive strength, load deflection, moisture, freeze-thaw durability, and others. Lists are provided that define the success of SPC in various acids and salts, and the time of exposure.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Sulfur polymer cement   |
| <b>Waste Type:</b>                | --  |
| <b>Waste Loading:</b>             | --  |
| <b>Development Status:</b>        | SPC developed 13 years earlier  |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | Successfully resists freeze-thaw cycles   |
| <b>Radiation Stability:</b>       | --  |
| <b>Biological Stability:</b>      | No reaction   |
| <b>Leach Resistance:</b>          | --  |
| <b>Immersion Stability:</b>       | Routinely used as tanks for acid and salt solutions   |
| <b>Free Liquids:</b>              | None; impermeable to water  |
| <b>Chemical Durability:</b>       | Resists most acids and salts that destroy Portland cement   |
| <b>Compositional Flexibility:</b> | Resists most acids and salts that destroy Portland cement   |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | This report does not discuss encapsulation of wastes, but indirectly addresses the problem that has plagued the waste stabilization industry: concrete degradation. |

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**CERAMIC SUMMARIES**

## CERAMIC SUMMARIES

NO. 1

TITLE: An ICPP Aluminum Phosphate Ceramic Waste Form: Synthesis and Room-Temperature Aqueous Stability

AUTHORS: Paul Silva; Barry E. Scheetz

PUBLICATION: Advances in Ceramics 8, American Ceramic Society, 1984

DATE: 1974

ORGANIZATION: Pennsylvania State University

SUMMARY: Describes the synthesis and dissolution behavior of a berlinite-based ( $\text{AlPO}_4$ ) ceramic encapsulant for the high-alumina nuclear waste at the Idaho Chemical Processing Plant (ICPP).

|                            |   |
|----------------------------|---|
| Waste Form:                | Aluminum phosphate (berlinite) low-temperature ceramic  |
| Waste Type:                | High-alumina nuclear waste at the Idaho Chemical Processing Plant (ICPP)  |
| Waste Loading:             | 42% (Simulated ICPP alumina waste)  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | --  |
| Leach Resistance:          | Very low for Cs and Sr over the pH range 3-9  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | Very low solubility ( $<10^{-5}$ mol/l) in the pH range 3-9   |
| Compositional Flexibility: | --  |
| Gas Generation:            | None  |
| RCRA Compliance:           | --  |
| Conclusions:               | The extreme insolubility of berlinite, coupled with the very low processing temperatures makes this a very attractive candidate waste form for high-alumina defense wastes. |

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**ORGANIC BINDER SUMMARIES**

**ORGANIC BINDER SUMMARIES INDEX**

- 
- |       |              |   |
|-------|--------------|---|
| NO. 1 | AUTHOR:      |   |
|       | TITLE:       | DOW Polymer   |
|       | PUBLICATION: | Literature Provided by Heather Holmes Burns on CIF Project, |
|       | DATE:        | 5/22/92   |
- 
- |       |              |   |
|-------|--------------|---|
| NO. 2 | AUTHOR:      |   |
|       | TITLE:       | Low-Level Radioactive Waste Volume Reduction and Stabilization Technologies and Resource Manual |
|       | PUBLICATION: | DOE/LLW-76T   |
|       | DATE:        | 12/88   |
- 
- |       |              |   |
|-------|--------------|---|
| NO. 3 | AUTHOR:      |   |
|       | TITLE:       | Treatment of Spent Ion-Exchange Resins for Storage and Disposal |
|       | PUBLICATION: | Technical Report Series No. 254                                 |
|       | DATE:        | 1985  |
- 
- |       |              |   |
|-------|--------------|---|
| NO. 4 | AUTHOR:      |   |
|       | TITLE:       | A Waste Inventory Report for Reactor and Fuel Fabrication Facility Wastes |
|       | PUBLICATION: | ONWI-20   |
|       | DATE:        | 1979  |
- 
- |       |              |   |
|-------|--------------|---|
| NO. 5 | AUTHOR:      | Akins, M.J.; Costomiris, G.; Delete, P.; Wilson, R.B. |
|       | TITLE:       | The Options for Solidifying Low Level Waste           |
|       | PUBLICATION: | Nuclear Engineering International, Vol. 28            |
|       | DATE:        | 3/83  |
- 
- |       |              |   |
|-------|--------------|---|
| NO. 6 | AUTHOR:      | Barletta, R.; Bowerman, B.; Davis, R.; Shea, C. |
|       | TITLE:       | Biodegradation Testing of Bitumen               |
|       | PUBLICATION: | BNL-NUREG-38999                                 |
|       | DATE:        | 1986  |
- 
- |       |              |   |
|-------|--------------|---|
| NO. 7 | AUTHOR:      | Bennet, B., et al.                                  |
|       | TITLE:       | Conditioning Options for Magnox Fuel Element Debris |
|       | PUBLICATION: | Radioactive Waste Management                        |
|       | DATE:        | 1984  |
-



- NO. 8 AUTHOR: Buckley, L.; Clegg, B.; Oldham, W.  
TITLE: Microbial Activity on Bituminized Radioactive Waste  
PUBLICATION: Radioactive Waste Management and the Nuclear Fuel  
Cycle, Vol. 6  
DATE: 3/85
- 
- NO. 9 AUTHOR: Colombo, P.; Neilson Jr., R.M.  
TITLE: Properties of Radioactive Wastes and Waste Containers-  
First Topical Report  
PUBLICATION: BNL-NUREG-50957  
DATE: 1979
- 
- NO. 10 AUTHOR: Colombo, P.; Neilson Jr., R.M.  
TITLE: Properties of Radioactive Wastes and Waste Containers  
PUBLICATION: BNL-NUREG-50957  
DATE: 1977
- 
- NO. 11 AUTHOR: Colombo, P.; Neilson Jr., R.M.  
TITLE: Critical Review of the Properties of Solidified Radioactive  
Waste  
PUBLICATION: BNL-NUREG-50591  
DATE: 1976
- 
- NO. 12 AUTHOR: Dougherty, D.R.; Pietrzak, R.F.; Fuhrmann, M.; Colombo,  
P.  
TITLE: Accelerated Leach Test(s) Program Annual Report  
PUBLICATION: BNL-52042  
DATE: 1986
- 
- NO. 13 AUTHOR: Fenner, O.H.  
TITLE: Chemical and Environmental Properties of Plastics and  
Elastomers  
PUBLICATION: Handbook of Plastics and Elastomers, C.A. Harper, ed.  
DATE: 1975
- 
- NO. 14 AUTHOR: Filter, H.E.  
TITLE: Vinyl Ester Solidification of Low-Level Radioactive  
Wastes  
PUBLICATION:  
DATE: 1979
-

- NO. 15    AUTHOR:            Fitzgerald, C.L.; Godbee, H.W.; Blanco, R.E.  
          TITLE:            The Feasibility of Incorporating Radioactive Wastes in  
                          Asphalt or Polyethylene  
          PUBLICATION:    Nuclear Applications and Technology, Vol. 9, No. 12  
          DATE:            1970
- 
- NO. 16    AUTHOR:            Franz, E.M.; Colombo, P.  
          TITLE:            Waste Form Evaluation Program, Final Report  
          PUBLICATION:    BNL-51954  
          DATE:            9/84
- 
- NO. 17    AUTHOR:            Franz, E.M.; Colombo, P.  
          TITLE:            Waste Form Evaluation Program, Final Report  
          PUBLICATION:    BNL-51954  
          DATE:            9/84
- 
- NO. 18    AUTHOR:            Franz, E.M.; Colombo, P.  
          TITLE:            Waste Form Evaluation Program, Final Report  
          PUBLICATION:    BNL-51954  
          DATE:            9/84
- 
- NO. 19    AUTHOR:            Franz, E.M.; Colombo, P.  
          TITLE:            Waste Form Evaluation Program, Final Report  
          PUBLICATION:    BNL-51954  
          DATE:            9/84
- 
- NO. 20    AUTHOR:            Franz, E.M.; Heiser, J.H.; Colombo, P.  
          TITLE:            Solidification of Problem Wastes, Annual Progress Report  
          PUBLICATION:    BNL-52078  
          DATE:            1987
- 
- NO. 21    AUTHOR:            Fuhrmann, M.; Neilson, R.M.; Colombo, P.  
          TITLE:            A Survey of Agents and Techniques Applicable to the  
                          Solidification of LLW  
          PUBLICATION:    BNL 51521  
          DATE:            12/81
- 
- NO. 22    AUTHOR:            Graves, J.J.  
          TITLE:            The Chemical Cleaning of Dresden Unit 1  
          PUBLICATION:    Transactions of the American Nuclear Society, Vol. 28  
          DATE:            1978
-



- NO. 29    AUTHOR:            Mattus, A.J.; Kaczmarzky, M.M.  
          TITLE:            Laboratory Performance Testing of an Extruded Bitumen  
                                  Containing a Surrogate, Sodium Nitrate-Based, Low-  
                                  Level Aqueous Waste  
          PUBLICATION:    Proceedings of the Symp. on Waste Management at  
                                  Tucson, AZ,  
          DATE:            3/1-5/87
- 
- NO. 30    AUTHOR:            Miyanoga, I.; Sakata, S.; Ito, A.; Amano, H.  
          TITLE:            Development of Radioactive Waste Management at the  
                                  Japan Atomic Energy Research Institute  
          PUBLICATION:    Nuclear Power and Its Fuel Cycle, Vol. 4, IAEA CN-  
                                  36/156  
          DATE:            1977
- 
- NO. 31    AUTHOR:            Moriyama, N.; Dojiri, S.; Honda, T.  
          TITLE:            Solidification of Powdered Ion Exchange Resins with  
                                  Polyethylene  
          PUBLICATION:    Nuclear and Chemical Waste Management, Vol. 3, No. 2  
          DATE:            1982
- 
- NO. 32    AUTHOR:            Moriyama, N.  
          TITLE:            Incorporation of an Evaporator Concentrate on  
                                  Polyethylene for BWR  
          PUBLICATION:    Nuclear Chemical Waste Management  
          DATE:            1982
- 
- NO. 33    AUTHOR:            Moriyama, N.  
          TITLE:            Solidification of Powdered Ion Exchange Resins with  
                                  Polyethylene  
          PUBLICATION:    Nuclear Chemical Waste Management  
          DATE:            1982
- 
- NO. 34    AUTHOR:            Moriyama, N.; Dojiri, S.; Matsuzuru, H.  
          TITLE:            Incorporation of BWR's Evaporator Concentrate in  
                                  Polyethylene  
          PUBLICATION:    Nuclear and Chemical Waste Management, Vol. 3, No. 1  
          DATE:            1982
-





**ORGANIC BINDER SUMMARIES****NO. 1****TITLE:** DOW Polymer**AUTHORS:****PUBLICATION:** Literature Provided by Heather Holmes Burns on CIF Project**DATE:** May 22, 1992**ORGANIZATION:****SUMMARY:** Discussion of waste form and process consideration.

**Waste Form:** DOW polymer (vinyl ester resin)  
**Waste Type:** Sodium sulfate salts and 5% ash fines  
**Waste Loading:** Waste/binder = 2.0/ 1.0  
**Development Status:** Commercial  
**Compressive Strength:** Solidified waste forms meet the NRC structural stability requirements.

**Thermal Stability:** --  
**Radiation Stability:** --  
**Biological Stability:** --  
**Leach Resistance:** Recent hazardous leachability test (TCLP tests) have proven that DOW Polymer waste forms can meet the EPA requirements for disposal for some waste types.

**Immersion Stability:** --  
**Chemical Durability:** --  
**Compositional Flexibility:** The solid waste product is a uniform mixture of finely dispersed liquid particles in a continuous matrix.

**Free Liquids:** When cured there is no free water  
**Gas Generation:** --  
**RCRA Compliance:** --  
**Conclusions:** DOW system is capable of safely processing wet solid radioactive waste generated by light water reactors.

## ORGANIC BINDER SUMMARIES

NO. 2

**TITLE:** Low-Level Radioactive Waste Volume Reduction and Stabilization Technologies and Resource Manual

**AUTHOR:**

**PUBLICATION:** DOE/LLW-76T

**DATE:** December 1988

**ORGANIZATION:**

**SUMMARY:** Waste volume minimized during solidification process. Water content of liquid waste streams is removed during the solidification process.

**Waste Form:** Bitumen

**Waste Type:** Various (summary of systems) compatible with most low-level radioactive waste streams.

**Waste Loading:** High waste loading capability

**Development Status:** --

**Compressive Strength:** Documented results (ASTM D-1074) show a range of compressive strengths between 55 and 300 psi with oxidized bitumen generally higher.

**Thermal Stability:** Test Method: ASTM B553. Data supported conclusion that thermal cycling has no effect on bituminized waste forms.

**Radiation Stability:** Samples exposed to total dose of  $10^8$  rad IAW NRC tests. All samples stabilized using bitumen performed well during radiation testing. Post-test compressive strengths were unchanged compared to pretest. Radiation has little or no effect on waste stabilized in bitumen.

**Biological Stability:** Initial biodegradation tests detected bacterial growth on some test samples. Bitumen vendors, IAW, the NRC BTP, are conducting long-term tests to determine effects of growth on waste form stability.

**Leach Resistance:** Test Method: ANS 16-1. Typical wastes with 60 wt% waste had range of leach indices from 8 to 14.



**ORGANIC BINDER SUMMARIES****NO. 2 (cont.)**

|                                   |  |
|-----------------------------------|--|
| <b>Immersion Stability:</b>       | Bitumen waste forms containing certain waste streams may crack or swell when exposed to water. Immersion can have a severe impact on compressive strength of waste forms. On the average, the test showed a decrease in compressive strength of 10-50%. The loss of compressive strength was directly proportional to increase in the amount of waste loading. |
| <b>Free Liquid:</b>               | None   |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | Some types of bitumen can be ignited at temperatures as low as 315° C  |
| <b>Conclusions:</b>               | Bitumen's performance in the tests prescribed in the NRC BTP indicates that bitumen is a good stabilization agent for most low-level waste streams. Its potential application to wastes other than those reported in the reviewed topical reports, including aqueous mixed, or hazardous wastes, must be evaluated on a case-by case basis.                    |

## ORGANIC BINDER SUMMARIES

NO. 3

**TITLE:** Treatment of Spent Ion-Exchange Resins for Storage and Disposal

**AUTHOR:**

**PUBLICATION:** Technical Report Series No. 254

**DATE:** 1985

**ORGANIZATION:** International Atomic Energy Agency

**SUMMARY:** For a 55 wt% loading of ion-exchange resins, volumes are 0.75 times the initial volume of ion-exchange resin. For ion-exchange ash in bitumen, final volumes are less than initial volumes.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Bitumen (direct-distilled, oxidized, cracked, or emulsified)  |
| <b>Waste Type:</b>                | Spent ion-exchange resins   |
| <b>Waste Loading:</b>             | --  |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | Bitumen can be successfully used for low and medium activity when the radiation dose does not exceed $10^7$ Gy. Some $H_2$ generation.  |
| <b>Biological Stability:</b>      | Any bacteriological attack can be regarded as generally insignificant.  |
| <b>Leach Resistance:</b>          | Leach rates generally increase with harder bitumens, increased waste loading, and residual water in the waste form.   |
| <b>Immersion Stability:</b>       | Ion-exchange resins will absorb water with time once they are disposed of, although bitumen will impede this process. The swelling could lead to damage to the waste form repository. |
| <b>Chemical Durability:</b>       | --  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | --  |

## ORGANIC BINDER SUMMARIES

NO. 4

TITLE: A Waste Inventory Report for Reactor and Fuel Fabrication Facility Wastes

AUTHOR:

PUBLICATION: ONWI-20

DATE: 1979

ORGANIZATION: ONWI Library

SUMMARY: Description of the physical and chemical characteristics of waste forms generated at light-water-cooled nuclear reactor power plants and nuclear fuel fabrication facilities.

|                            |   |
|----------------------------|---|
| Waste Form:                | Urea formaldehyde   |
| Waste Type:                | Liquid waste  |
| Waste Loading:             | 1:1   |
| Development Status:        | Commercial  |
| Compressive Strength:      | 50 kg/cm <sup>2</sup>   |
| Thermal Stability:         | --  |
| Radiation Stability:       | Moderate damage at 3 x 10 <sup>6</sup> rad; severe damage at 2 x 10 <sup>7</sup> rad  |
| Biological Stability:      | Susceptible to biodegradation   |
| Leach Resistance:          | Poor for <sup>85</sup> Sr and <sup>137</sup> Cs   |
| Immersion Stability:       | --  |
| Free Liquids:              | Produced during polymerization as low as 1%   |
| Chemical Durability:       | Resistant to oils, solvents, and grease, but dissolves in strong alkalis and acids  |
| Compositional Flexibility: | Good in sealed system, water loss when exposed to air   |
| Gas Generation:            | Hydrogen gas produced during radiolysis   |
| RCRA Compliance:           | Self-extinguishing  |
| Conclusions:               | Means for the stabilization of low-level waste containing free liquids are needed to minimize the potential release of radionuclides during handling, offsite shipping, and disposal. |

## ORGANIC BINDER SUMMARIES

NO. 5

TITLE: The Options for Solidifying Low Level Waste

AUTHOR: M.J. Akins; G. Costomiris; P. Delete; R.B. Wilson

PUBLICATION: Nuclear Engineering International, Vol. 28

DATE: March 1983

ORGANIZATION:

SUMMARY: Solidification of resin in bitumen results in volume reduction. Ash in bitumen and evaporator concentrates in bitumen do not result in volume reduction.

Waste Form: Bitumen

Waste Type: Various (summary of studies)

Waste Loading: --

Development Status: --

Compressive Strength: Bitumen at ambient temperature exhibits good mechanical strength. On impact, the material will deform but will not normally break apart. The material will flow at high temperatures.

Thermal Stability: Bitumen has demonstrated good thermal stability, but it is subject to melting/flowing at elevated temperatures which could be reached if exposed to fire or very hot sun-conditions.

Radiation Stability: Bitumen can be successfully used for solidification of low and intermediate level wastes when integrated radiation dose does not exceed  $\sim 10^9$  rad. As the dose approaches  $10^9$  rad, the chemical structure breaks down, leading to gas generation and a volume increase.

Biological Stability: Based on the amount of organic constituents that leach from test specimens, it was determined that bitumen is not likely to be subject to significant biological attack.

Leach Resistance: Some test results indicate bitumen leach rates are an order of magnitude lower than cements. However, if catastrophic swelling occurs as a result of immersion, the leach resistance is negligible.

Immersion Stability: --

Free Liquids: None

Chemical Durability: --

**ORGANIC BINDER SUMMARIES****NO. 5 (cont.)**

- Compositional Flexibility:** Has low sensitivity to variations on waste feed and waste to binder ratio, but oil in feed must be <1%.
- Gas Generation:** --
- RCRA Compliance:** Ignition temperature for pure bitumen is ~600°F, and 750 to 900°F when mixed with radioactive waste.
- Conclusions:** Bitumen has some advantages and some disadvantages when compared to other solidification agents. Each system must be evaluated on its merits in conjunction with all the specific parameters of its intended use.

## ORGANIC BINDER SUMMARIES

NO. 6

TITLE: Biodegradation Testing of Bitumen

AUTHOR: P. Barletta; B. Bowerman; R. Davis; C. Shea

PUBLICATION: BNL-NUREG-38999

DATE: 1986

ORGANIZATION: Brookhaven National Laboratory

## SUMMARY:

|                            |   |
|----------------------------|---|
| Waste Form:                | Bitumen (oxidized)  |
| Waste Type:                | N/A, the bitumen was tested without loading   |
| Waste Loading:             | --  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | Used Bartha-Pramer test method. See summary for results.  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Environmental factors have very little effect on the rate of biodegradation of bitumen. The mean rate is on the order of $5.5 \times 10^{-4}$ cm/yr. Summary of other studies indicates microbial attack is more likely under aerobic conditions, and that many fungal and bacterial strains can degrade bitumen. |

## ORGANIC BINDER SUMMARIES

NO. 7

**TITLE:** Conditioning Options for Magnox Fuel Element Debris  
**AUTHOR:** B. Bennet; et al.  
**PUBLICATION:** Radioactive Waste Management  
**DATE:** 1984  
**ORGANIZATION:** British Nuclear Energy Society, London  
**SUMMARY:** Work carried out on processes for the treatment of fuel element debris arising at Magnox nuclear power plants.

**Waste Form:** Epoxy resin  
**Waste Type:** Magnox fuel element debris  
**Waste Loading:** 40 vol%  
**Development Status:** Commercial  
**Compressive Strength:** 19 MPa  $\pm$  1  
**Thermal Stability:** Reduction of compressive strength to 17 MPa  $\pm$  1 after 20 freeze/thaw cycles  
**Radiation Stability:** Increased compressive strength to 20 MPa  $\pm$  1 after irradiation to 1,000 Mrads  
**Biological Stability:** --  
**Leach Resistance:** For  $^{137}\text{Cs}$ : 7 days =  $1.0 \times 10^{-4}$  cm/day; 100 days =  $1.2 \times 10^{-5}$  cm/day  
**Immersion Stability:** Decreased compressive strength to 17 MPa  $\pm$  1 after immersion inversion  
**Free Liquids:** --  
**Chemical Durability:** --  
**Compositional Flexibility:** --  
**Gas Generation:** 90°C for 24 hours: 0.7 mL/kg,  
**RCRA Compliance:** Spontaneous ignition at 600°  
**Conclusions:** Epoxy resins exhibit superior chemical stability and leach resistance.

## ORGANIC BINDER SUMMARIES

NO. 8

TITLE: Microbial Activity on Bituminized Radioactive Waste

AUTHOR: L. Buckley; B. Clegg; W. Oldham

PUBLICATION: Radioactive Waste Management and the Nuclear Fuel Cycle, Vol. 6

DATE: March 1985

## ORGANIZATION:

## SUMMARY:

|                            |   |
|----------------------------|---|
| Waste Form:                | Bitumen   |
| Waste Type:                | Nitrate salts   |
| Waste Loading:             | 38 wt%  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability:         | --  |
| Radiation Stability:       | --  |
| Biological Stability:      | Exposed to pseudomonas, bacillus, and other bacteria chosen for their affinity for bitumen.   |
| Leach Resistance:          | Test Method: "Leach Testing of Immobilized Radioactive Waste Solids, A Proposal for a Standard Method," E. Hesse, 1971. There were no statistical differences in the releases from the specimens inoculated with the culture, and those which were not. |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | Statistically, there appeared to be no effect of microbial action on the bituminized waste forms.   |



## ORGANIC BINDER SUMMARIES

NO. 9

**TITLE:** Properties of Radioactive Wastes and Waste Containers-First Topical Report

**AUTHOR:** P. Colombo; R.M. Neilson, Jr.

**PUBLICATION:** BNL-NUREG-50957

**DATE:** 1979

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Properties of waste forms and packages, solidification agents, and various processing parameters are discussed.

**Waste Form:** Urea formaldehyde

**Waste Type:** Bead resin waste slurry

**Waste Loading:** Waste/UF = 2.6

**Development Status:** Commercial

**Compressive Strength:** ASTM C39-72: 78 psi  $\pm$  5

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** High concentration of organic carbon in free standing water

**Leach Resistance:** IAEA Leach Test:  $^{137}\text{Cs} = 1.4 \times 10^{-6}$  cm/day;  $^{85}\text{Sr} = 2.4 \times 10^{-8}$  cm/day

**Immersion Stability:** --

**Free Liquids:** Often formed during polymerization

**Chemical Durability:** --

**Compositional Flexibility:** Weight loss due to water evaporation in ambient air

**Gas Generation:** .915 cm<sup>3</sup>/g from 10<sup>7</sup> R at 104 R/hr; .756 cm<sup>3</sup>/g from 10<sup>7</sup> R at 105 R/hr

**RCRA Compliance:** Flammability: ASTM D635-74; self-extinguishing

**Conclusions:** This information is required to assess the safety and environmental impacts of radioactive waste disposal and to support quality control and assurance standards for radioactive waste treatment systems in operating facilities.

## ORGANIC BINDER SUMMARIES

NO. 10

**TITLE:** Properties of Radioactive Wastes and Waste Containers

**AUTHOR:** P. Colombo; R.M. Nielson, Jr.

**PUBLICATION:** BNL-NUREG-50957

**DATE:** 1977

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** The properties of waste forms and packages resulting from the solidification of liquid concentrate and solid wastes generated as by-products of the liquid radioactive treatment systems in commercial BWRs and PWRs have been determined.

**Waste Form:** Vinyl ester-styrene

**Waste Type:** BWR chemical regenerative waste

**Waste Loading:** Waste/binder = 1.65

**Development Status:** Commercial

**Compressive Strength:** ASTM C39-72: 2140 lb/in.<sup>2</sup>

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** After four days immersion, small amounts of organic carbon was detected.

**Leach Resistance:** Cumulative fraction release: Cs = 0.0548; Sr = 0.0513; Co = 0.155

**Immersion Stability:** --

**Free Liquids:** None

**Chemical Durability:** Non-corrosive

**Compositional Flexibility:** Weight loss due to evaporation of water

**Gas Generation:** 21 molecules/100 eV at an exposure of 10<sup>5</sup> rad

**RCRA Compliance:** --

**Conclusions:** Vinyl ester-styrene compares favorably with other types of solidifying agents.

## ORGANIC BINDER SUMMARIES

NO. 11

TITLE: Critical Review of the Properties of Solidified Radioactive Waste

AUTHOR: P. Colombo; R.M. Neilson, Jr.

PUBLICATION: BNL-NUREG-50591

DATE: 1976

ORGANIZATION: Brookhaven National Laboratory

SUMMARY: Review and evaluation of existing information on the properties of various solidified wastes and waste packages being generated by commercial nuclear power reactors.

|                            |  |      |        |
|----------------------------|--|------|--------|
| Waste Form:                | Urea formaldehyde  |      |        |
| Waste Type and Loading:    | Deep bed demineralizer resins  | 2.2% |        |
|                            | Powdered demineralizer resins  | 2.6% | Weight |
|                            | 25 wt% sodium sulfate  | 1.6% | Ratio  |
|                            | Borates  | 1.5% | to     |
|                            | Diatomaceous earth   | 2.7% | UF     |
| Development Status:        | Commercial   |      |        |
| Compressive Strength:      | --   |      |        |
| Thermal Stability:         | Poor due to high water content   |      |        |
| Radiation Stability:       | Mild damage at $3 \times 10^6$ rad; moderate to severe damage at $2 \times 10^7$ rad                   |      |        |
| Biological Stability:      | --   |      |        |
| Leach Resistance:          | Cs - $3 \times 10^{-2}$ to $2 \times 10^{-1}$ fractional per day                                       |      |        |
| Immersion Stability:       | --   |      |        |
| Free Liquids:              | .5% to 1% by volume, slightly acidic   |      |        |
| Chemical Durability:       | Resists chemical attack  |      |        |
| Compositional Flexibility: | Prolonged drying leads to degradation of strength and increased leachability.                          |      |        |
| Gas Generation:            | Low toxicity gases produced (.5 molecules/100 eV)  |      |        |
| RCRA Compliance:           | Self extinguishing   |      |        |
| Conclusions:               | Conclusions are drawn regarding the current state of knowledge of solidified reactor waste properties. |      |        |

## ORGANIC BINDER SUMMARIES

NO. 12

TITLE: Accelerated Leach Test(s) Program Annual Report

AUTHOR: D.R. Dougherty; R.F. Pietrzak; M. Fuhrmann; P. Colombo

PUBLICATION: BNL-52042

DATE 1986

ORGANIZATION: Brookhaven National Laboratory

SUMMARY: Leachability evaluation of various solidification/waste form combination.

Waste Form: Bitumen

Waste Type: Sodium tetraborate (produced by neutralizing boric acid with sodium hydroxide)

Waste Loading: 40 wt%

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: --

Biological Stability: --

Leach Resistance: 129 days: conductance ( $\mu\text{mhos/cm}$ ) = 1630

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: For waste form samples containing 40 wt% salt, the cumulative fraction released was 450 times greater than the baseline after 129 days. This is due to the number of interconnected salt particles providing a pathway for water.

## ORGANIC BINDER SUMMARIES

NO. 13

**TITLE:** Chemical and Environmental Properties of Plastics and Elastomers

**AUTHOR:** O.H. Fenner

**PUBLICATION:** Handbook of Plastics and Elastomers, C.A. Harper, ed.

**DATE:** 1975

**ORGANIZATION:** McGraw-Hill, New York

**SUMMARY:** Provides a listing of physical properties and characteristics for various plastics and elastomers.

**Waste Form:** Polyethylene

**Waste Type:** --

**Waste Loading:** --

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** Unprotected polyethylene undergoes rapid ultraviolet degradation.

**Gas Generation:** --

**RCRA Compliance:** If ignited, polyethylene burns in a slow controlled manner at about 1 in./min. under ASTM D-635 conditions.

**Conclusions:** Control of environmental exposure is critical to the integrity of polyethylene.

## ORGANIC BINDER SUMMARIES

NO. 14

TITLE: Vinyl Ester Solidification of Low-Level Radioactive Wastes

AUTHOR: H.E. Filter

PUBLICATION:

DATE: 1979

ORGANIZATION: Dow Chemical Company

## SUMMARY:

Waste Form: Vinyl Ester-Styrene  
Waste Type: --  
Waste Loading: --  
Development Status: Commercial  
Compressive Strength: --  
Thermal Stability: --  
Radiation Stability: Low exposures have no significant effect on leachability. Exposures of  $4 \times 10^8$  and  $6 \times 10^8$  rad as much as tripled the leachability; however, this was still below a cumulative release of 5% in 100 days. Similar doses had no effect on the leachability of  $^{60}\text{Co}$ .  
Biological Stability: --  
Leach Resistance: Prior heating to  $538^\circ\text{C}$  for ten minutes resulted in an acceleration of initial leaching of  $^{137}\text{Cs}$ ; however, the overall cumulative fraction leached began to approach that of the control after 90 days.  
Immersion Stability: --  
Free Liquids: --  
Chemical Durability: --  
Compositional Flexibility: --  
Gas Generation: --  
RCRA Compliance: --  
Conclusions: The Dow Polymer Solidification Process has advantages of low leachability and radiation stability.

## ORGANIC BINDER SUMMARIES

NO. 15

**TITLE:** The Feasibility of Incorporating Radioactive Wastes in Asphalt or Polyethylene

**AUTHOR:** C.L. Fitzgerald; H.W. Godbee; R. E. Blanco

**PUBLICATION:** Nuclear Applications and Technology, Vol. 9, No. 12

**DATE:** 1970

**ORGANIZATION:** Oak Ridge National Laboratory

**SUMMARY:** Thermoplastic of chemical formula  $(CH_2-CH_2)^x$  is melted and mixed with waste to form homogeneous solid (solid plastic monolith).

**Waste Type:** Sodium borate, organic and inorganic solvents, low-density polyethylene lab-scale

**Waste Form:** --

**Waste Loading:** 40 wt% evaporator concentrates, 50 wt% tributyl-phosphate

**Development Status:** Experimental, limited application (at time of publication)

**Compressive Strength:** 300 kg/cm<sup>2</sup>

**Thermal Stability:** --

**Radiation Stability:** 10<sup>6</sup> rad, no effect; 10<sup>7</sup> - 10<sup>9</sup> rad caused slight shrinkage and discoloration

**Biological Stability:** --

**Leach Resistance:** Depends on waste type and loading, polyethylene type. Ranged from 3 x 10<sup>-3</sup> g/cm<sup>2</sup> day to 1.5 x 10<sup>-6</sup> g/cm<sup>2</sup> day.

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** No gas was generated.

**RCRA Compliance:** --

**Conclusions:** Discusses feasibility of polyethylene for waste encapsulation. Limitations of their results can be attributed to the fact that only two types (of the hundreds available) of polyethylene were used, and processing technology was not optimized (only thin film evaporator used, which is not common practice in plastics industry).

## ORGANIC BINDER SUMMARIES

NO. 16

TITLE: Waste Form Evaluation Program, Final Report

AUTHOR: E.M. Franz; P. Colombo

PUBLICATION: BNL-51954

DATE: September 1984

ORGANIZATION: Brookhaven National Laboratory

SUMMARY: Presents data for assessing the acceptability of polyethylene waste forms to meet the requirements of 10 CFR 61.

Waste Form: Polyethylene

Waste Type: Sodium sulfate

Waste Loading: 50 wt%

Development Status: Commercial

Compressive Strength: ASTM D-1074: 2000 psi  $\pm$  100

Thermal Stability: ASTM B-553

Radiation Stability:  $^{60}\text{Co}$  gamma radiation to  $10^8$  rad at  $3.6 \times 10^6$  rad/hr then compressive testing ASTM D-1074

Biological Stability: ASTM G21 and G22

Leach Resistance: ANS 16.1: leach index  $^{85}\text{Sr}=10.2$ ,  $^{137}\text{Cs}=9.9$ ,  $^{60}\text{Co}=10.1$

Immersion Stability: 54 wt%: 0.0% swelling after 90 days immersion in water

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: The major criterion the waste form must meet after exposure (immersion, irradiation, etc.) is a compressive strength  $\geq$  50 psi.



**ORGANIC BINDER SUMMARIES**

NO. 17

**TITLE:** Waste Form Evaluation Program, Final Report

**AUTHOR:** E.M. Franz; P. Colombo

**PUBLICATION:** BNL-51954

**DATE:** September 1984

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Presents data for assessing the acceptability of polyethylene waste forms to meet the requirements of 10 CFR 61.

**Waste Form:** Polyethylene

**Waste Type:** Boric acid

**Waste Loading:** 35 wt%

**Development Status:** Commercial

**Compressive Strength:** ASTM D-1074: 1800 psi  $\pm$  100

**Thermal Stability:** ASTM B-553

**Radiation Stability:**  $^{60}\text{Co}$  gamma radiation to  $10^8$  rad at  $3.6 \times 10^6$  rad/hr then compressive testing ASTM D-1074

**Biological Stability:** ASTM G21 and G22

**Leach Resistance:** --

**Immersion Stability:** 0.2% swelling after 90 days immersion in water

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** The major criterion the waste form must meet after exposure (immersion, irradiation, etc.) is a compressive strength  $\geq$  50 psi.

## ORGANIC BINDER SUMMARIES

NO. 18

TITLE: Waste Form Evaluation Program, Final Report

AUTHOR: E.M. Franz; P. Colombo

PUBLICATION: BNL-51954

DATE: September 1984

ORGANIZATION: Brookhaven National Laboratory

SUMMARY: Presents data for assessing the acceptability of polyethylene waste forms to meet the requirements of 10 CFR 61.

Waste Form: Polyethylene

Waste Type: Incinerator ash

Waste Loading: 20 wt%

Development Status: Commercial

Compressive Strength: ASTM D-1074: 1750 psi  $\pm$  140

Thermal Stability: ASTM B-553

Radiation Stability:  $^{60}\text{Co}$  gamma radiation to  $10^8$  rad at  $3.6 \times 10^6$  rad/hr, then compressive testing ASTM D-1074

Biological Stability: ASTM G21 and G22

Leach Resistance: ANS 16.1 (25 wt%) : leach index  $^{85}\text{Sr}$ =15.5,  $^{137}\text{Cs}$ =12.5,  $^{60}\text{Co}$ =13.9

Immersion Stability: 25 wt%: 0.5% swelling after 90 days immersion in water

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: The major criterion the waste form must meet after exposure (immersion, irradiation, etc.) is a compressive strength  $\geq$  50 psi.

**ORGANIC BINDER SUMMARIES**

NO. 19

**TITLE:** Waste Form Evaluation Program, Final Report

**AUTHOR :** E.M. Franz; P. Colombo

**PUBLICATION:** BNL-51954

**DATE:** September 1984

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Presents data for assessing the acceptability of polyethylene waste forms to meet the requirements of 10 CFR 61.

**Waste Form:** Polyethylene

**Waste Type:** Ion exchange resin

**Waste Loading:** 10 wt%

**Development Status:** Commercial

**Compressive Strength:** ASTM D-1074: 2100 psi  $\pm$  170

**Thermal Stability:** ASTM B-553

**Radiation Stability:**  $^{60}\text{Co}$  gamma radiation to  $10^8$  rad at  $3.6 \times 10^6$  rad/hr, then compressive testing ASTM D-1074

**Biological Stability:** ASTM G21 and G22

**Leach Resistance:** ANS 16.1 : leach index  $^{85}\text{Sr}=16.2$ ,  $^{137}\text{Cs}=18.2$ ,  $^{60}\text{Co}=13.6$

**Immersion Stability:** 20 wt%: 0.0% swelling after 90 days immersion in water

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** The major criterion the waste form must meet after exposure (immersion, irradiation, etc.) is a compressive strength  $\geq$  50 psi.

## ORGANIC BINDER SUMMARIES

NO. 20

TITLE: Solidification of Problem Wastes, Annual Progress Report

AUTHOR: E.M. Franz; J.H. Heiser; P. Colombo

PUBLICATION: BNL-52078

DATE: 1987

ORGANIZATION: Brookhaven National Laboratory

SUMMARY: Development and evaluation of solidification systems for sodium nitrate waste and compacted waste.

|                            |  |                            |
|----------------------------|--|----------------------------|
| Waste Form:                | Polyethylene   |                            |
| Waste Type:                | Sodium nitrate   |                            |
| Waste Loading:             | 30-70%   |                            |
| Development Status:        | Laboratory scale   |                            |
| Compressive Strength:      | ASTM D-695:  | Compressive Strength (psi) |
|                            | 30 wt%   | 2370                       |
|                            | 50 wt%   | 1920                       |
|                            | 60 wt%   | 2200                       |
|                            | 70 wt%   | 1020                       |
| Thermal Stability:         | --   |                            |
| Radiation Stability:       | --   |                            |
| Biological Stability:      | --   |                            |
| Leach Resistance:          | ANS 16.1:  | Leach Index                |
|                            | 30 wt%   | 11.1                       |
|                            | 50 wt%   | 9.7                        |
|                            | 60 wt%   | 9.0                        |
|                            | 70 wt%   | 7.8                        |
| Immersion Stability:       | 90 days in water:  | Compressive Strength (psi) |
|                            | 30 wt%   | 2550                       |
|                            | 50 wt%   | 1920                       |
|                            | 60 wt%   | 2310                       |
|                            | 70 wt%   | 720                        |
| Free Liquids:              | --   |                            |
| Chemical Durability:       | --   |                            |
| Compositional Flexibility: | --   |                            |
| Gas Generation:            | --   |                            |
| RCRA Compliance:           | --   |                            |
| Conclusions:               | Polyethylene waste forms exhibit good leaching characteristics, high compressive strength (700-2,600 psi) and high leaching efficiencies (70 wt% sodium nitrate) |                            |

**ORGANIC BINDER SUMMARIES**

NO. 21

**TITLE:** A Survey of Agents and Techniques Applicable to the Solidification of LLW

**AUTHOR:** M. Fuhrmann; R.M. Neilson; P. Colombo

**PUBLICATION:** BNL 51521

**DATE:** December 1981

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Use of a twin screw extruder reported at Japan Atomic Energy Research Institute, and in the Netherlands .

**Waste Stream:** Ion exchange resin beads

**Waste Form:** Low-density polyethylene

**Waste Loading:** 50 wt% bead resin

**Development Status:** Experimental, bench-scale

**Compressive Strength:** 300 kg/cm<sup>2</sup>

**Thermal Stability:** --

**Radiation Stability:** 33% drop in compressive strength reported after gamma irradiation dose of 10<sup>8</sup> rad.

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** Polyethylene by itself does not spontaneously ignite on heating to temperatures of 550°C.

**Conclusions:** This is a literature study comparing solidification agents applicable to LLW. Report is a comprehensive overview of many solidification agents including hydraulic cements, bitumen, polyethylene, urea-formaldehyde, polyester and epoxy thermosetting polymers, mineralization processes, glass, polymer modified gypsum cement, and polymer impregnated concrete. Numerous figures and tables reprinted from original sources enhance readability.

## ORGANIC BINDER SUMMARIES

NO. 22

TITLE: The Chemical Cleaning of Dresden Unit I

AUTHOR: J.J. Graves

PUBLICATION: Transactions of the American Nuclear Society, Vol. 28

DATE: 1978

ORGANIZATION:

SUMMARY: Discussion of Dow Solidification system use in decontaminating a nuclear power reactor.

Waste Form: Dow polymer

Waste Type: Various wastes produced during decontamination of nuclear power plants

Waste Loading: --

Development Status: Commercial

Compressive Strength: --

Thermal Stability: --

Radiation Stability: Exposure to 600 Mrad has no significant effect on mechanical properties or leachability.

Biological Stability: --

Leach Resistance: Low leachability

Immersion Stability: --

Free Liquid: No free liquid

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: The Dow solidification process is superior to any other commercial solidification process in waste form behavior and characteristics.

## ORGANIC BINDER SUMMARIES

NO. 23

**TITLE:** An Economic Analysis of a Volume Reduction/Polyethylene Solidification System for Low-Level Radioactive Wastes

**AUTHOR:** P.D. Kalb; P. Colombo

**PUBLICATION:** BNL-51866

**DATE:** January 1985

**ORGANIZATION:** Brookhaven National Lab

**SUMMARY:** Fluidized bed calcination/incineration as a waste pretreatment, coupled, with encapsulation in LDPE. Levelized revenue requirement technique used to compare volume reduction/polyethylene encapsulation with solidification of aqueous streams in cement.

**Waste Form:** Low-density polyethylene

**Waste Type:** Aqueous evaporator concentrates and ion exchange resins

**Waste Loading:** 60 wt% sodium sulfate, 30 wt% boric acid, and 25 wt% incinerator ash (conservative waste loadings selected intentionally)

**Development Status:** Study of economic feasibility for full-scale implementation

**Compressive Strength:** Not applicable

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** --

**Immersion Stability:** --

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** If current trends in escalation rates of cost components continue, the volume reduction/polyethylene solidification option will be cost effective for both boiling water and pressurized water reactors. Data indicate that a minimum net annual savings of \$0.8 million/yr (for a PWR with short shipping distance to disposal site) and a maximum savings of \$9 million/yr for a BWR with long shipping distance) can be achieved (savings in 1984 dollars).

## ORGANIC BINDER SUMMARIES

NO. 24

**TITLE:** Polyethylene Solidification of Low-Level Wastes

**AUTHOR:** P.D. Kalb; P. Colombo

**PUBLICATION:** BNL - 51867

**DATE:** 1984

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Low density polyethylene is melted at 130 - 150°C and mixed with waste to form homogeneous solid, using single screw extruder. Static hopper replaced with volumetric feeders to provide precise control of waste/binder ratios.

**Waste Form:** Low density polyethylene

**Waste Type:** Sodium sulfate, boric acid, incinerator ash, ion exchange resin beads

**Waste Loading:** 70 wt% sodium sulfate, 50 wt% boric acid, 40 wt% incinerator ash, 30 wt% ion exchange resin beads.

**Development Status:** Bench-scale (1.25-in.) single screw extruder

**Compressive Strength:** ASTM D-621 (deformation of plastics under 100 psi load)  
Maximum deformation when subjected to 100 psi load was 0.36%.

**Thermal Stability:** Thermal cycling conducted according to NRC protocol. After 30 cycles between -40 and +60°C, no change in waste form mechanical properties was detected.

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** ANS 16.1 90 day (<sup>137</sup>Cs, <sup>85</sup>Sr, <sup>60</sup>Co). Dependent on waste type and loading. Leaching from ash waste forms ranged between  $1 \times 10^{-3}$  and  $4 \times 10^{-4}$  for loadings up to 35 wt%. All 3 isotopes had similar releases from sodium sulfate waste forms (about  $5 \times 10^{-3}$  for 10 wt%) which increased with waste loading (about  $2 \times 10^{-2}$  50 wt%) for sodium sulfate.

**Immersion Stability:** 90 day water immersion testing conducted according to NRC protocol. Samples were then measured for any change in dimensions. Swelling was less than 2% for all samples, except ion exchange resin waste forms with loadings exceeding 30 wt%.

**Free Liquids:** --

**Chemical Durability:** --



**ORGANIC BINDER SUMMARIES**

NO. 24 (cont.)

Compositional Flexibility: --

Gas Generation: --

RCRA Compliance: --

Conclusions: This work examined many types of low density polyethylene (varying in density and melt index) to optimize the selection of a plastic with optimal properties for waste encapsulation. Process development study compared effectiveness of several processing technologies - single screw extrusion was found to be most effective and efficient. Feed system modified for improved QA. Detailed process control parameters (temperature zone setting, screw speeds, motor draw, melt temperatures and pressures) are all reported at length in the appendix.

## ORGANIC BINDER SUMMARIES

NO. 25

**TITLE:** Polyethylene Encapsulation of Nitrate Salt Wastes: Waste Form Stability, Process Scale-up, and Economics

**AUTHOR:** P.D. Kalb; J.H. Heiser; P. Colombo

**PUBLICATION:** BNL 52293

**DATE:** July 1991

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Low density polyethylene is melted at 130 - 150°C and mixed with waste to form homogeneous solid, using single screw extruder. Static hopper replaced with volumetric feeders to provide precise control of waste/binder ratios. This report assessed potential long-term durability of polyethylene waste forms by examining potential failure mechanisms including biodegradation, radiation, chemical attack, flammability, environmental stress cracking, and photodegradation. A review of the literature was supplemented by waste form performance testing for several key potential failure mechanisms including biodegradation, radiation and flammability.

**Waste Form:** Low-density polyethylene

**Waste Type:** Nitrate salts

**Waste Loading:** 70 dry wt% sodium nitrate

**Development Status:** Bench-scale: formulation development and performance testing complete. Full-scale: feasibility test at production-scale successful. Full-scale demonstration using surrogate waste planned for FY-1993.

**Compressive Strength:** ASTM D-695, Standard Method of Test for Compressive Properties of Rigid Plastics. Polyethylene waste form compressive yield strength varies with loading from a minimum of > 1,000 psi (70 wt% NaNO<sub>3</sub>) to a maximum of >2,350(30 wt% NaNO<sub>3</sub>); NRC minimum strength for waste forms other than cement is 60 psi. Compressive strength data for cores taken from full-scale test specimen were within 8% of lab-scale data.

**Thermal Stability:** --

## ORGANIC BINDER SUMMARIES

NO. 25 (cont.)

- Radiation Stability:** Low density polyethylene polymer chains become more heavily cross-linked when subjected to ionizing radiation, which increases mechanical strength and reduces permeability. Polyethylene waste forms subjected to  $10^8$  rad total absorbed dose of gamma irradiation increased in strength by about 18% over un-irradiated control samples.
- Biological Stability:** Results from literature which indicate that polyethylene is highly resistant to microbial degradation were confirmed by biodegradation testing on polyethylene waste forms (ASTM G-21 and G-22).
- Leach Resistance:** ANS 16.1 90 day, sodium leachability. Leaching of soluble salts is dependent on waste loading. Leaching index ranged from 11.1 for polyethylene waste forms containing 30 wt%  $\text{NaNO}_3$  to 7.8 for 70 wt%  $\text{NaNO}_3$ . These results represent 2 to 5 orders of magnitude lower leachability than NRC minimum leach index value of 6.0.
- Immersion Stability:** --
- Free Liquids:** --
- Chemical Durability:** --
- Compositional Flexibility:** --
- Gas Generation:** --
- RCRA Compliance:** Flammability testing indicated that polyethylene self-ignition occurs at  $430^\circ\text{C}$  (the material is self-extinguishing below this temperature) compared with normal process temperatures of  $<150^\circ\text{C}$ .
- Conclusions:** Following bench-scale development and testing, a scale-up feasibility test was conducted for simulated  $\text{NaNO}_3$  waste using a 4.5-in. production-scale extruder with a maximum output of 2000 lbs/hr. This test indicated that at least 60 wt%  $\text{NaNO}_3$  could be successfully encapsulated in polyethylene using production equipment, bench- and full-scale process data were comparable, and the waste form product was homogeneously mixed. QA/performance testing for cored specimens from the full-scale waste form including compressive strength, thermal cycling, irradiation and biodegradation testing were in close agreement with lab-scale data and were well above minimum NRC standards for low-level waste. An economic analysis for encapsulation of Rocky Flats nitrate salt resulted in net cost savings ranging between \$1.5 and 2.7 million, compared with conventional cement processes.

## ORGANIC BINDER SUMMARIES

NO. 26

TITLE: Characterization of Bituminous Intermediate-Level Waste Products

AUTHOR: Z. Kopajtic; D. Laske; H.P. Linder; et al.

PUBLICATION: Materials Research Society Symposium Proceedings, Vol. 21

DATE: 1989

ORGANIZATION: Materials Research Society Summary:

## SUMMARY:

Waste Form: 60 wt% distilled bitumen

Waste Type: Swiss intermediate level radioactive waste

Waste Loading: 40 wt% solids

Development Status: --

Compressive Strength: --

Thermal Stability: --

Radiation Stability: Gamma irradiated to approximately 5 MGy at approximately 3.8 kGy/hr

Biological Stability: --

Leach Resistance: Test Method: ISO-6961-1982 Long Term Leach Testing of Solidified Radioactive Waste Forms. Swelling and cracking observed, relatively low leach rates, although leach rates are slightly higher for irradiated forms.

Immersion Stability: --

Free Liquids: --

Chemical Durability: --

Compositional Flexibility: --

Gas Generation: As an irradiation effect, swelling and gas evolution (mainly H<sub>2</sub>)

RCRA Compliance: --

Conclusions: External gamma irradiation at a high dose rate caused intense swelling in both oxidized and distilled bituminous waste products. The amount of swelling depends on factors such as dose rate, total dose, type of bitumen, and quantity of the waste.

## ORGANIC BINDER SUMMARIES

NO. 27

**TITLE:** Characterization of Bituminous Intermediate-Level Waste Products

**AUTHOR:** Z. Kopajtic; D. Laske; H.P. Linder; et al.

**PUBLICATION:** Materials Research Society Symposium Proceedings, Vol. 21

**DATE:** 1989

**ORGANIZATION:** Materials Research Society

**SUMMARY:**

**Waste Form:** 63 wt% blown bitumen  
**Waste Type:** Simulated inactive  $\text{NaNO}_3$   
**Waste Loading:** 37 wt%  
**Development Status:** --  
**Compressive Strength:** --  
**Thermal Stability:** --  
**Radiation Stability:** Gamma irradiated to approximately 5 MGy at approximately 3.8 kGy/hr  
**Biological Stability:** --  
**Leach Resistance:** Test Method: ISO-6961-1982 Long Term Leach Testing of Solidified Radioactive Waste Forms. Swelling and cracking observed, relatively low leach rates, although leach rates are slightly higher for irradiated forms.  
**Immersion Stability:** --  
**Free Liquids:** --  
**Chemical Durability:** --  
**Compositional Flexibility:** --  
**Gas Generation:** As an irradiation effect, swelling and gas evolution (mainly  $\text{H}_2$ )  
**RCRA Compliance:** --  
**Conclusions:** External gamma irradiation at a high dose rate caused intense swelling in both oxidized and distilled bituminous waste products. The amount of swelling depends on factors such as dose rate, total dose, type of bitumen, and quantity of the waste.

## ORGANIC BINDER SUMMARIES

## TITLE:

AUTHOR: F.A. Makhlis

PUBLICATION: Radiation Physics and Chemistry of Polymers

DATE: 1975

ORGANIZATION: John Wiley and Sons, New York

SUMMARY: Influence of the physical state and temperature on radiochemical transformations in polymers.

|                            |   |
|----------------------------|---|
| Waste Form:                | Polyethylene  |
| Waste Type:                | --  |
| Waste Loading:             | --  |
| Development Status:        | --  |
| Compressive Strength:      | --  |
| Thermal Stability :        | --  |
| Radiation Stability:       | Observed changes in melting point and gel content depend only on total dose and are not affected by the rate the dose is delivered. |
| Biological Stability:      | --  |
| Leach Resistance:          | --  |
| Immersion Stability:       | --  |
| Free Liquids:              | --  |
| Chemical Durability:       | --  |
| Compositional Flexibility: | --  |
| Gas Generation:            | --  |
| RCRA Compliance:           | --  |
| Conclusions:               | The use of polymers in atomic technology is limited due to their relatively low radiation stability.                                |

## ORGANIC BINDER SUMMARIES

NO. 29

**TITLE:** Laboratory Performance Testing of an Extruded Bitumen Containing a Surrogate, Sodium Nitrate-Based, Low-Level Aqueous Waste

**AUTHORS:** A.J. Mattus; M.M. Kaczmarzsky

**PUBLICATION:** Proceedings of the Symposium on Waste Management at Tucson, Arizona

**DATE:** March 1-5, 1987

**ORGANIZATION:**

**SUMMARY:** Always results in volume reduction for wastes which contain volatile solvents.

**Waste Form:** Bitumen (ASTM D-312) Type III (oxidized)

**Waste Type:** Surrogate low-level mixed liquid waste containing ~30 wt% sodium nitrate, plus eight heavy metals, cold Cs and Sr.

**Waste Loading:** Waste Loading of 40 wt%, 50 wt%, and 60 wt%

**Development Status:** --

**Compressive Strength:** Test Method ASTM D-1074. Compressive strengths range from ~249 psi for pure bitumen to ~622 for the 60 wt% loading.

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Test Method ANS 16. 1. Leach indices for Cs, Sr, as well as other heavy metals range from 8.0 to >14.1 for all samples.

**Immersion Stability:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** Test Method: EP Tox. All waste forms/waste loading easily passed the test. For most of the 8 EPA metals, the concentration in the leachates was at/near the LLD.

**Conclusions:** Regulatory performance testing of extruded bitumen has shown that the relatively viscous form of oxidized bitumen used performed well under several required performance tests. The extruder bitumen process has been able to achieve high waste loading and high VRE, and has still resulted in a waste form capable of offering superior leach resistance.

## ORGANIC BINDER SUMMARIES

NO. 30

**TITLE:** Development of Radioactive Waste Management at the Japan Atomic Energy Research Institute

**AUTHOR:** I.S. Miyango; A.I. Sakata; H. Amano

**PUBLICATION:** Nuclear Power and Its Fuel Cycle, Vol. 4, IAEA CN-36/156

**DATE:** 1977

**ORGANIZATION:** Japan Atomic Energy Research Institute

**SUMMARY:** Discussion of waste forms and their process considerations.

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Low-density polyethylene  |
| <b>Waste Type:</b>                | Ion exchange resin beads  |
| <b>Waste Loading:</b>             | 50 wt% bead resin   |
| <b>Development Status:</b>        | Bench-scale   |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | --  |
| <b>Radiation Stability:</b>       | --  |
| <b>Leach Resistance:</b>          | Release of $^{137}\text{Cs}$ was as low as 0.1% after one year. |
| <b>Immersion Stability:</b>       | --  |
| <b>Free Liquids:</b>              | --  |
| <b>Chemical Durability:</b>       | --  |
| <b>Compositional Flexibility:</b> | --  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | --  |



## ORGANIC BINDER SUMMARIES

NO. 31

**TITLE:** Solidification of Powdered Ion Exchange Resins with Polyethylene

**AUTHOR:** N. Moriyama; S. Dojiri; T. Honda

**PUBLICATION:** Nuclear and Chemical Waste Management, Vol. 3, No. 2

**DATE:** 1982

**ORGANIZATION:** Japan Atomic Energy Research Institute

**SUMMARY:** Low density polyethylene is melted at 160°C and mixed with waste to form homogeneous solid, using batch wiped film stirrer.

**Waste Form:** Low-density polyethylene

**Waste Type:** Powdered ion exchange resins

**Waste Loading:** 40 - 50 wt% powdered ion exchange resins

**Development Status:** Experimental (lab-scale), limited application at time of publication

**Compressive Strength:** ASTM D-6952 test. 31- 237 kg/cm<sup>2</sup>

**Thermal Stability:** --

**Radiation Stability:** Radiation resistance testing showed slight increases in strength with doses ranging from 10<sup>7</sup> - 10<sup>9</sup> rad and 50 wt% powdered resin.

**Biological Stability:** --

**Leach Resistance:** IAEA leach test. Fraction of <sup>60</sup>Co released in 200 days: 3.1 x 10<sup>-4</sup>. <sup>137</sup>Cs fraction release below detection limit of 3 x 10<sup>-5</sup>. <sup>60</sup>Co diffusion coefficient: 4.5 x 10<sup>-10</sup> cm<sup>2</sup>/day for initial 64 days, 6.7 x 10<sup>-12</sup> cm<sup>2</sup>/day for remaining data.

**Immersion Stability:** Water immersion for 1200 days caused some swelling and cracking in samples containing waste loadings of more than 40 wt%.

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** Minimal gas generation measured: 2 x 10<sup>-2</sup> cm<sup>3</sup>/g Mrad.

**RCRA Compliance:** --

**Conclusions:** --

## ORGANIC BINDER SUMMARIES

NO. 32

**TITLE:** Incorporation of an Evaporator Concentrate on Polyethylene for BWR

**AUTHOR:** N. Moriyama

**PUBLICATION:** Nuclear Chemical Waste Management

**DATE:** 1982

**ORGANIZATION:**

**SUMMARY:** The adaptability of polyethylene solidification method to an evaporator concentrate produce on a boiling water reactor (BWR).

**Waste Form:** Polyethylene

**Waste Type:** Sodium sulfate

**Waste Loading:** 40 to 70 wt%

**Development Status:** --

**Compressive Strength:** ASTM D-695 and D-256

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** 40-50 wt%: diffusion coefficient of sodium range from  $10^{-7}$  to  $10^{-6}$  cm<sup>2</sup>/day

**Immersion Stability:** No swelling, destruction or deterioration after 120 days

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Polyethylene products are superior to both cement and bitumen products with respect to mechanical property, leachability, water resistance, and volume reduction effect.

## ORGANIC BINDER SUMMARIES

NO. 33

**TITLE:** Solidification of Powdered Ion Exchange Resins with Polyethylene

**AUTHOR:** N. Moriyama

**PUBLICATION:** Nuclear Chemical Waste Management

**DATE:** 1982

**ORGANIZATION:**

**SUMMARY:** The adaptability of polyethylene solidification method to spent powdered ion exchange resins.

**Waste Form:** Polyethylene

**Waste Type:** Spent powdered ion exchange resin

**Waste Loading:** 50 wt%

**Development Status:** --

**Compressive Strength:** ASTM D-695 and D-256: 230 kg/cm<sup>2</sup>

**Thermal Stability:** --

**Radiation Stability:** Compressive strength increased with increased irradiation. Samples become hard and brittle in doses above  $6.9 \times 10^8$  rad.

**Biological Stability:** --

**Leach Resistance:** Fractional after 200 days:  $^{60}\text{Co} = 3.1 \times 10^{-4}$ ,  $^{137}\text{Cs} =$  undetectable

**Immersion Stability:** Crack formation after 1,200 days

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:**  $2.0$  to  $2.4 \times 10^{-2}$  cm<sup>3</sup>/g·Mrad

**RCRA Compliance:** --

**Conclusions:** The product with 50 wt% waste has a compressive strength of 203 kg/cm<sup>2</sup>, and low leachability.

## ORGANIC BINDER SUMMARIES

NO. 34

**TITLE:** Incorporation of BWR's Evaporator Concentrate in Polyethylene

**AUTHOR:** N. Moriyama; S. Dojiri; H. Matsuzuru

**PUBLICATION:** Nuclear and Chemical Waste Management, Vol. 3, No. 1

**DATE:** 1982

**ORGANIZATION:** Japan Atomic Energy Research Institute

**SUMMARY:** Low density polyethylene is melted at 190°C and mixed with waste to form homogeneous solid, using batch wiped film stirrer.

**Waste Form:** Low-density polyethylene

**Waste Type:** Sodium sulfate

**Waste Loading:** 50 - 70 wt% sodium sulfate evaporator concentrates

**Development Status:** Experimental, limited lab-scale application at time of publication

**Compressive Strength:** ASTM D-695: 137 - 327 kg/cm<sup>2</sup>

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Depends on waste loading, polyethylene type. Diffusion coefficient ranged from  $3 \times 10^{-4}$  cm<sup>2</sup>/day (70 % loading) to  $8.5 \times 10^{-8}$  cm<sup>2</sup>/day (40 % loading).

**Immersion Stability:** Water immersion for 120 days caused no increased swelling or weight gain.

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Batch-wiped film mixer used instead of a screw extruder, which is commonly used in plastics industry.

**ORGANIC BINDER SUMMARIES****NO. 35**

**TITLE:** Testing of the Bituminized Ion-Exchange Resin Waste Products from a Nuclear Power Plant

**AUTHOR:** A.K. Muurinen; U.S. Vuorinen

**PUBLICATION:**

**DATE:** 1981

**ORGANIZATION:** International Atomic Energy Agency

**SUMMARY:**

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | <b>Bitumen</b>   |
| <b>Waste Type:</b>                | Finnish nuclear power plant low- and medium-level wet wastes (ion-exchange resins, filter masses, and evap. conc.)   |
| <b>Waste Loading:</b>             | --   |
| <b>Development Status:</b>        | --   |
| <b>Compressive Strength:</b>      | --   |
| <b>Thermal Stability:</b>         | --   |
| <b>Radiation Stability:</b>       | Samples exposed to ~52 Mrad over ~6 days. The waste swelled ~20 to 25% as a result of irradiation.   |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | --   |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | --   |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | Radiation somewhat affects the properties of waste products and of pure bitumen. Post-irradiation tests performed included relative density, softening point, breaking point, flash point, and water content. Activity distributions were also measured. |

## ORGANIC BINDER SUMMARIES

NO. 36

TITLE: Properties of Radioactive Waste Containers Progress Report

AUTHOR: R.M. Neilson, Jr.; P. Colombo

PUBLICATION: No. 11, BNL-NUREG-51042

DATE: 1979

ORGANIZATION: Brookhaven National Laboratory

SUMMARY: Effects of radiation exposure on Urea Formaldehyde waste form leachability.

|                            |   |                  |               |
|----------------------------|---|------------------|---------------|
| Waste Form:                | Urea Formaldehyde   |                  |               |
| Waste Type:                | Diatomaceous earth slurry   |                  |               |
| Waste Loading:             | waste/UF = 2.0 by weight  |                  |               |
| Development Status:        | Commercial  |                  |               |
| Compressive Strength:      | --  |                  |               |
| Thermal Stability:         | --  |                  |               |
| Radiation Stability:       | --  |                  |               |
| Biological Stability:      | --  |                  |               |
| Leach Resistance:          | Cumulative Fraction Release   |                  |               |
|                            | <u>Rad Exposure (<sup>60</sup>Co)</u>   | <u>Strontium</u> | <u>Cesium</u> |
|                            | 10 <sup>5</sup>   | 2.824            | 0.043         |
|                            | 10 <sup>6</sup>   | 2.741            | 0.038         |
|                            | 10 <sup>7</sup>   | 3.165            | 0.057         |
|                            | 10 <sup>8</sup>   | 3.706            | 0.151         |
| Immersion Stability:       | --  |                  |               |
| Free Liquids:              | --  |                  |               |
| Chemical Durability:       | --  |                  |               |
| Compositional Flexibility: | --  |                  |               |
| Gas Generation:            | --  |                  |               |
| RCRA Compliance:           | --  |                  |               |
| Conclusions:               | A radiation dose dependence was found for both cesium and strontium release from urea formaldehyde waste forms. |                  |               |

**ORGANIC BINDER SUMMARIES****NO. 37**

**TITLE:** Solidification of Liquid Concentrate and Solid Waste Generated as By-Products of the Liquid Radwaste Treatment Systems in Light Water Reactors

**AUTHOR:** R.M. Neilson, Jr.; P. Colombo

**PUBLICATION:** BNL-NUREG-22839

**DATE:** 1977

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Overview of solidification process and non-specific comparison of products.

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Urea Formaldehyde  |
| <b>Waste Type:</b>                | Various  |
| <b>Waste Loading:</b>             | Highly dependent on waste type   |
| <b>Development Status:</b>        | Commercial   |
| <b>Compressive Strength:</b>      | --   |
| <b>Thermal Stability:</b>         | --   |
| <b>Radiation Stability:</b>       | --   |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | $5 \times 10^{-1}$ to $5 \times 10^3$ g/(cm <sup>2</sup> -day)   |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | Acidic free standing water produced during polymerization  |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | Properties relevant to the evaluation of solidified waste forms are identified and relative comparisons are made for wastes solidified by various processes. |

## ORGANIC BINDER SUMMARIES

NO. 38

**TITLE:** Solidification of Liquid Concentrate and Solid Waste Generated as By-Products of the Liquid Radwaste Treatment System in Light Water Reactors

**AUTHOR:** R.M. Nielson, Jr.; P. Colombo

**PUBLICATION:** BNL-NUREG-22839

**DATE:** 1977

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** Table listing nonspecific properties of Dow polymer waste forms.

|                          |  |
|--------------------------|--|
| Waste Form:              | Dow polymer  |
| Waste Type:              | Various  |
| Waste Loading:           | --   |
| Development Status:      | Commercial   |
| Compressive Strength:    | --   |
| Thermal Stability:       | --   |
| Radiation Stability:     | --   |
| Biological Stability:    | --   |
| Leach Resistance:        | $10^{-3}$ to $10^{-5}$ g/(cm <sup>2</sup> -day)  |
| Immersion Stability:     | --   |
| Free Liquids:            | --   |
| Chemical Durability:     | --   |
| Compositional Stability: | --   |
| Gas Generation:          | --   |
| RCRA Compliance:         | Not ignitable  |
| Conclusions:             | When compared to other waste solidification systems, the Dow polymer is a viable choice. |



## ORGANIC BINDER SUMMARIES

NO. 39

**TITLE:** Long-Term Behavior of Bituminized waste

**AUTHOR:** M. Snellman; M. Valkiainen

**PUBLICATION:** Proceedings of the Symposium on Waste Management at Tucson, Arizona

**DATE:** March 1-5, 1987

## ORGANIZATION:

## SUMMARY:

|                                   |   |
|-----------------------------------|---|
| <b>Waste Form:</b>                | Bitumen   |
| <b>Waste Type:</b>                | Ion-exchange resins   |
| <b>Waste Loading:</b>             | --  |
| <b>Development Status:</b>        | --  |
| <b>Compressive Strength:</b>      | --  |
| <b>Thermal Stability:</b>         | Below 0°C bitumen waste forms are brittle and may be cracked by mechanical impact. Cracks may heal if forms are reheated to room temperature.   |
| <b>Radiation Stability:</b>       | Radiolysis gas (mostly hydrogen) is generated at a rate and in an amount proportional to the absorbed radiation dose.   |
| <b>Biological Stability:</b>      | Anaerobic microbial attack on a monolithic bitumen block takes place extremely slowly.  |
| <b>Leach Resistance:</b>          | Samples were leached in cement equilibrated and distilled water using a test method developed by K. Brodersen. Leaching tends to be much faster in "cement water."  |
| <b>Immersion Stability:</b>       | If swelling of bitumenized ion-exchange resins is restricted, but uptake of water is possible, considerable pressure may develop.   |
| <b>Free Liquids:</b>              | --  |
| <b>Chemical Durability:</b>       | --  |
| <b>Compositional Flexibility:</b> | Waste loadings up to 50%  |
| <b>Gas Generation:</b>            | --  |
| <b>RCRA Compliance:</b>           | --  |
| <b>Conclusions:</b>               | Behavior of waste product depends on the solidification process parameters. In the context of this study, no unacceptable long-term effects have been found caused by bitumen itself, but swelling must be taken into account for disposal. |

**ORGANIC BINDER SUMMARIES**

NO. 40

**TITLE:** The Leachability and Mechanical Integrity of Simulated Decontamination Resin Wastes Solidified in Cement and Vinyl Ester-Styrene

**AUTHOR:** P. Soo; L.W. Milian; P.L. Piciulo

**PUBLICATION:** Topical Report, BNL-NUREG-52149

**DATE:** 1988

**ORGANIZATION:** Brookhaven National Laboratory

**SUMMARY:** The release rates of organic decontamination reagents were measured for mixed-bed resins solidified in vinyl ester styrene.

**Waste Form:** Vinyl ester-styrene

**Waste Type:** Simulated LOMI Waste

**Waste Loading:** 50 wt%

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** --

**Biological Stability:** --

**Leach Resistance:** Cumulative release fraction = .088 after 144 days for waste form size of 5 cm x 10 cm.

**Immersion Stability:** No cracking, spalling, swelling, etc. were observed after 540 to 630 days immersion.

**Free Liquids:** --

**Chemical Durability:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** Vinyl ester-styrene waste forms have superior leaching and compositional stability properties compared to cement based waste forms.

## ORGANIC BINDER SUMMARIES

NO. 41

**TITLE:** High-Strength Asphalt Solidification Process for Low-Level Radioactive Wastes

**AUTHOR:** Letter from M. Tokar (NRC) to O. Wong encloses NRC Technical Evaluation Report on USE-61-002-P

**DATE:** August 2, 1991

**SUMMARY:**

**Waste Form:** Bitumen (distilled)  
**Waste Type:** Boric acid concentrates  
**Waste Loading:** --  
**Development Status:** --  
**Compressive Strength:** Test Method: ASTM D-1074. Pure asphalt compressive strength was 430 psi. Compressive strength for boric acid wastes were 1055 to 1185 psi.  
**Thermal Stability:** Test Method: ASTM B553. Post-test compressive strengths for boric acid wastes were >900 psi  
**Radiation Stability:** Samples exposed to  $10^8$  rad. Post-test compressive strengths for boric acid wastes were >800 psi.  
**Biological Stability:** Test Method: ASTM G21 & G22. Post-test compressive strengths were 850 to 1085 psi. One sample had growth and was retested using Bartha Pramer, which it passed.  
**Leach Resistance:** Test Method: ANS 16. 1. Leach indices ranged from 7.77 to 10.91 in demineralized water and from 7.65 to 9.90 in simulated sea-water  
**Immersion Stability:** Following immersion for 90 days, one waste form (pH = 7, waste loading = 50.68 wt%) had a compressive strength of 96 psi. Another (pH = 9, waste loading = 43.85 wt%) had a compressive strength of ~1075 psi.  
**Free Liquids:** None  
**Chemical Durability:** --  
**Compositional Flexibility:** Up to 12 wt% solids. Limited to pH=9, waste loading < 40 wt%.  
**Gas Generation:** --  
**RCRA Compliance:** Under normal conditions, the waste form will not cause fires. Minimum flash point is 500°F  
**Conclusions:** The boric acid waste form has been approved by the NRC, under certain conditions. Other wastes may be qualified for use with this binder system following future testing.

**ORGANIC BINDER SUMMARIES****NO. 42**

**TITLE:** "10 CFR 61" Waste Form Conformance Program for Solidified Process Waste Products Produced by a WasteChem Corporation Volume Reduction and Solidification (VRS) System

**AUTHOR:** Letter from M. Tokar (NRC) to Klein (WasteChem), encloses NRC Technical Evaluation Report on VRS-002

**PUBLICATION:**

**DATE:** January 22, 1988

**SUMMARY:**

**Waste Form:** Bitumen (ASTM D-312, Type III)

**Waste Type:** Simulated generic process wastes from commercial PWRs and BWRs

**Waste Loading:** --

**Development Status:** --

**Compressive Strength:** Test Method: ASTM D-1074. At 10% sample deformation, results ranged from 108 psi to 262 psi.

**Thermal Stability:** Test Method: ASTM B553. Post-test compressive strengths range from 81.2 psi to 276 psi.

**Radiation Stability:** Samples exposed to 108 Mrad, over a 239.9 hour period had post-irradiation compressive strengths of 220 and 270 psi.

**Biological Stability:** Test Method: Bartha-Pramer. Total carbon loss from waste forms was projected to be <10% over 300 years.

**Leach Resistance:** Test Method: ANS 16. 1. Leach index range was 8.07 to 13.76

**Immersion Stability:** Immersion for 90 days in deionized water. Compressive strength range after immersion, 74 psi to 250 psi. Waste loading for simulated evaporator concentrate was decreased to 25%. Higher loadings led to swelling and disintegration.

**Free Liquids:** None

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** NRC approved the WasteChem topical report with limitations on waste loading, (none are greater than 60 wt% waste). In addition, waste forms must be contained in 55-gal drums or approved HICs. Backfill should be used if 55-gal drums are used.

**ORGANIC BINDER SUMMARIES**

NO. 43

**TITLE:****AUTHOR:** U.S. Environmental Protection Agency**PUBLICATION:** Toxicity Characteristic Leaching Procedure, 40 CFR 261**DATE:** March 29, 1990**ORGANIZATION:** U.S. Environmental Protection Agency**SUMMARY:** Testing to determine the mobility of both organic and inorganic analytes present in liquids, solids, and multiphase wastes.

|                                   |  |
|-----------------------------------|--|
| <b>Waste Form:</b>                | Polyethylene   |
| <b>Waste Type:</b>                | Sodium nitrate salt containing sufficient chromium to be classified as hazardous                 |
| <b>Waste Loading:</b>             | --   |
| <b>Development Status:</b>        | --   |
| <b>Compressive Strength:</b>      | --   |
| <b>Thermal Stability:</b>         | --   |
| <b>Radiation Stability:</b>       | --   |
| <b>Biological Stability:</b>      | --   |
| <b>Leach Resistance:</b>          | --   |
| <b>Immersion Stability:</b>       | --   |
| <b>Free Liquids:</b>              | --   |
| <b>Chemical Durability:</b>       | --   |
| <b>Compositional Flexibility:</b> | --   |
| <b>Gas Generation:</b>            | --   |
| <b>RCRA Compliance:</b>           | --   |
| <b>Conclusions:</b>               | Toxicity Characteristic Leaching Procedure recently replaced the EP Toxicity Test in 40 CFR 261. |

**ORGANIC BINDER SUMMARIES****NO. 44****TITLE:****AUTHOR:** U.S. Department of Transportation**PUBLICATION:** 49 CFR 173, Appendix C, Federal Register 46, 31294**DATE:** June 15, 1981**ORGANIZATION:** U.S. Department of Transportation**SUMMARY:** Vibration testing for hazardous waste form packaging that is to be transported.**Waste Form:** Polyethylene**Waste Type:** Not specified**Waste Loading:** --**Development Status:** --**Compressive Strength:** --**Thermal Stability:** --**Radiation Stability:** --**Biological Stability:** --**Leach Resistance:** --**Immersion Stability:** --**Free Liquids:** --**Chemical Durability:** --**Compositional Flexibility:** --**Gas Generation:** --**RCRA Compliance:** --**Conclusions:** Replaced by "Test for Solid Oxidizing Substances," for DOT use to qualify hazardous wastes.

## ORGANIC BINDER SUMMARIES

NO. 45

**TITLE:** Long-Term Properties of TVO's Bituminized Resins

**AUTHOR:** M. Valkiainen; U. Vuorinen

**PUBLICATION:** YJT-89-06

**DATE:** June 1989

**ORGANIZATION:** Nuclear Waste Commission of Finnish Power Companies

**Waste Form:** Bitumen

**Waste Type:** Spent ion-exchange resins (Nordic)

**Development Status:** --

**Compressive Strength:** --

**Thermal Stability:** --

**Radiation Stability:** In individual drums (200 l), the amount of generated radiolytic gas may be appreciable in terms of drum volume leading to possible swelling of the product and possible pressurizing of the drum.

**Leach Resistance:** Leaching in cement equilibrated water with Brodersen method. Samples have been in leachant for several years. Leaching tends to increase with decreasing pH.

**Immersion Stability:** Pretreating anion ion-exchange resin by drying at 140°C for 14 hours reduces rewetting and swelling capacity. Only in products with resin contents below 20%, can the resin particles be completely bitumen surrounded.

**Free Liquids:** --

**Chemical Durability:** --

**Compositional Flexibility:** --

**Gas Generation:** --

**RCRA Compliance:** --

**Conclusions:** --

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