PATENT SPECIFICATION

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NO DRAWINGS

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COMPLETE SPECIFICATION

A Salt of a Nitrogen-Containing Ester of Phosphoric Acid and the Therapeutic Applications thereof

We, Hans Alfred Nieper, of Sedanstrasse 21, 3 Hannover, Germany, (formerly of Krankenhaus Silbersee, Langerhagen-Hanover, Germany), and Franz Köhler, of Dr. Köhler Chemie, Alsbach, Germany, both German citizens, do hereby declare the invention for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:—

This invention relates to new compositions of therapeutic value.

According to the present invention there is provided, as a new therapeutic agent, a composition comprising the mono calcium salt of phosphoric acid monoaminoethyl ester and potassium and/or magnesium aspartates. For brevity this ester compound is referred to herein as Ca—AEP. Its formula may be written as:

However the stability of the 2-aminoethyl phosphoric acid, and its salts, to hydrolysis suggests that the compound may have an internal ring structure and be represented by the formula:

The compound may be made by the treatment of 2-aminoethylphosphoric acid with calcium carbonate or other convenient water-soluble calcium salt or calcium hydroxide. When calcium carbonate is added gradually to an equimolecular amount of 2-aminoethylphosphoric acid in aqueous medium the solution remains clear for a short time and then the Ca—AEP precipitates rapidly. The precipitate may conveniently be separated by filtration and dried *in vacuo* at a temperature not exceeding 40°C. The product may thus be obtained in the form of a white or greyish-white powder of polygonal crystals. It has no ascertainable melting point as it decomposes on

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<u>, </u>	heating. It is insoluble in most common solvents but soluble in water to the extent	
1828	of 3.8 to 3.9% at room temperature.	
	The combination at Ca-AEP and K and/or Mg aspartates has valuable thera-	
	peutic properties and can be used for the treatment of various clinical conditions as set	5
5	out below. New therapeutic compositions according to the invention may comprise in)
	addition to the Ca—AEP and K and Mg aspartates, a therapeutically acceptable	
7 .	diluent or extender. The diluent or extender may be an aqueous medium, e.g. a	
	sterile aqueous medium, or a solid, the composition being in the form of tablets, or	
:	a resorbable material such as cocoa butter.	
10	Suitably, such compositions may take any of the following forms. In each of	10
	these the quantity of Ca-AEP is indicated; the quantity of K and Mg aspartates is	
	in each case adjusted to provide a daily dosage of 1 to 2g. for internal administration.	
	a) Sterile aqueous vials (preferably 10 ml) for intravenous and/or intramuscular	
	injections, containing, for example, 4 per cent of Ca—AEP,	
15	b) Liquid oral preparations containing, for example, 1 to 4 per cent of Ca—AEP in	15
1.7	solution in a suitable vehicle, or up to 10 per cent as a suspension. The vehicle	
	may be flavoured.	
	c) Tablets, uncoated, scored or not, or entericoated (gastric juice resistant), containing,	
•	for example from 0.2 to 0.4 gm (preferably 0.35 gm) of Ca—AEP. Such tablets	20
20	may contain for example talc, and/or magnesium stearate. A suitable daily dosc is	~~
	0.8 to 2 gm.	
	d) Suppositories containing, for example, from 0.4 to 1 gm. of Ca-AEP, prefer-	
	ably in a highly resorbable mass e.g. cocoa butter. A suitable daily dose is from	
	0.8 to 8 gm.	25
25	e) Compositions for application to the skin or to mucous membranes containing, for	23
	example, from 0.5 to 10 percent of Ca-AEP, such as a lotion in aqueous	
	resorbable solution, emulsion or ointment bases.	
100	f) Ophthalmic preparations, such as ointments in a suitable base or collyrium, con-	
	taining, for example, 0.05 to 4 per cent of Ca—AEP in a solution of adequate	- 40
30	tonicity and buffer capacity.	30
-	The compound Ca—AEP was found to have a toxicity (determined l.p in the	
	mouse) of $LD_{50} = 0.5$ gm/kg and $LD_{100} = 1.25$ gm/kg.	
-	The compositions of the present invention, in their various dosage forms have	
12 1 1	been found to be active in the treatment of:	
35	(1) Autoimmune Diseases	35
- F	Colitis ulceros and mucosa	
-	Hepatitis (chronic and non cirrhotic)	
	 Chronic nephritis and associated hypertension 	
21.5	 Nephrosclerosis, malignant, and associated fixed hypertension 	
40	 Myocarditis, Jaffe's myocarditis, post-infarct syndrome 	40
	 Post-cardiotomy syndrome 	
4	- Multiple sclerosis	
	 Osteonecrosis 	
	- Rheumatic manifestations, including rheumatoid arthritis, rheumatic fever,	
45	myocarditis	45
	- Sclerodermia	
-	— Chronic inflammation with or without tuberculosis.	
	(2) Allergic Diseases	
	(3) Inflammatory Diseases	
50	Hemorrhoids (suppositories)	50
- 20	Skin inflammatory diseases (topical). Dermatitis	•
	- Eye diseases.	
12 T. T.	(4) Eczema	
	(5) Smooth Muscle Spasma	
55	— Intestinal	EE.
55	— Gastric	55
	— Bronchial (asthma)	
	(6) Lupus erythematosus	
40	(7) Gastritis	
6 U	(8) Tuberculosis	60
	(9) Osteoporosis	
	(10) Aging	
	(11) Juvenile Diabetis	
	(12) Treatment and diagnosis of Cancer	
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	(13) Progressive Muscle Dystrophy	
	(14) Angiosmastic Hypertension	
	(15) Chronic Encephalitis	
5	(16) Spondylitis Osteoporosis (17) Intersticial Pulmonary Fibrosis	_
J	(17) Intersticial Pulmonary Fibrosis (18) Myalgia	5
	(19) Breast Induration	
	(20) Consolidation of Bone Fracture	
	In copending Application No. 48613/64 (Serial No. 1,079,569) we have described	
10	as a new therapeutic agent, the compound calcium di-aspartate. In the preferred form	10
	the said agent consists of a mixture of 25% by weight of the D-form of that salt and	
	75% by weight of the L-form of that salt.	
	For some indications such as in dermatology, arthritis and colitis Ca-AEP	
15	represents a therapeutic improvement over such Ca aspartate. The therapeutic applica-	
15	tions are similar to those described for Ca aspartate, but in some cases, therefore, the	15
	therapeutic doses of Ca—AEP are lower. Comparison of effects have been conducted	
	on the same patients.	
	In the treatment of inflammatory conditions of internal organs and limbs, 3 gm	
20	of Ca 1,dl, aspartate are equivalent to 1 gm of Ca—AEP. In the treatment of eczema or skin inflammation, very high doses of Ca 1, dl, aspartate, at least 10 times greater	20
	than doses of Ca—AEP, are unable to achieve the same results as the latter.	20
	Following clinical experience with calcium aspartate, and particularly with	
	Ca 1, dl-aspartate and the Mg and K-2-aminoethylphosphate (Mg-K-AEP), Ca-2-	
	aminoethyl-phosphate (Ca—AEP) was used in all inflammatory, rheumatic, auto-	
25	immune and allergic conditions. Since it has been observed in the past that cancer	25
	patients reacted to a mixture of Mg and K-AEP by episodes of intense shivering,	
	but that this sympton was only observed in these conditions, Ca—AEP was tested for	
	this purpose in cancer cases. It was concluded that tumor tissues are probably capable of	
30	liberating ethanolamine more rapidly from the AEP salt than other tissues. Pharma-	20
JU	cological tests have proven that ethanolamine leads to such temperature increase reaction and to shivering.	30
	It has been observed that the repeated use of Ca—AEP, particularly through the	
	i.v. route, leads occasionally to cardiolyascular side-effects or gall bladder side effects	
	that appear to be associated with K and Mg deficiency. This is corrected or avoided by	
35	the simultaneous administration of a mixture of K and Mg dl-aspartates i.e.	-35
	HOOC CH CHAMINGON IN O	
	HOOC—CH ₂ —CH(NH ₂)COOK.½H ₂ O and (HOOC—CH ₂ —CH(NH ₂)COO) ₂ Mg.4H ₂ O	
	(11000—011 ₂ —011(1411 ₂)C00) ₂ (41g,411 ₂ 0	
	preferably in equal parts by weight.	
	K and Mg, as their corresponding aspartic acid salts, have been demonstrated to	
4 0	be much more effective than their other salts in correcting disturbances associated	40
	with K and Mg deficiency. Moreover the combination of these two cations shows	
	definite potentiation of effects and occasionally therapeutic activity of a nature not observed with the use of either one of the cations alone.	
	It is generally recommended not to administer more than one i.v. injection of	
45	Ca—AEP every other day. I.V. administration should be alternated with oral or rectal	45
	administration.	40
	Ca—AEP:	
	All parts are parts by weight.	
	The following procedures will serve to illustrate the production of the compound	
50	Procedure 1.	50
	500 parts of calcium-carbonate are slowly added with continuous stirring to a	
	solution composed of 141 parts of phosphoric acid-mono-aminoethyl ester in 2000 parts	
	of water. The carbonate starts reacting with liberation of CO ₂ and then in a short time	
5 5	an almost clear solution is obtained in which the calcium salt of phosphoric acid-	
	mono-aminoethyl ester is rapidly precipitated. Once the liberation of CO ₂ has subsided,	55
	the reaction mixture is cooled at room temperature and the reaction product is	
	separated. After vacuum drying, 150 parts of the desired calcium salt is obtained in	
	the form of a snow-white crystalline product.	
60	Procedure 2.	60
	163 parts of the sodium salt of phosphoric acid-mono-aminoethyl ester are dis-	
	solved at room temperature in 1500 parts of water, to which are added with rapid	
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	stirring 73.5 parts of crystallized calcium chloride. The calcium chloride is rapidly dissolved and reacts with the sodium salt. The mixture is then warmed to about 50° and stirred for about one hour at this temperature, cooled, and the precipitate separated	
_	by filtration.	5
5	After drying of the precipitate under vacuum, 151 parts of calcium salt of	,
	phosphoric acid-mono-aminoethyl ester are obtained in the form of snow-white	
	glistening crystals.	
	WHAT WE CLAIM IS:—	
	1. A therapeutic composition which comprises the mono-calcium salt of phosphoric	10
10	acid monoaminoethyl ester and magnesium and/or potassium aspartates.	10
	2. A therapeutic composition according to claim 1 which comprises a therapeutic-	
	ally acceptable diluent.	
	3. A therapeutic composition according to claim 2 wherein the diluent is an	
	aqueous medium.	15
15	4. A therapeutic composition according to claim 2 wherein the diluent is a sterile	15
	aqueous medium.	
	5. A therapeutic composition according to claim 2 wherein the diluent is solid	
	and the composition is in the form of tablets.	
	6. A therapeutic composition according to claim 5 wherein the tablets contain	
20	talc and/or magnesium stearate.	20
	7. A therapeutic composition according to claim 2 wherein the diluent is a resorb-	
	able material and the composition is in the form of a suppository.	
	8. A therapeutic composition according to claim 7 wherein the diluent is cocoa	
	butter.	

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